

A mobile gasmixing unit

Creating precise gas mixtures

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Universal Gas Mixing Apparatus (UGMA)

Gas Studies

Performance

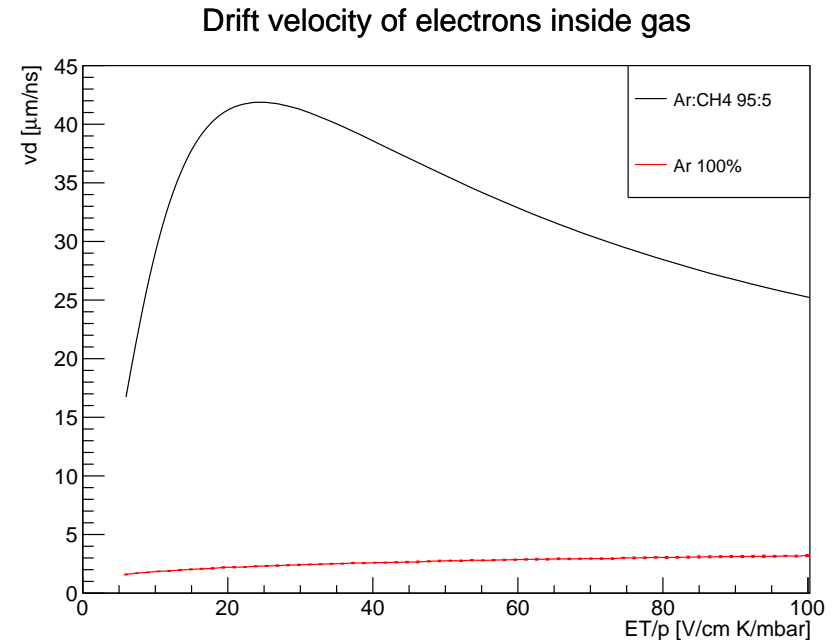
Summary & Outlook

Which gases are used inside a TPC?

Most mixtures used inside TPCs consist of two or three components!

Gas effects:

- Drift velocity: electrons and ions
- Diffusion: size of drifting electron cloud



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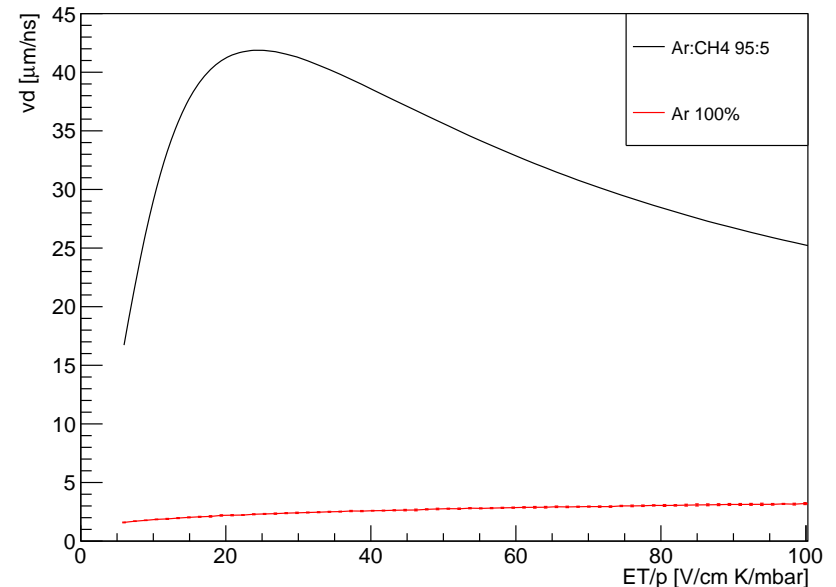
Gas effects:

- Drift velocity: electrons and ions
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Gas choice depends on:

- limits from environment (flammable gases allowed?)
- operating costs
- needed properties v_D and $D_{T,L}$
- interaction with detector materials
- ionisation properties (main component / noble gas)

Drift velocity of electrons inside gas



Need for a Gas System

Requests from R & D of gaseous detectors

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quick tests of new detectors with different mixtures and variations

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- ⇒ mobile system

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**SAFETY
FIRST**

Why a gassystem?

“Gas system? I’m using bottles!”

Bottled gases

Advantages

- high pressure (~200 bar)
- requires only little space

Disadvantages

- only fixed mixtures, no variations possible
- exotic mixtures have long delivery times
- normal precision only 10 %

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- variation of the mixture
- high precision < 0.1 vol.-%
- gas consumption can be reduced
- analysis, pressure regulation

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- space needed

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Costs for 1 m³ P10 (Ar:CH₄ 90:10)

$$1 \text{ m}^3 \text{ P10} \hat{=} 8.26 \text{ €}$$

ordering 10 ℓ 200 bar

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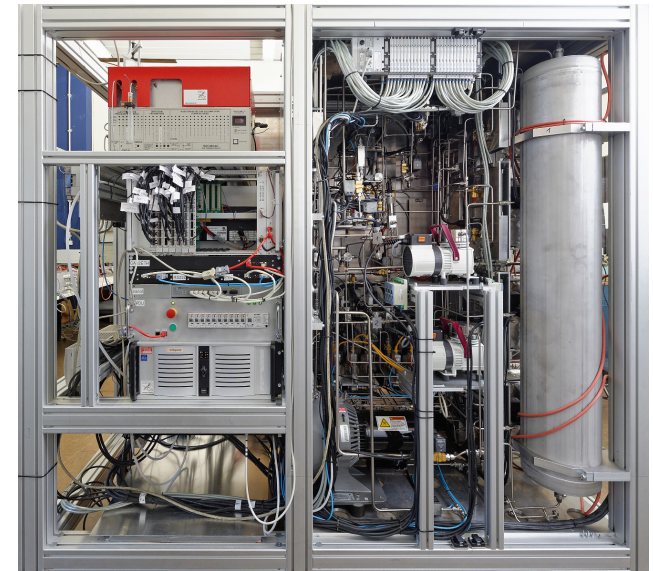
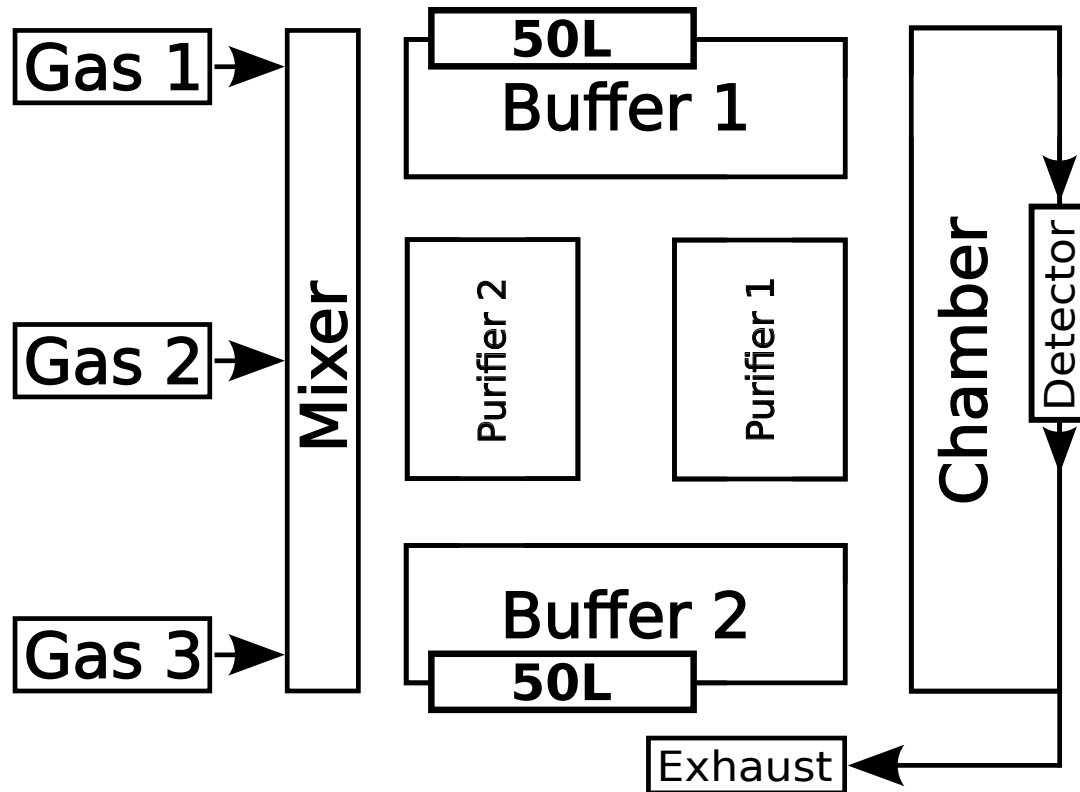
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UGMA

Universelle - Gas - Misch - Anlage
Universal - Gas - Mixing - Apparatus

Very basic schematic



Overview of the assembly groups

Mixer (3 lines input)

- two lines have three MFCs
max. 1, 10, 100 ℓ_n/h Ar
- one only 100 ℓ_n/h Ar main component

O₂ and H₂O Purifier (2 × 2 pcs.)

- each purifier is present twice
- one regenerating & one in use

Buffer (2 pcs.)

- $V_0 = 50 \ell$ equipped with pump
- partial pressure mixing
- gas storage

Monitoring:

temperature and pressure monitoring in all assembly groups

Chamber (detector interface)

- flow regulation (0 - 100 ℓ_n/h)
- pressure regulation (0 - 3 bara)
- continuous setpoint of the
gas recycle ratio 0-100 %

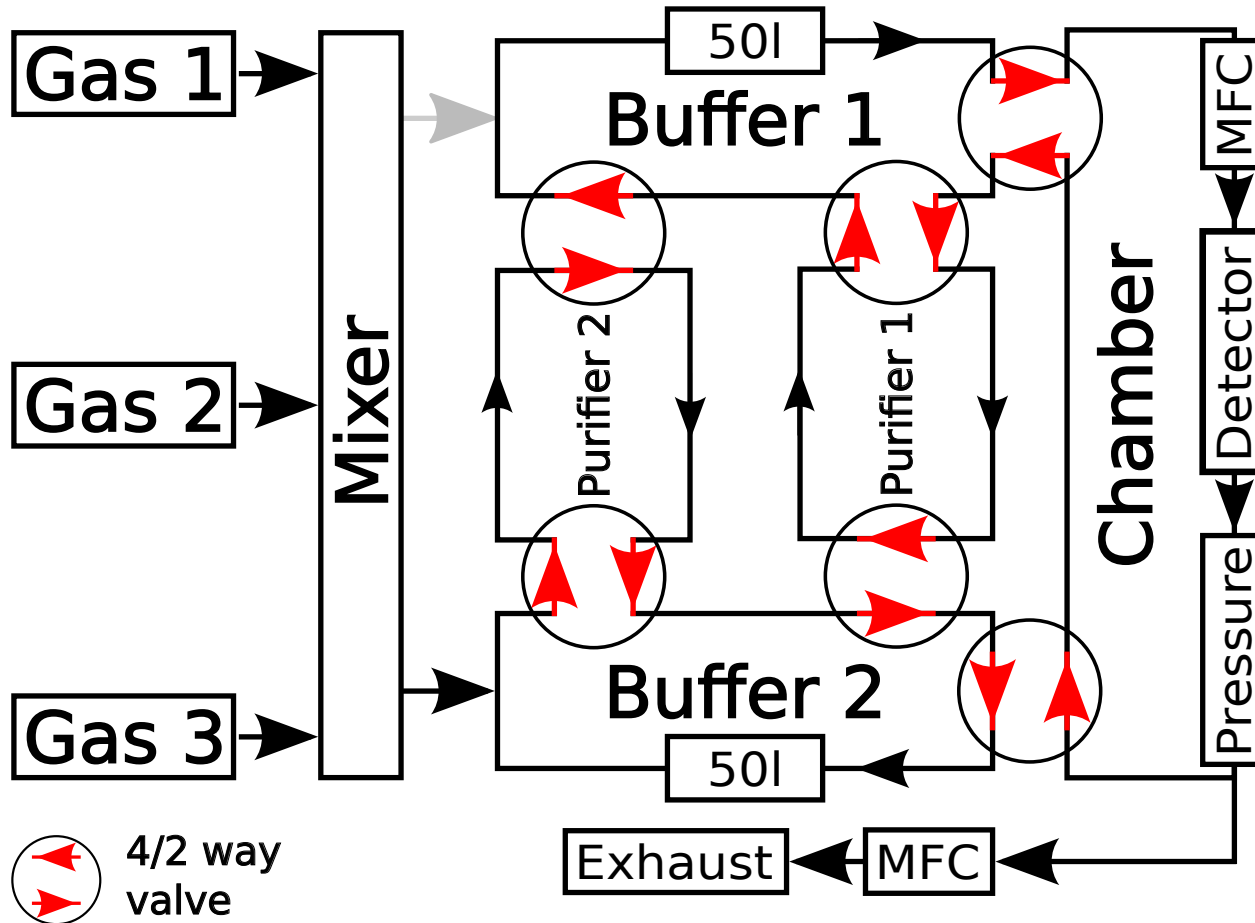
Analysis

- can be connected to every loop
- special sensors for O₂ & H₂O content
- gas chromatograph

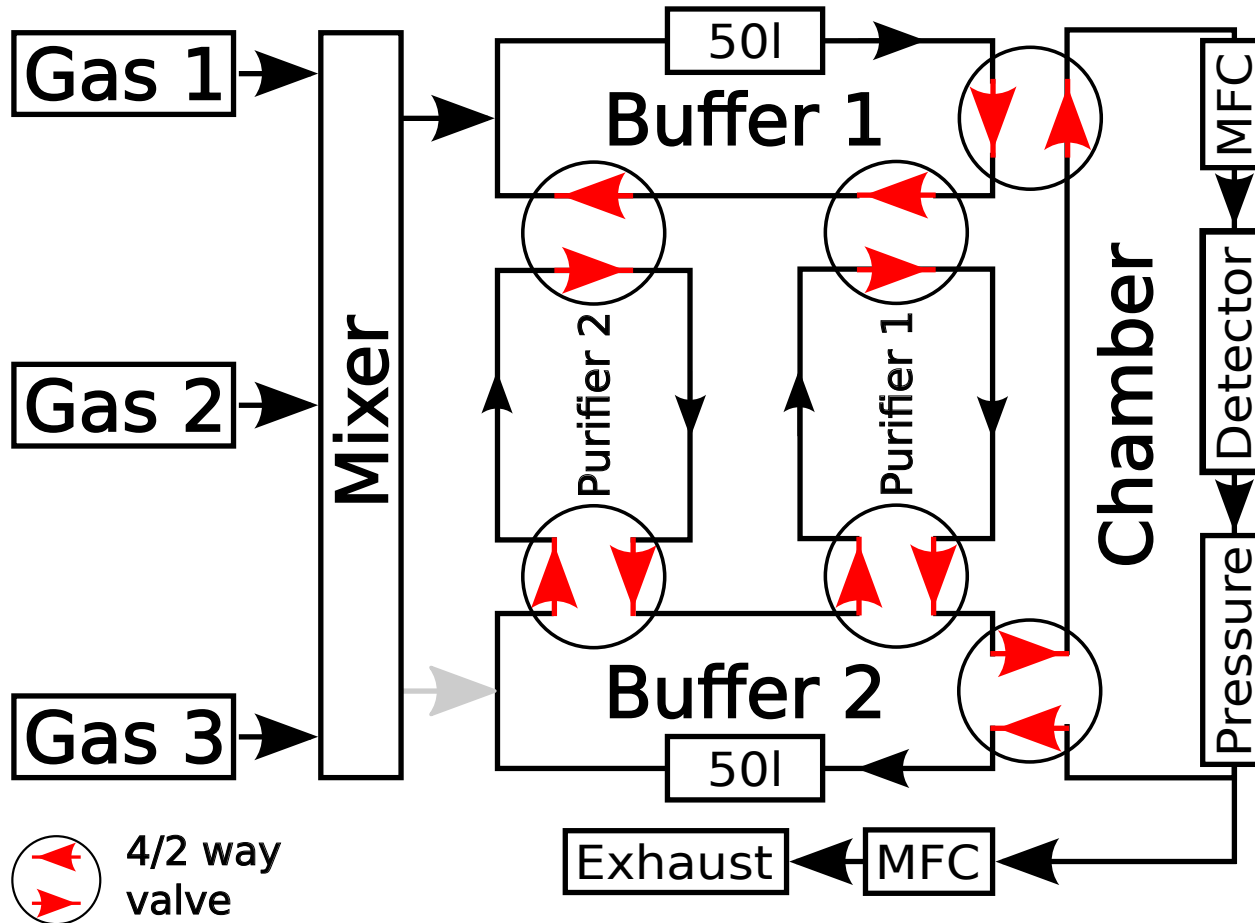
Safety

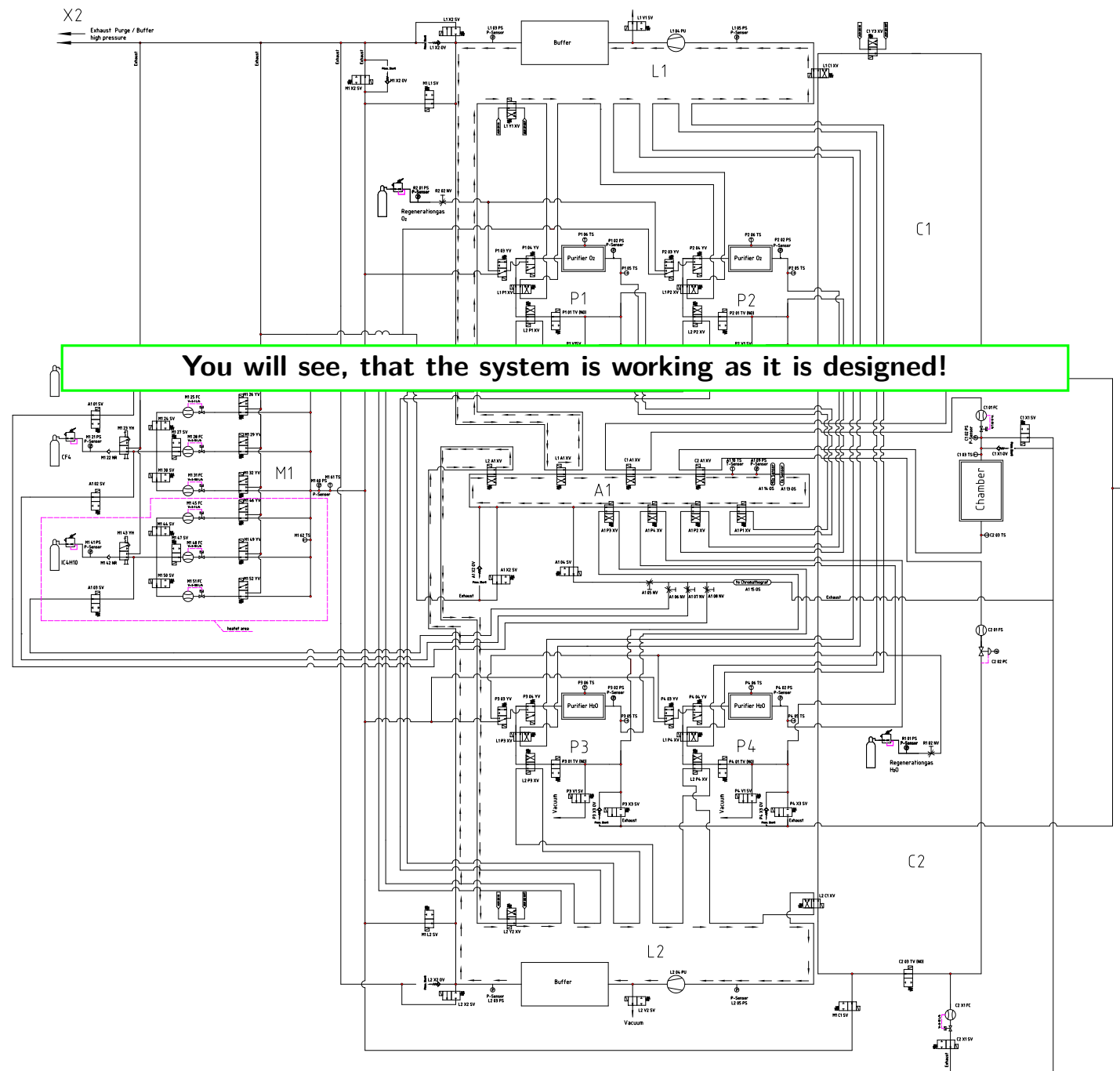
- flammable gas sensor
- interlock system

Looping

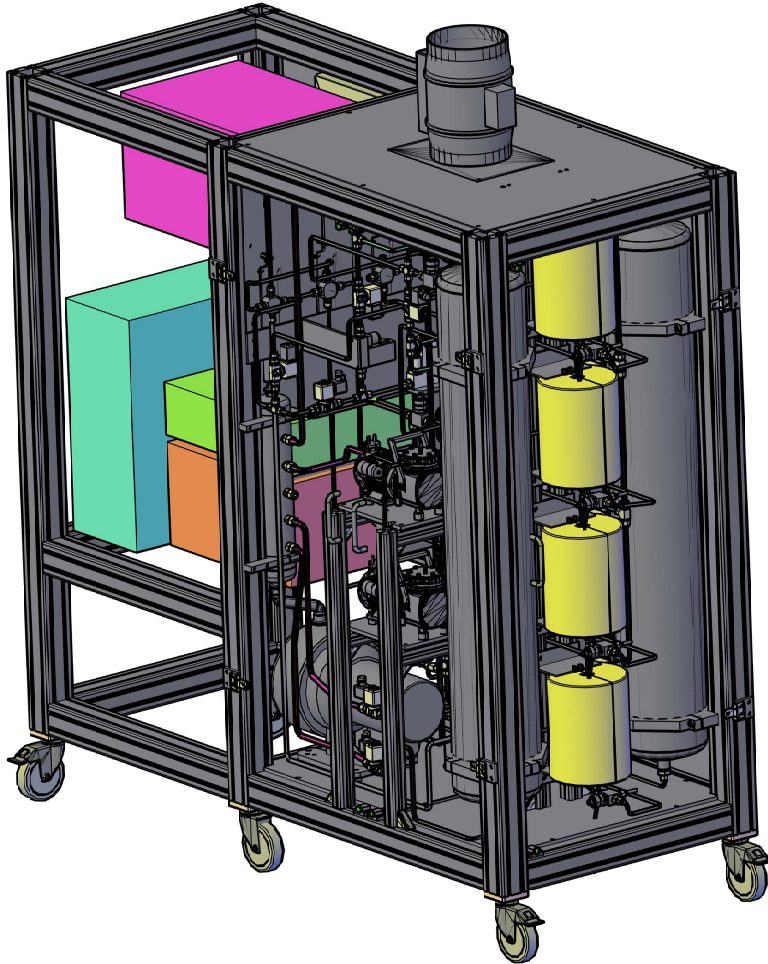


Looping





3D design



Overview - Mechanics

78 Valves

- 22 pneumatic 4/2-way valves
- 56 3/2-way & 2/2-way solenoid-valves

38 Sensors

- 18 pressure sensors
- 18 temperature sensors PT100
- 2 other sensors (H_2O , O_2)

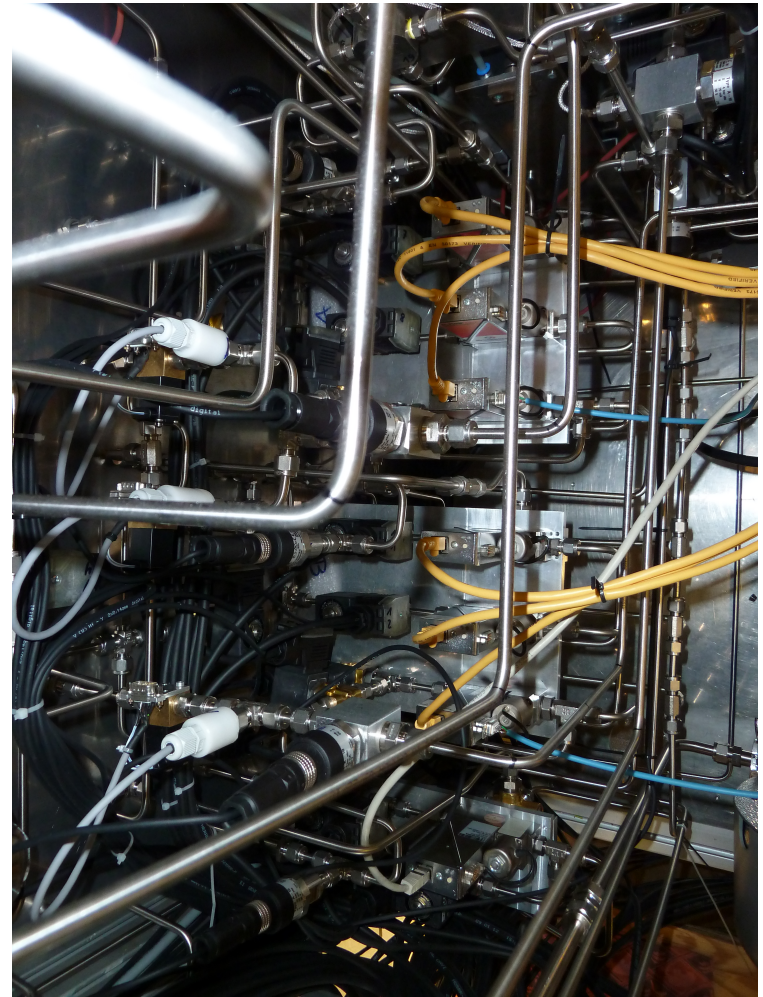
200 m Pipes

Stainless steel (for gas handling):

- ca. 120 m 6 mm
- ca. 20 m 10 mm

Flexible PU-tubing (for pneumatic valves):

- ca. 65 m 6 mm



Overview - Electronics

500 m Cables

- 240 m power cable
- 260 m signal cable

Data collection

- sampling all sensors with 0.5 Hz 12 Bit
- Data transfer via CAN-Bus and Ethernet

Power

- need three phase 230 V 16 A CEE
- only low voltage inside gas volume

Control

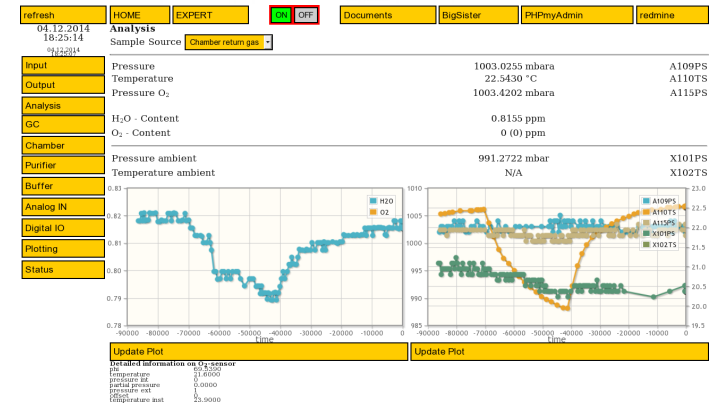
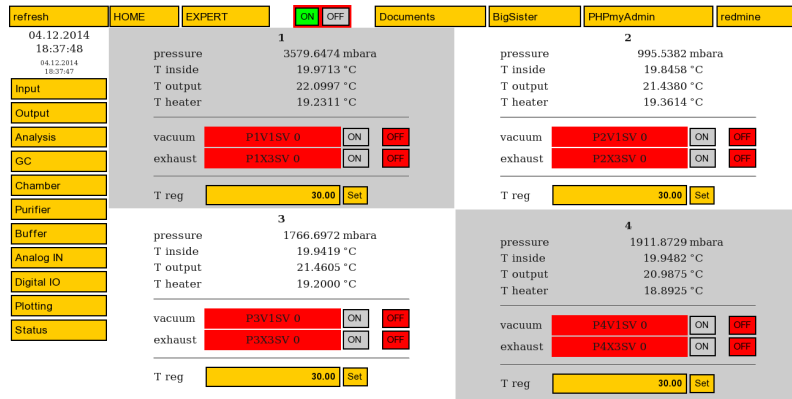
using a normal PC and some software



User Interaction

Control:

- remotely via web-browser
- locally via touch-screen
- same interface



Mixing Mode

Direct Flow Mixing

Application

- quick mix tests
- no changeover time

Method

Divide requested flow by fractions

$$\dot{V}_i = \eta_i \dot{V}_{\text{total}}$$

Gas dependence

Calibration in MFC or ext. conversion factor

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- long stable supply $\approx 12 \text{ h @ } 5 \ell_n/\text{h}$

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for ideal gas
further

$$V_i = p_i V_{\text{buffer}}$$

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for real gas using Redlich-Kwong equation

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Mixtures specified at:

$T_0 = 273.15 \text{ K}$ and $p_0 = 1013.25 \text{ mbar} \equiv \text{STP conditions DIN 1343}$
 \dot{V} is also given in norm-liters per hour ℓ_n/h

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Universal Gas Mixing Apparatus

Feature summary

Universal & Mobile

- can be moved through a normal office door
- has rolls to be moved easily
- can mix up to three gases each from 0 - 100 vol.-%

Gas Mixing - to be shown

- mixing better than commercial gas suppliers (10 %)
- accuracy ≤ 0.1 vol.-%

Apparatus

- using state of the art parts
- industrial standards

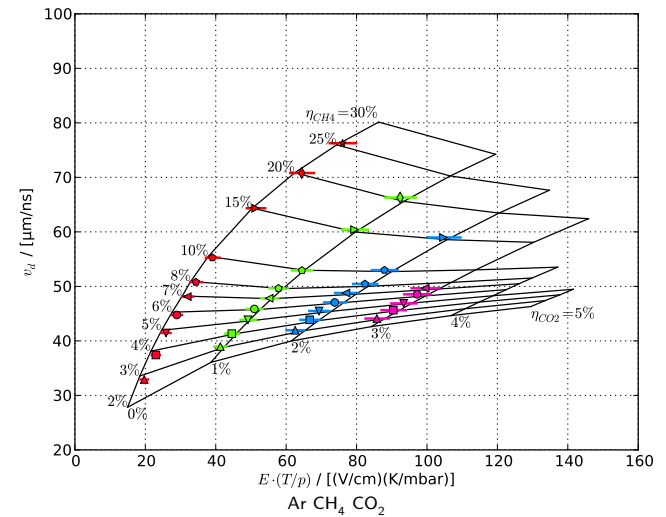
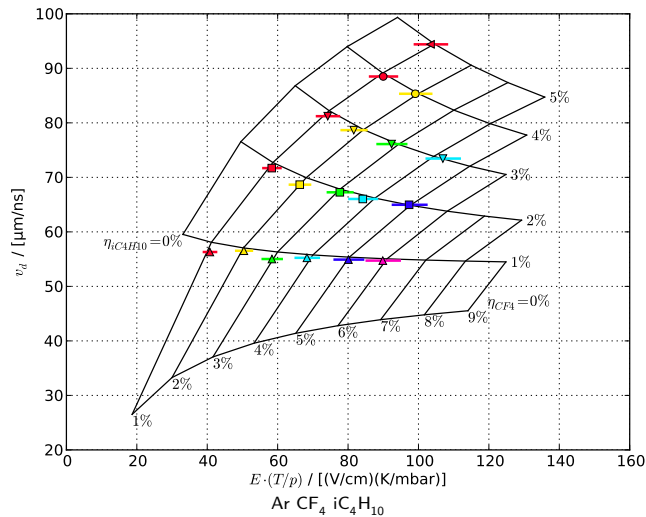
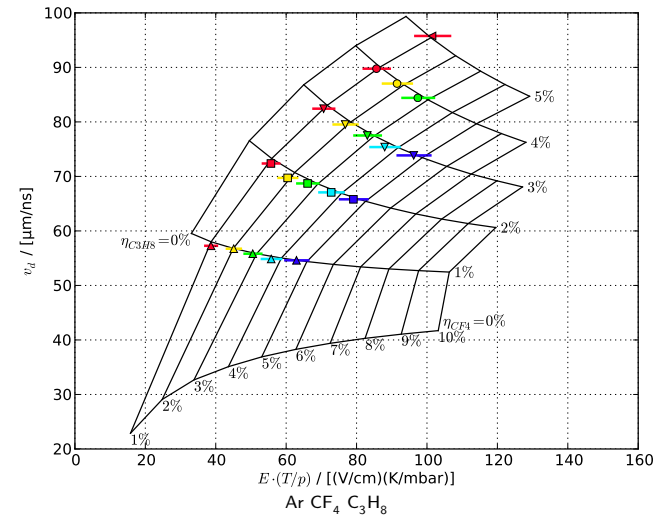
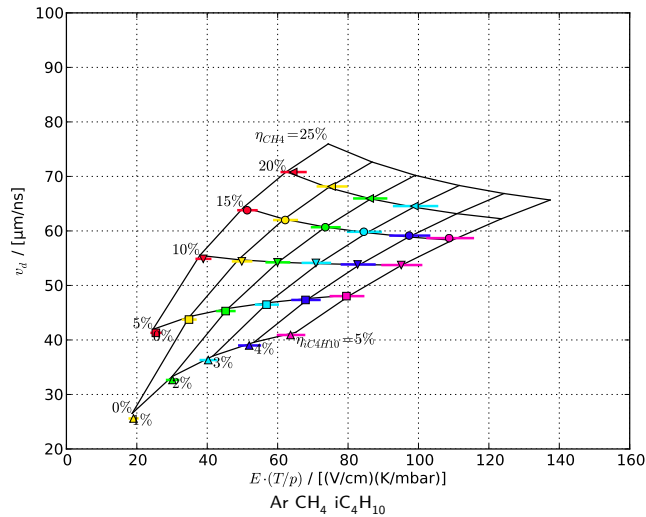


back to physics ...

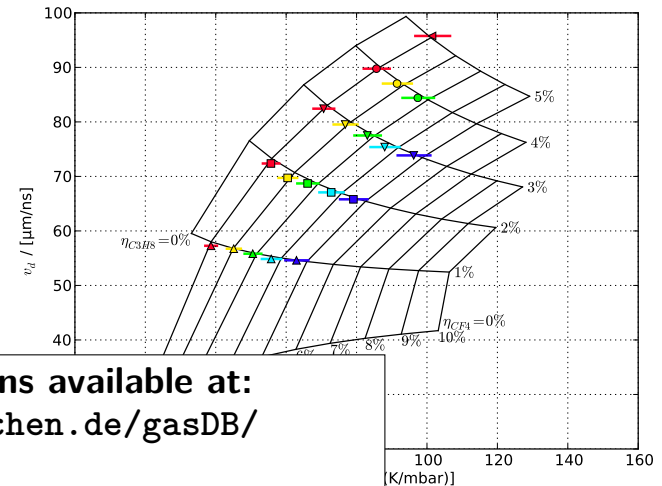
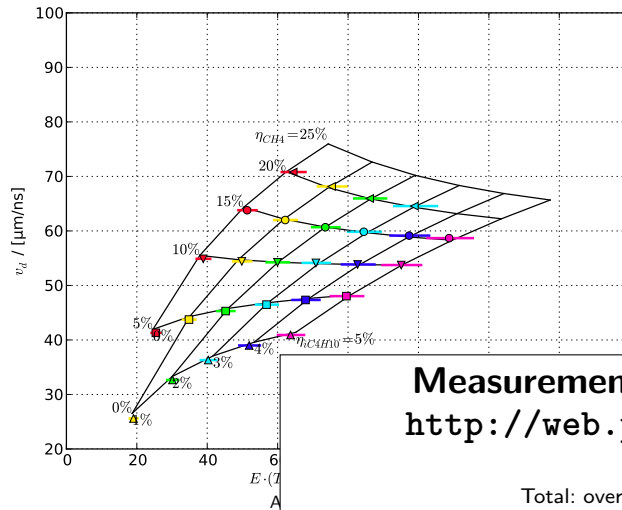
Gas Studies

focussing on electron drift velocity

Usable with lots of different gases



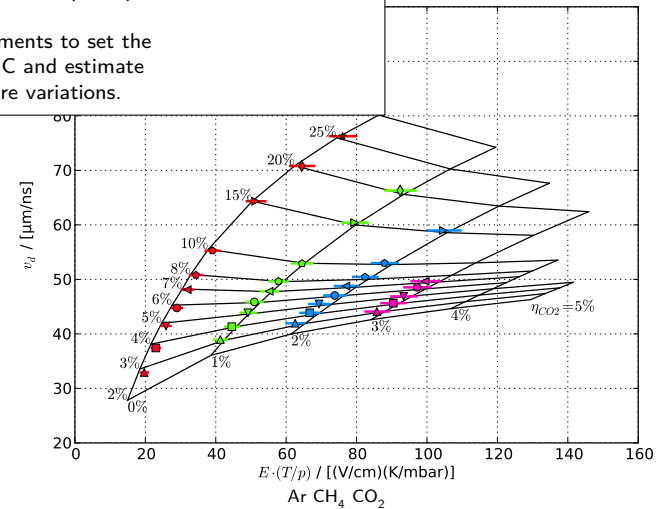
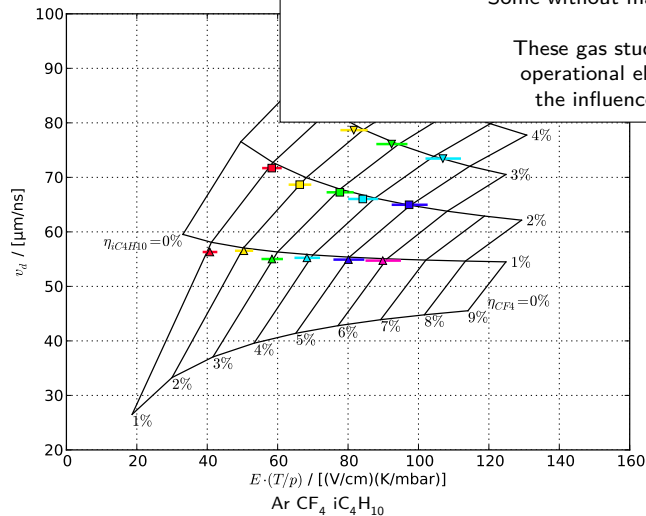
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Measurements and Simulations available at:
<http://web.physik.rwth-aachen.de/gasDB/>

Total: over 11200 simulations & over 280 measurements
 Some without maximum, so they are not shown in spider plots.

These gas studies may help further experiments to set the operational electrical field inside their TPC and estimate the influence of temperature and pressure variations.



Determination of the Performance

Howto determine the performance?

Problem: No analysis device with sufficient precision (0.01 vol.-%) for all gas types is available!

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Solution: We have great working and well understood monitoring chambers!

In addition we have an approved and powerful drift velocity simulation tool.

Magboltz supplies data for over 30 pure gases.

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Use the comparison of measurement and simulation to determine the performance of the UGMA!

Fitting the mixture...

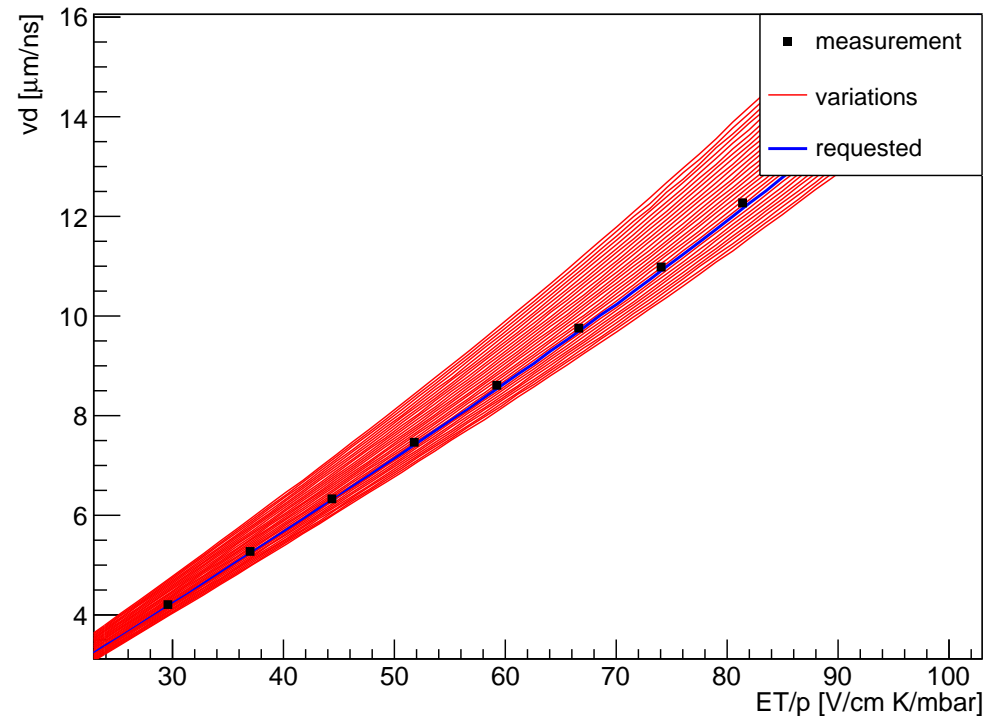
Needed: a sizeable amount of simulations

- Simulations done with Garfield++ which used FORTRAN Magboltz
 - step-size for binary mixtures $\Delta\eta = 0.01$ vol.-%
 - simulations around ± 0.5 vol.-% of set mixture
- ⇒ 100 Simulation curves per mixture each having 100 ET/p values

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CO2 16.00 Ar 84.00 - drift velocity



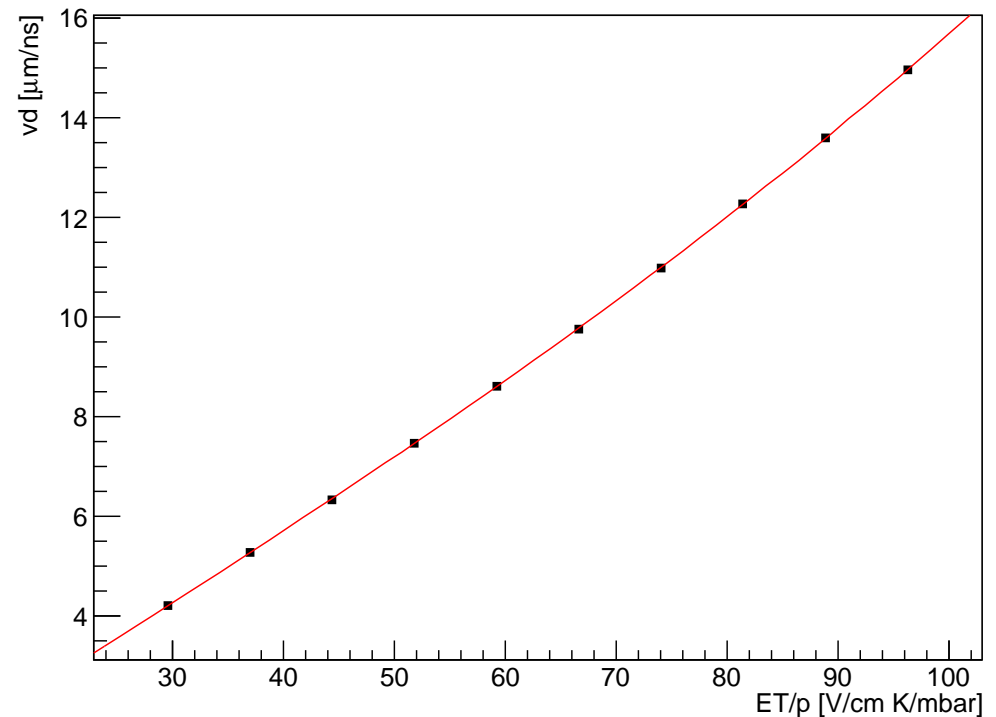
Fit the mixture via χ^2 minimisation:

$$\chi^2 = \sum_n \frac{(v_{d,\text{meas}}(E_n) - v_{d,\text{sim}}(E_n))^2}{\sigma_{v_{d,\text{meas}}}^2(E_n) + (v'_d(E_n) \sigma_{ET/p,\text{meas}}(E_n))^2}$$

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Skills, con
attitudes and
performance
Training is
view

First having a look at the UGMA

Calculated mixture

UGMA calculates the mixture from the measured values.

Direct Flow Mixing:

- measured flow of all lines

Partial Pressure Mixing:

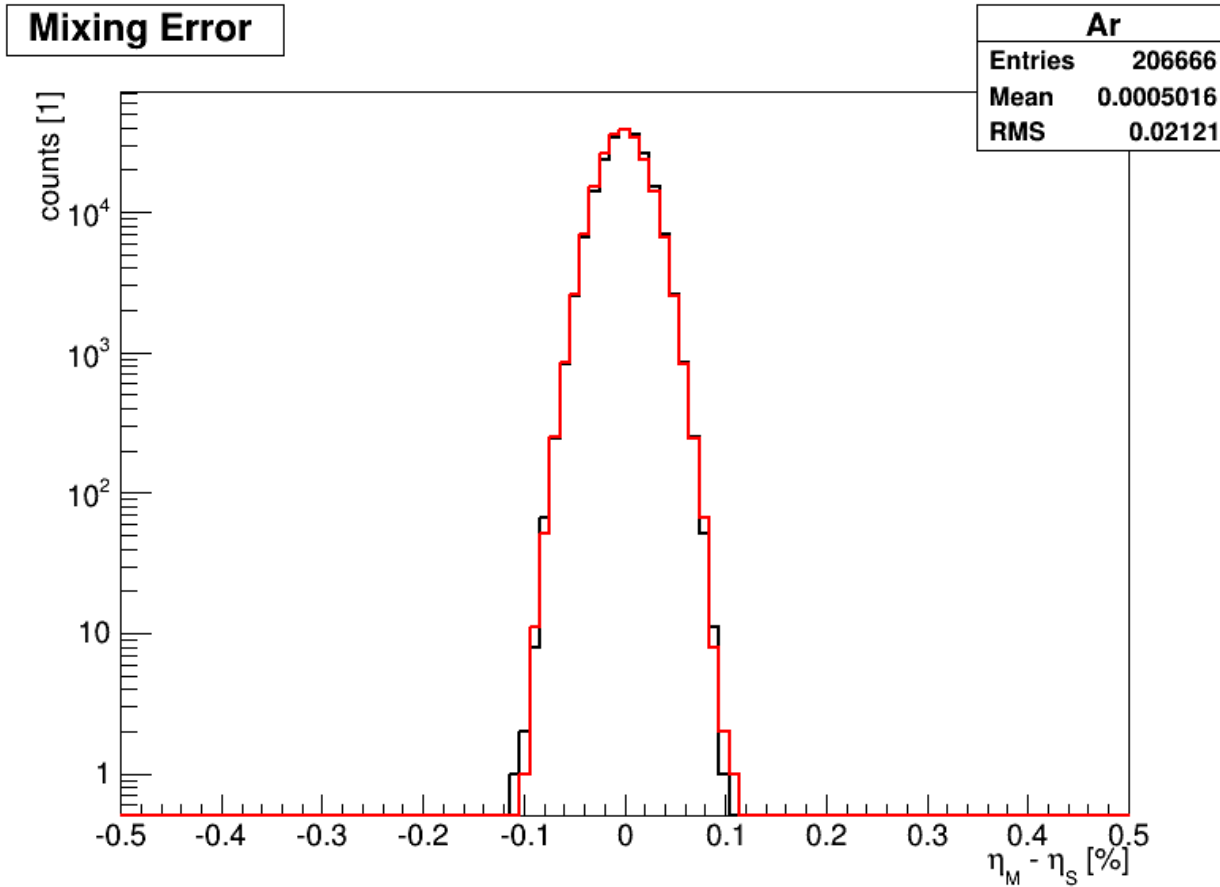
- measured pressure of the filled gases

and

Fitted mixture

Fitted mixture from comparison with simulation.

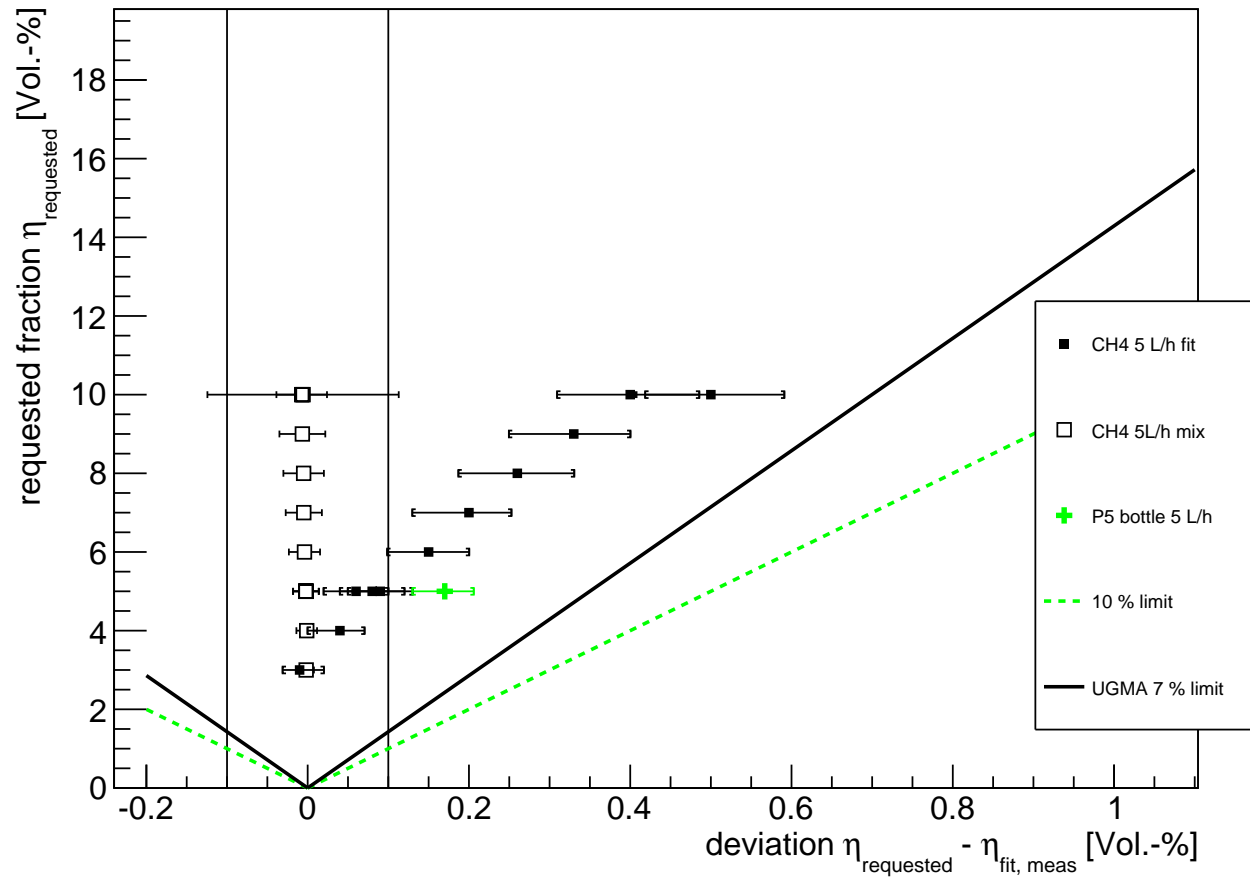
Direct Flow mixing



Ar:CH₄ 90:10 @ 10 $\ell_n/h \Rightarrow$ RMS of all lines < 0.03 vol.-%

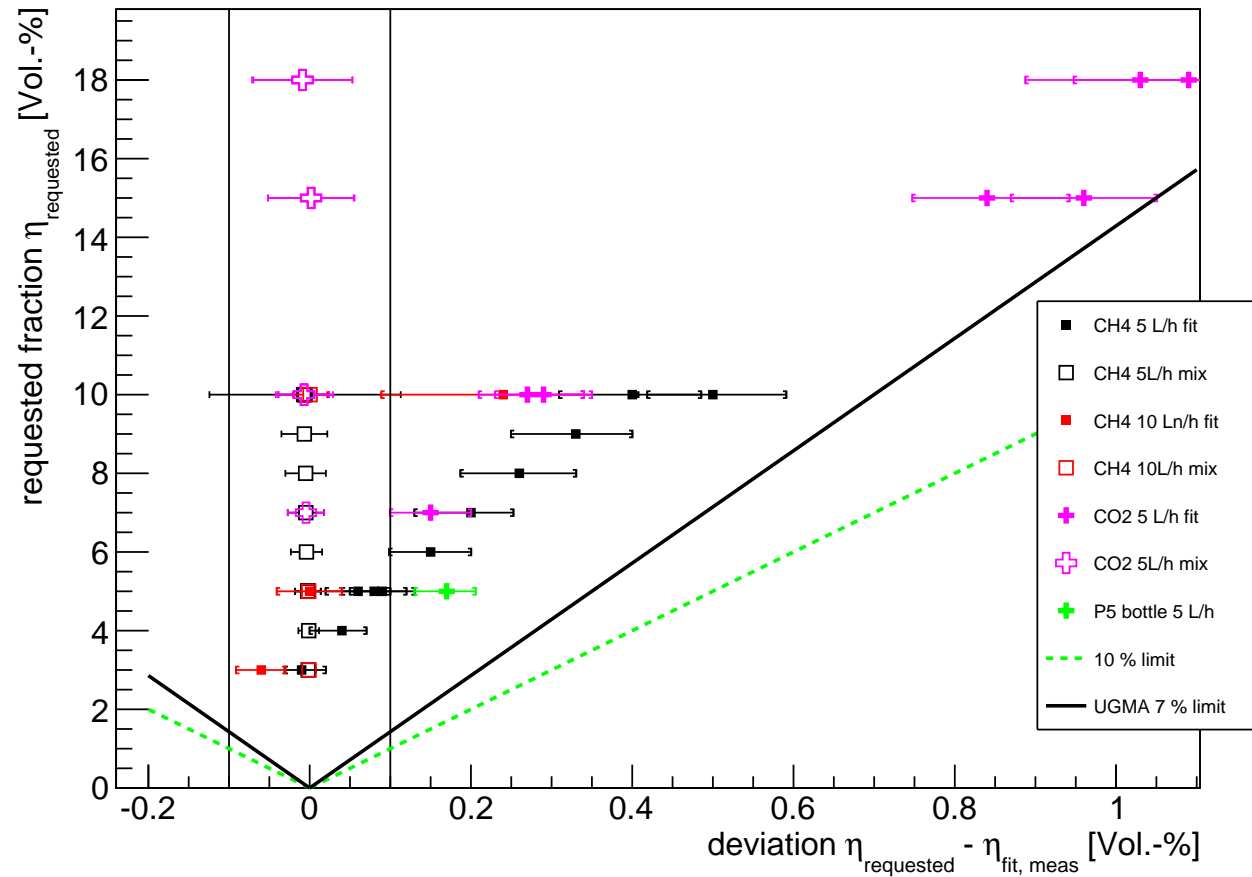
Performance for Ar CO₂ and Ar CH₄

UGMA performance (direct flow mixing)

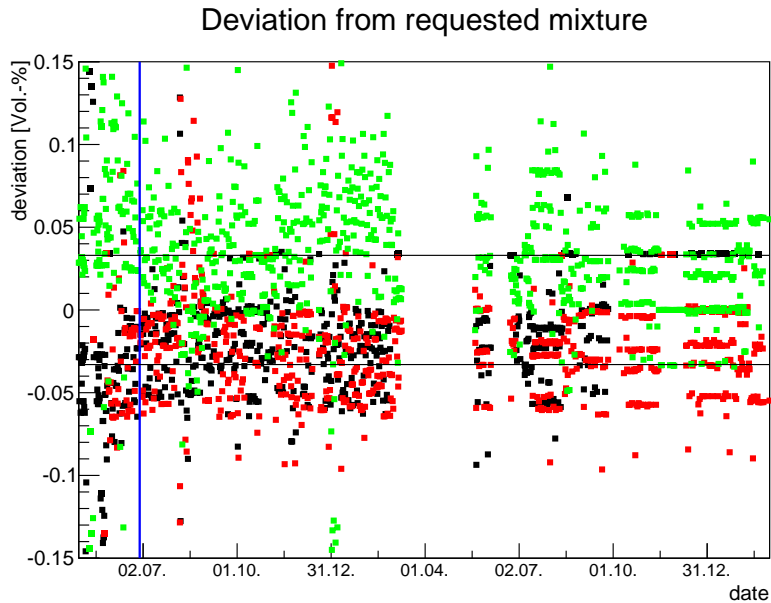


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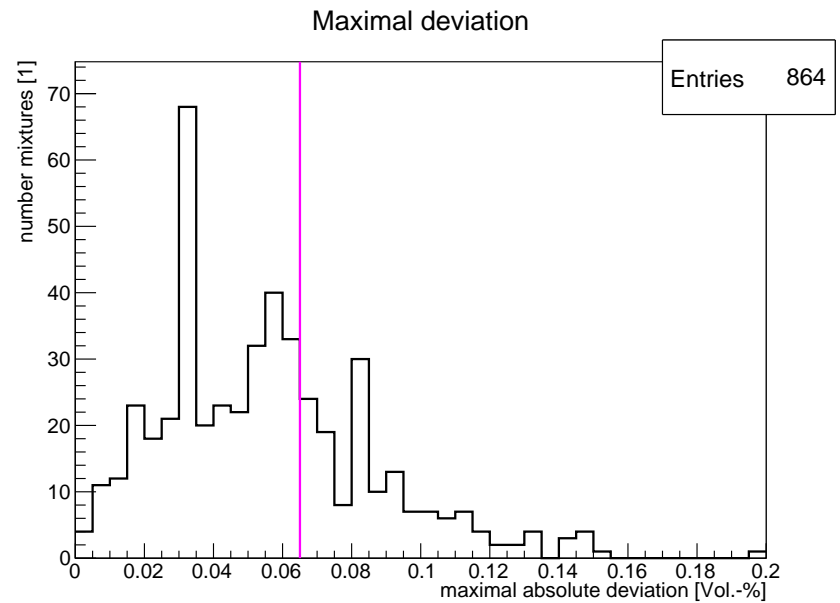


Partial pressure mixing



green - Argon, red and black - admixtures
black lines limit given by pressure sensors

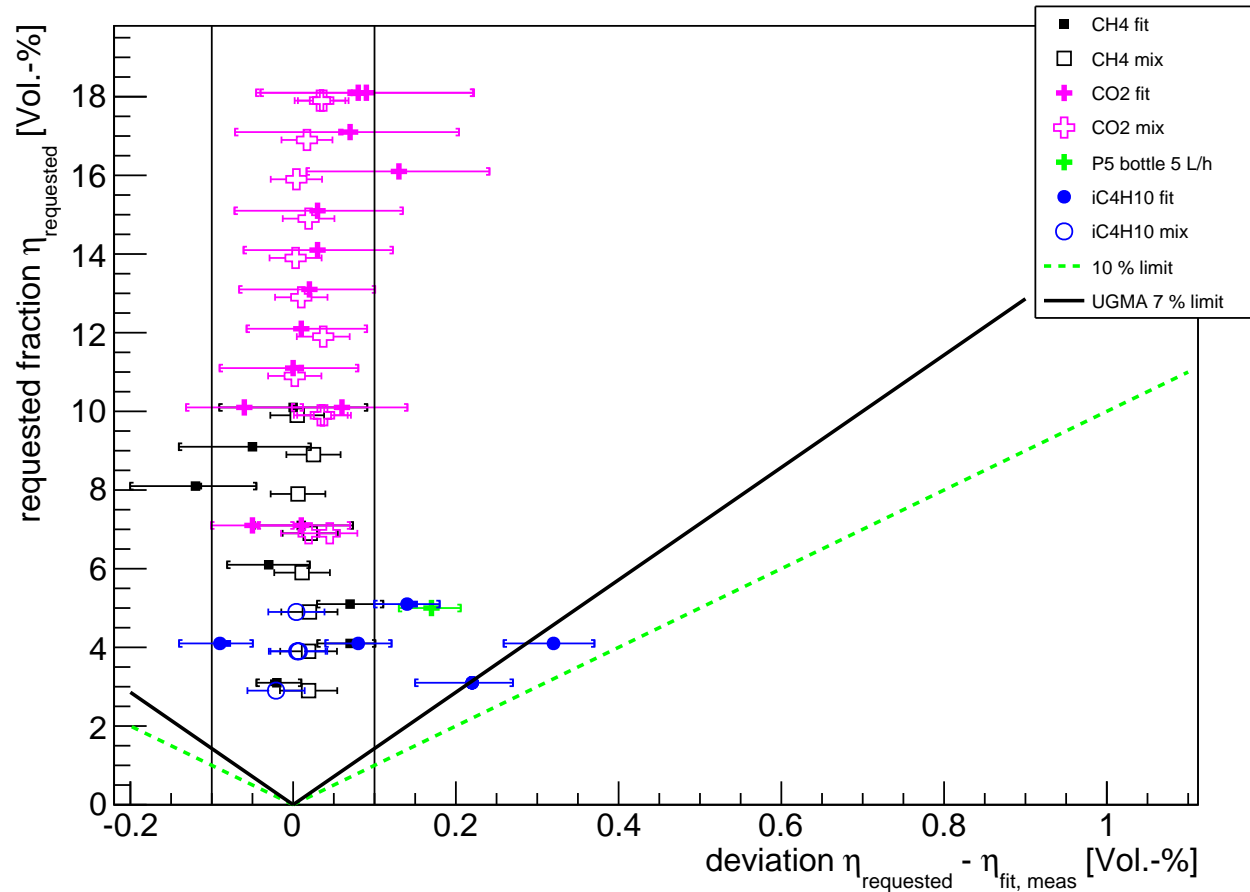
$$\Delta\eta_{\text{limit}} = \frac{\Delta p}{p_{\text{max}}} = \frac{1 \text{ mbara}}{3000 \text{ mbara}} = 0.033 \text{ vol.-%}$$



magenta 68 % limit

Performance for CO₂, CH₄ and iC₄H₁₀

UGMA performance (partial pressure mixing)



iC₄H₁₀ is rated within Magboltz with only 3 of 5 stars

Summary

Flexible Gas Mixing

Using Fitting the Mixture results (Tested within 3 - 18 vol.-%)

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$$\frac{\Delta\eta}{\eta} < 7\%$$

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- Water and Oxygen Contaminations well under control (≤ 1 ppm)

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User Friendly

- can be operated remotely via Web browser and locally via Touchscreen
- except of changing gas bottles it can run without user-interaction

Summary

System is safe

matches limitations from law & provisions:

- VDE, DIN, EU
- TA-Luft, ExSchV, BetrSichV, etc.

System is ready ...

- to do further gas studies
- to supply gas to new detectors and prototypes

Hopefully the system will be used and run over several years.
Further improvement of the system is not excluded ...

You can always make things better!

Jochen Steinmann
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