

SEMINAR

14th June 2013 – 12:30 CFEL-bldg. 99, seminar room IV (O1.111)

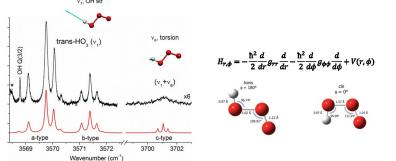
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Spectroscopy of the hydroxyl radical and HO-(O₂)_n clusters in helium nanodroplets

We are exploring the low temperature (0.4 K) *in-situ* association reactions between the hydroxyl radical (OH) and other small molecules, such as O_2 , C_2H_2 and H_2O . For example, upon sequential pick-up and solvation of OH and O_2 , the infrared spectrum in the 3500-3700 cm⁻¹ region reveals features that are assigned to the v_1 (OH stretch) fundamental and v_1+v_6 (OH stretch plus torsion) combination bands of the covalently bound *trans*-HOOO isomer. Despite the characteristic low temperature and rapid cooling of helium droplets, there is no evidence for the formation of a previously predicted, weakly bound OH-O₂ van der Waals complex, which implies the

absence of a kinetically significant barrier in the entrance channel of the reaction. There is also no spectroscopic evidence for the formation of *cis*-HOOO, which is predicted by theory to be nearly isoenergetic to the *trans* isomer. Stark spectroscopy of the *trans*-HOOO species provides



vibrationally averaged dipole moment components that qualitatively disagree with predictions obtained from CCSD(T) computations at the equilibrium, planar geometry, indicating a floppy complex undergoing significant large-amplitude motion. A Hamiltonian in internal bond-angle coordinates is derived to account for this large-amplitude motion, and the expectation values of the inertial dipole moment components are compared to those obtained from the Stark spectra, revealing a fundamental error in the underlying single-reference CCSD(T) potential energy surface.

Host: Jochen Küpper, CFEL Molecular Physics Seminar