



How biological catalysts make and break bonds - What X-ray spectroscopy can teach us...

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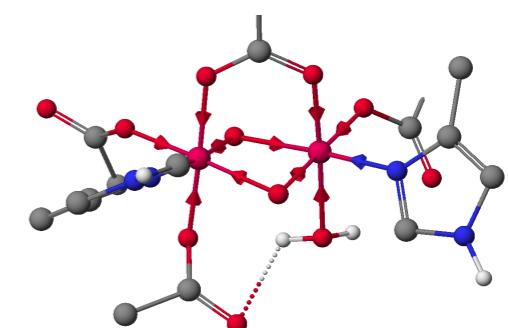
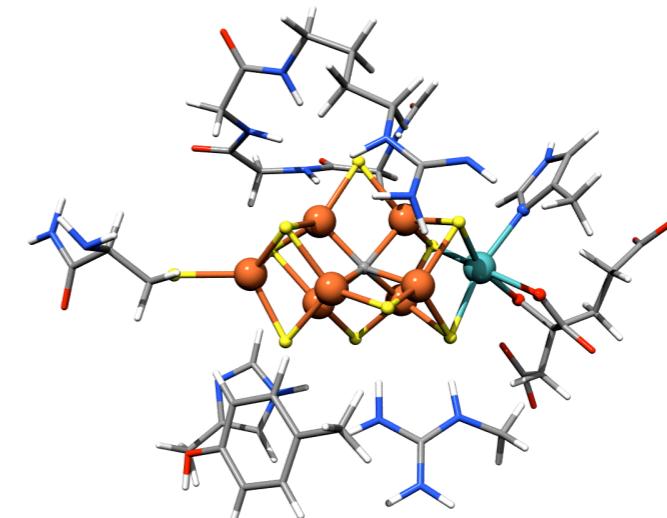
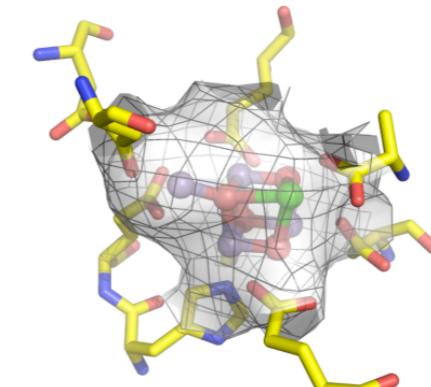
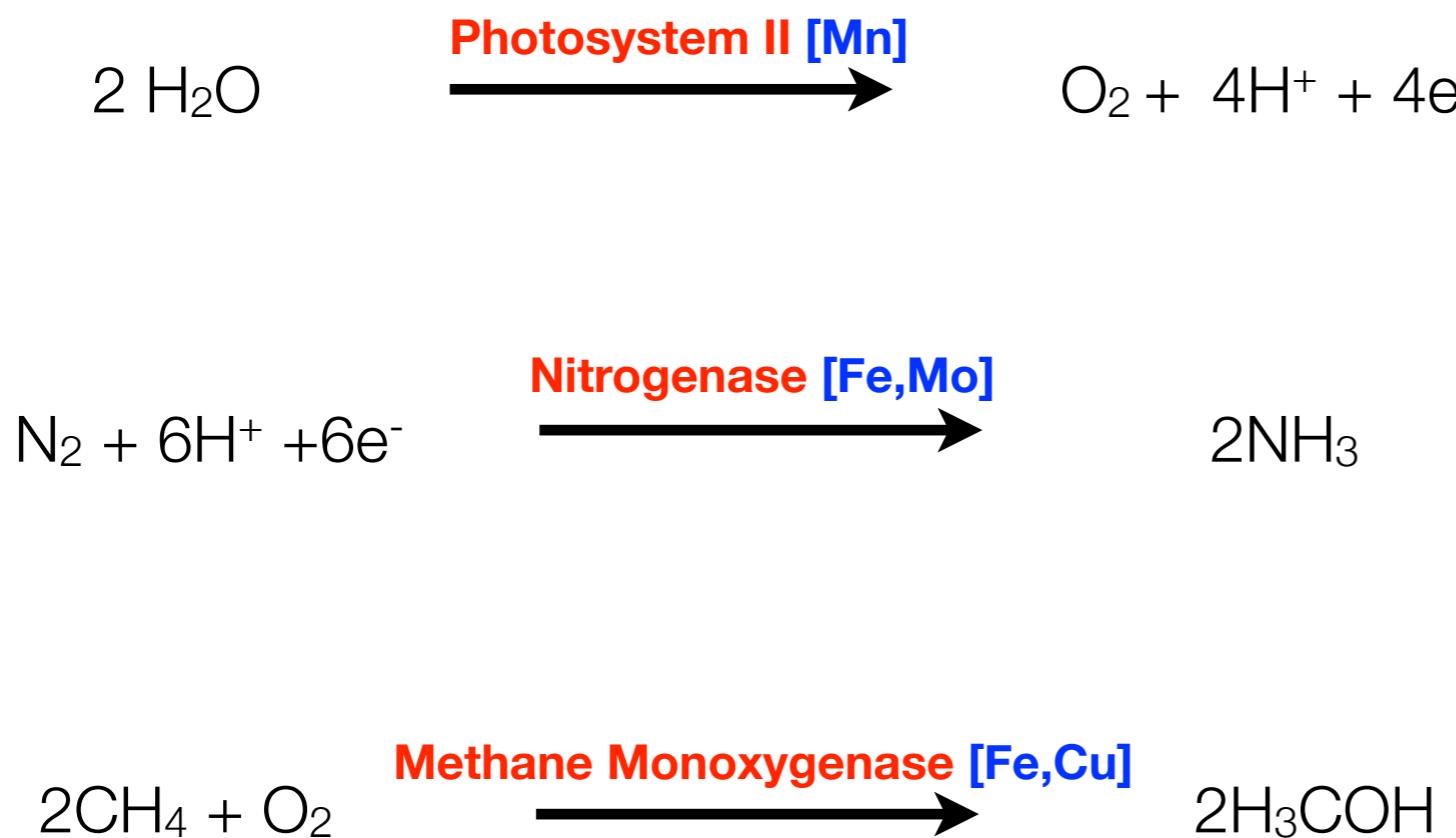


EUCALL Workshop: Biology at Advanced Laser Light Sources
30 November - 01 December 2017

Venue: European X-ray Free-Electron Laser Facility
Holzkoppel 4, 22869 Schenefeld, Germany

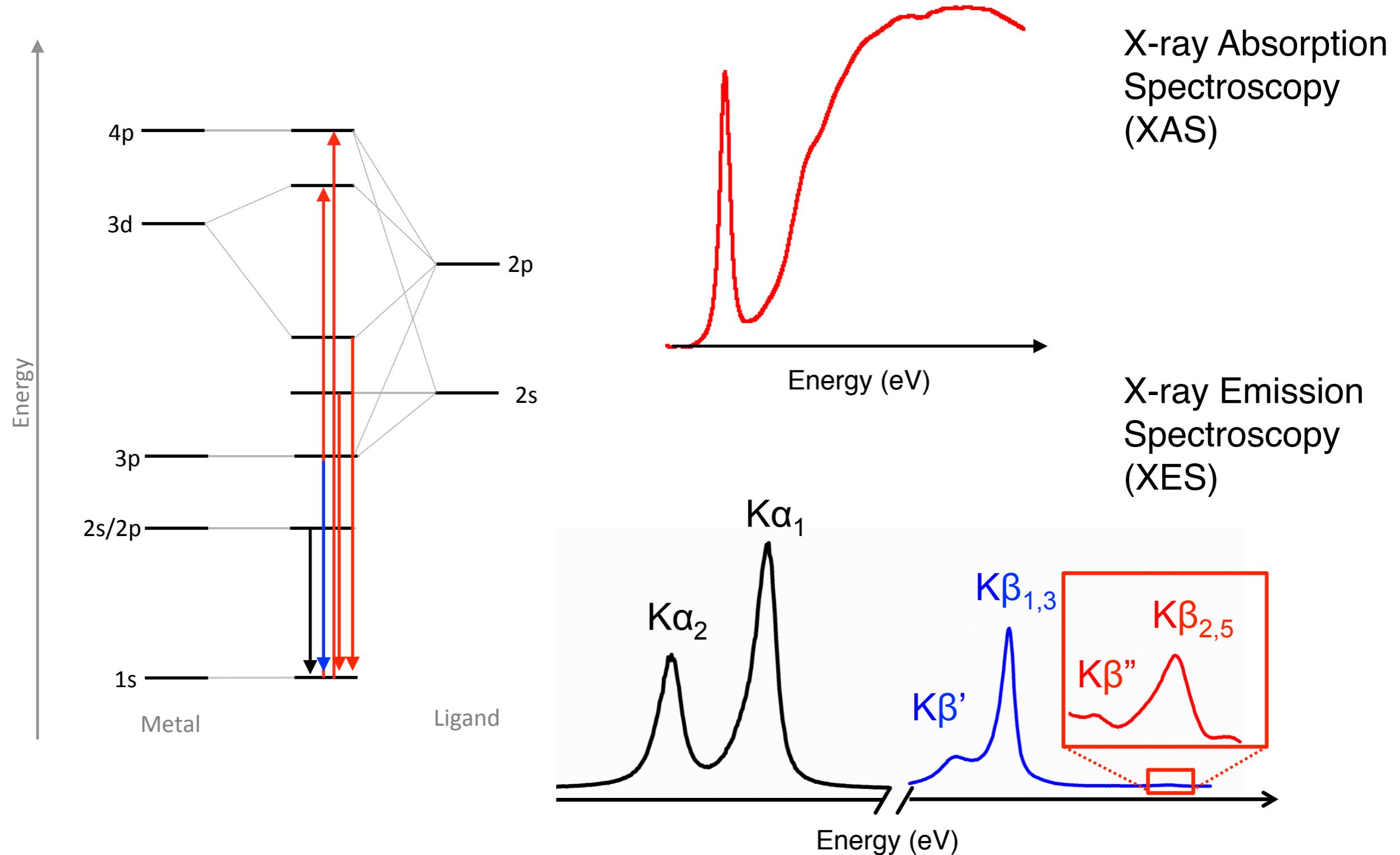
Motivation: From Biological to Chemical Catalysis

Taking inspiration from Nature....

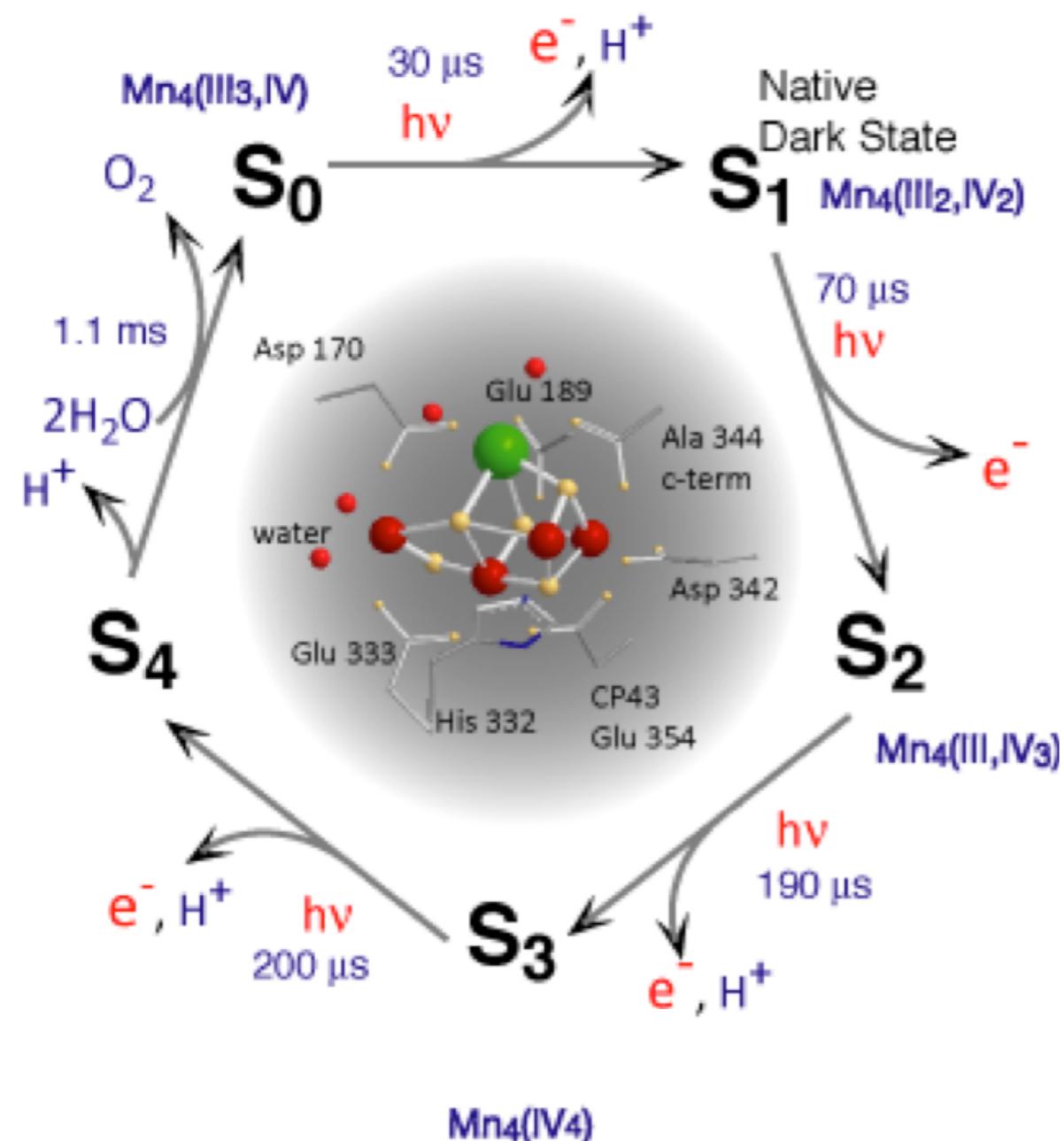


Goal: To obtain fundamental mechanistic insights that can be more broadly applied.

X-ray Spectroscopy



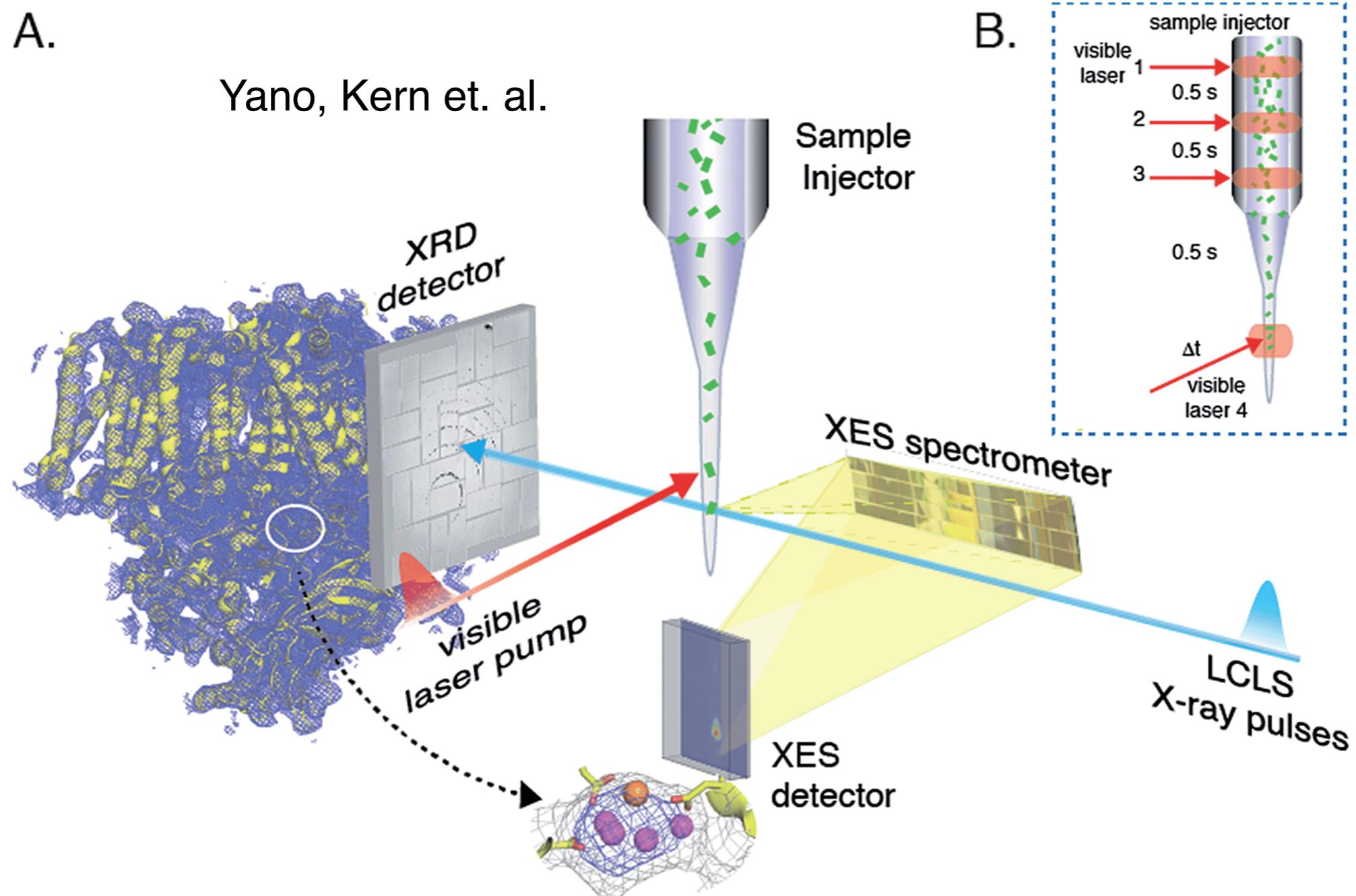
PSII: Mechanistic Understanding from Spectroscopy



- The OEC of PSII by far the most studied biological catalyst at XFELs (and elsewhere)
- Critical bond making and breaking processes on the order of micro to milliseconds
- Importance of XFEL due to damage rates
- S-states can be generated by laser flashing.

Adapted from:
Yano et. al.

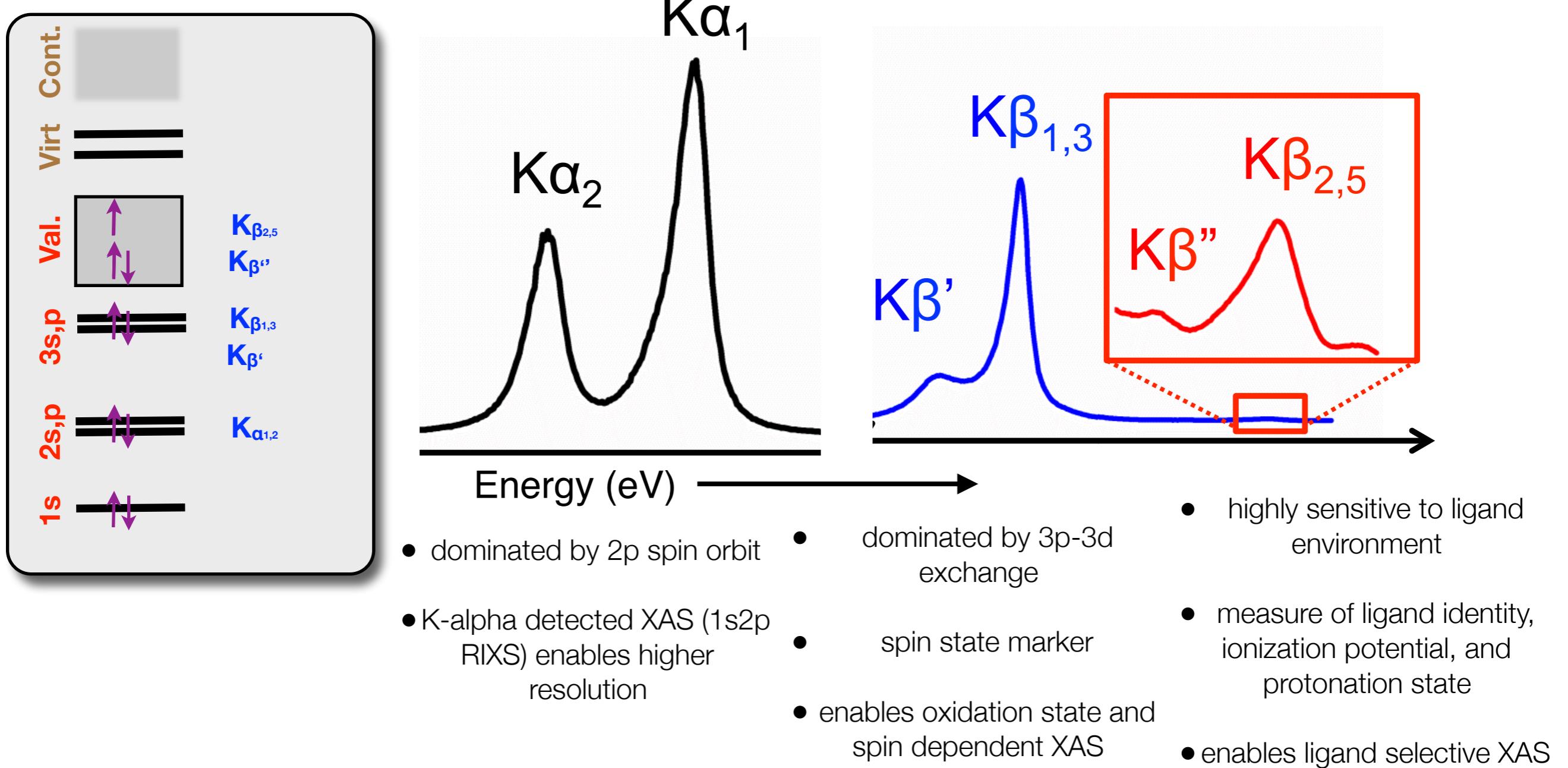
PSII: Combined XRD and XES



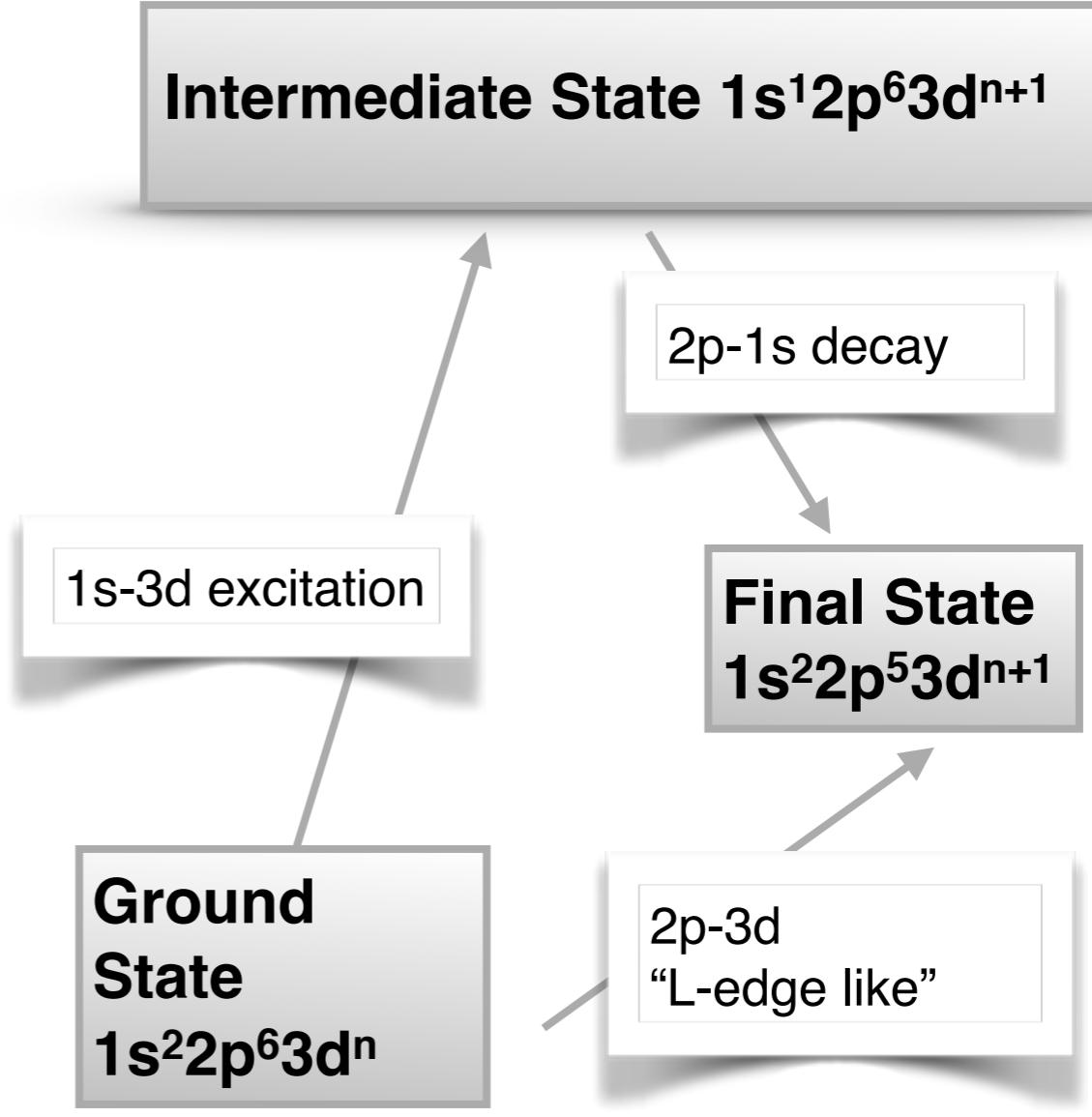
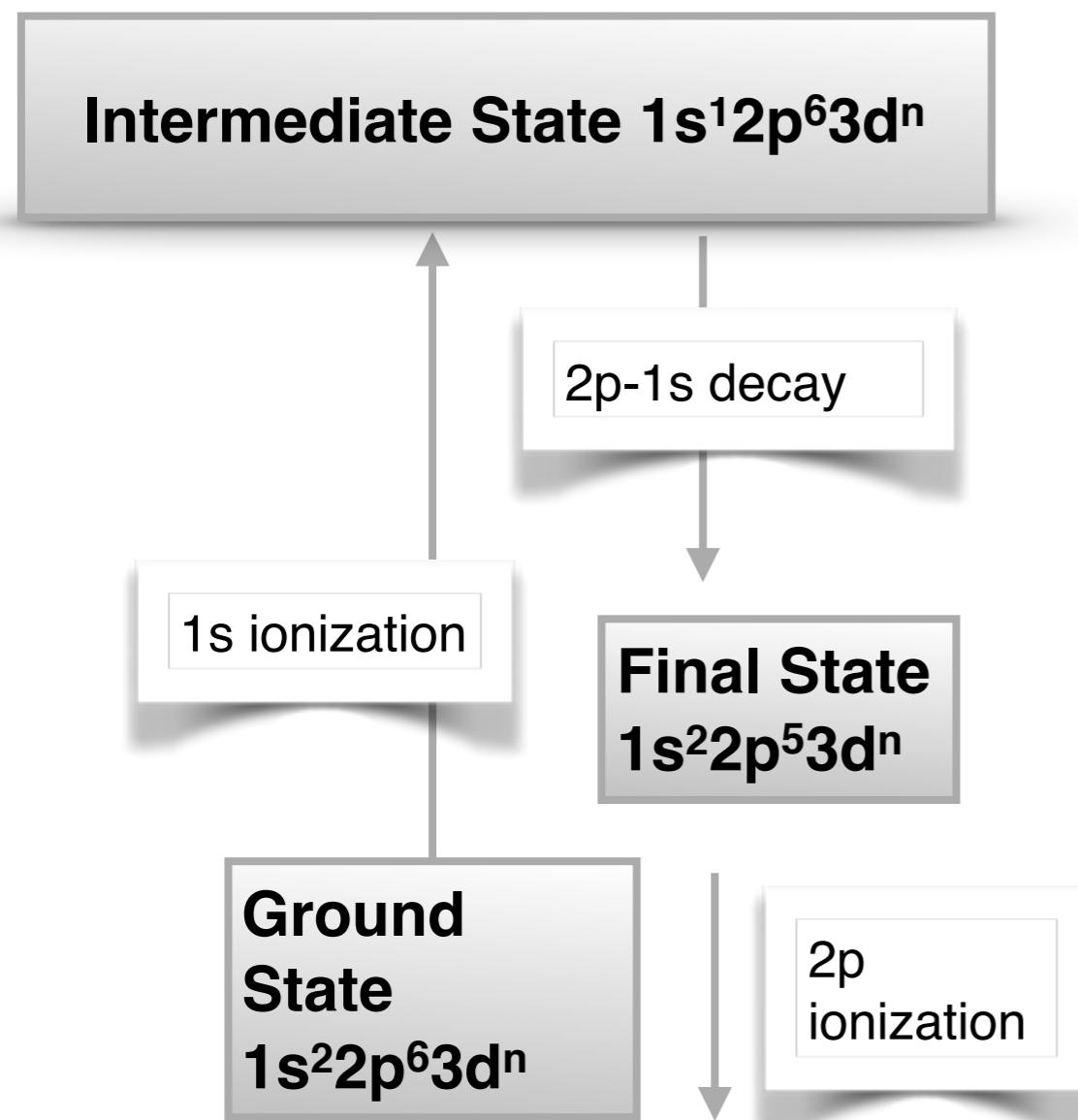
*Applications to other biological catalysts??
But first, more on XES....*

X-Ray Emission Spectra

N. Lee, T. Petrenko, U. Bergmann, F. Neese, S. DeBeer, J. Am. Chem. Soc., 2010, 132, 9715-9727.



Non-resonant $K\alpha$ XES vs. Resonant $K\alpha$ XES

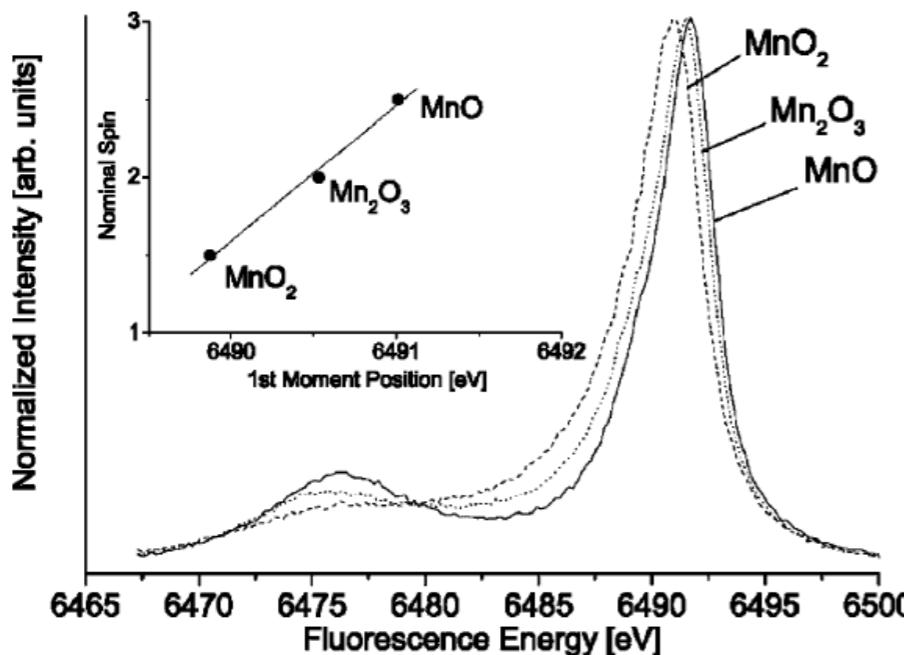
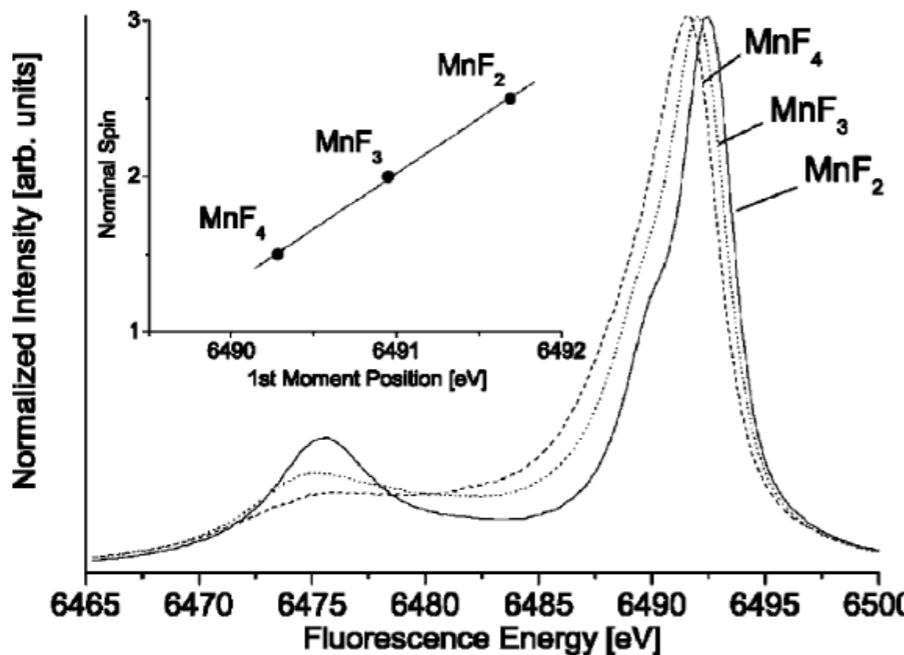


- incident beam at fixed energy above absorption edge
- core excitations into the continuum
- spectra effectively independent of incident energy

- incident beam tuned to excite into the absorption edge of interest
- excitations into bound states
- spectra highly energy dependent

K-Beta Mainlines as a Probe of Spin/Oxidation State

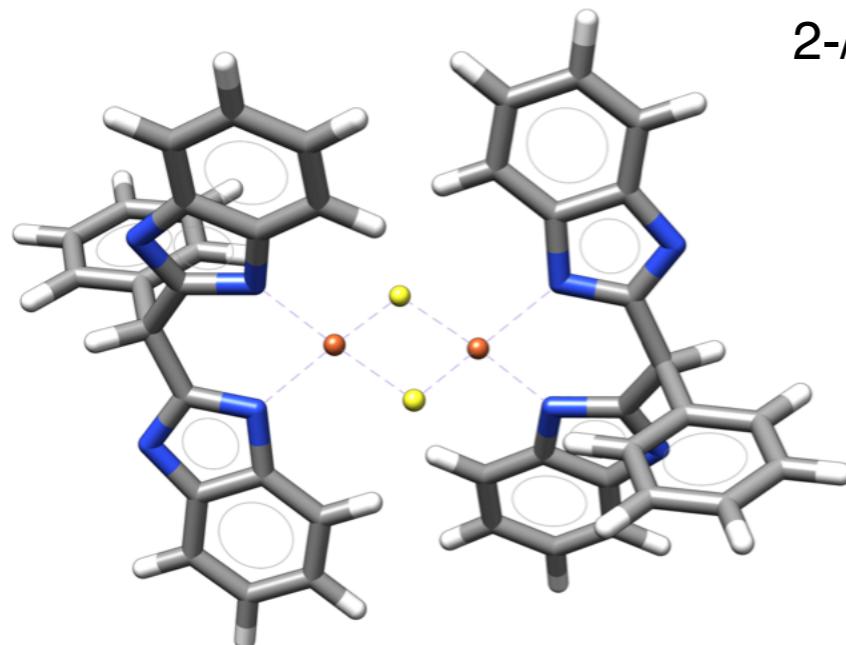
Mn(IV) d³, Mn(III) d⁴, Mn(II) d⁵



- * K β lines dominated by 3p-3d exchange interaction
- * K $\beta_{1,3}$ and K β' move closer together with decreasing spin state (i.e. as 3p-3d exchange interaction decreases)
- * K β lines reflect number of unpaired d electrons
- * Can be applied to any metal.
- * Has been used to “establish” damage free XRD data collection....

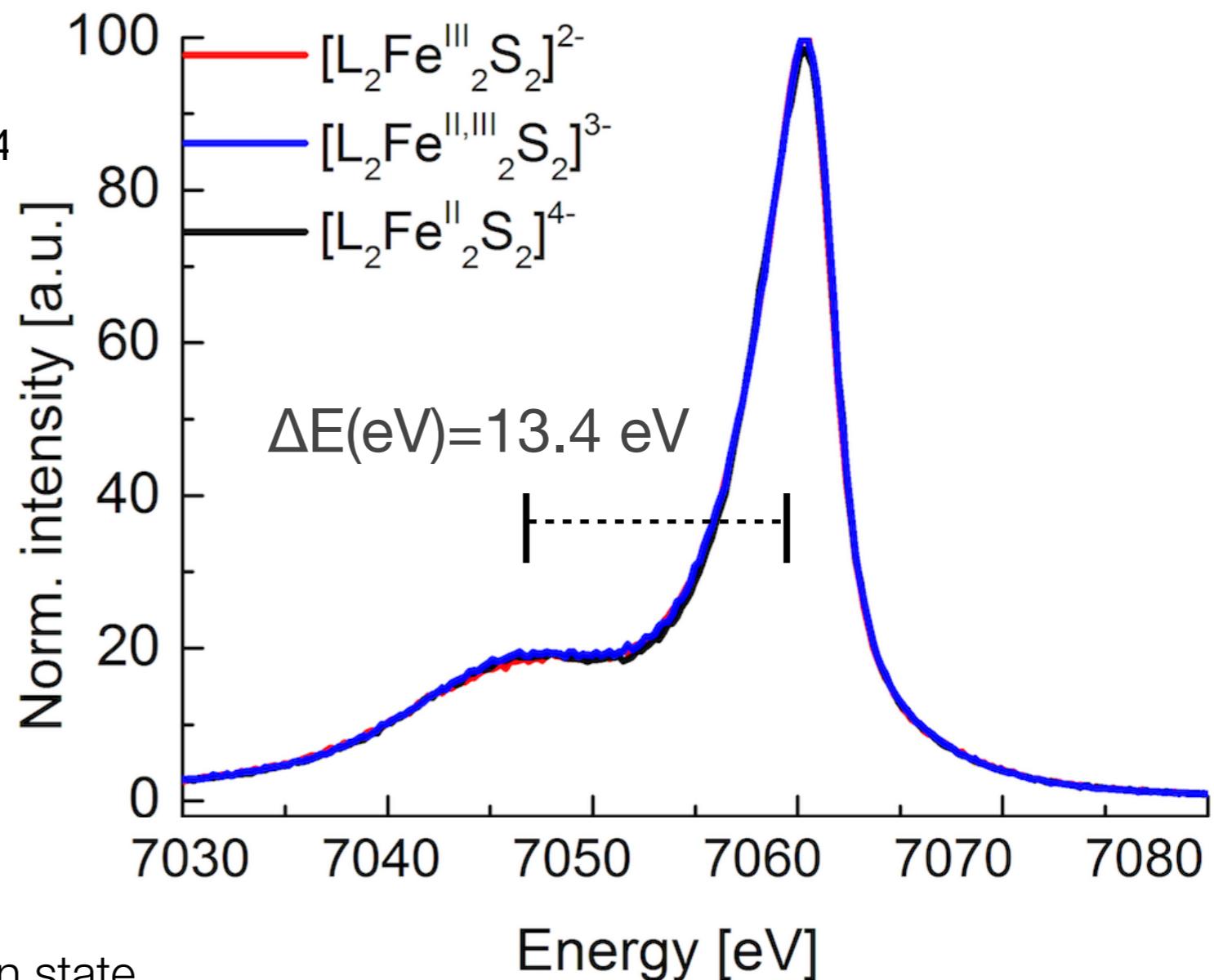
Lessons from Model Studies

In Collaboration with
F. Meyer (Göttingen)



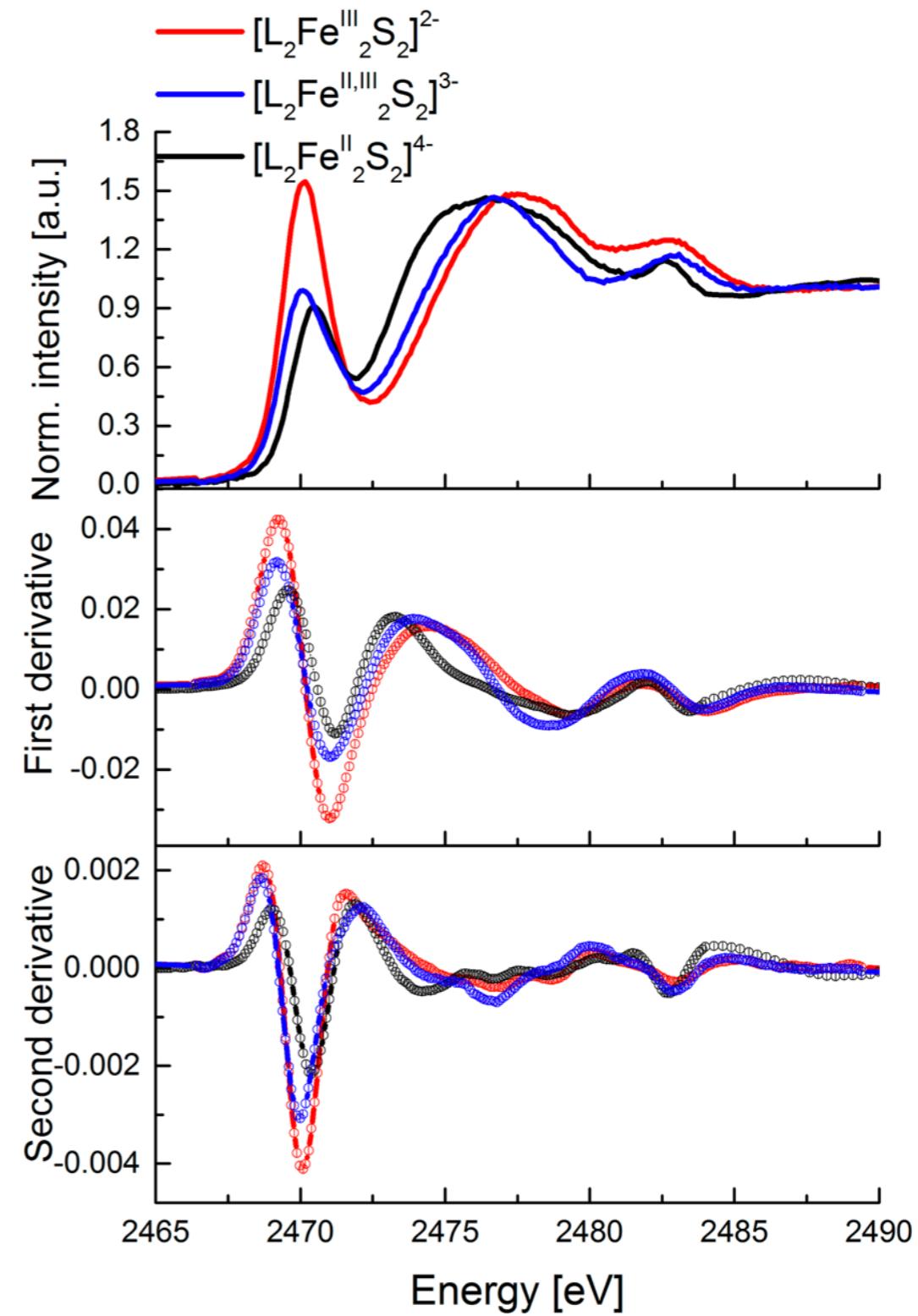
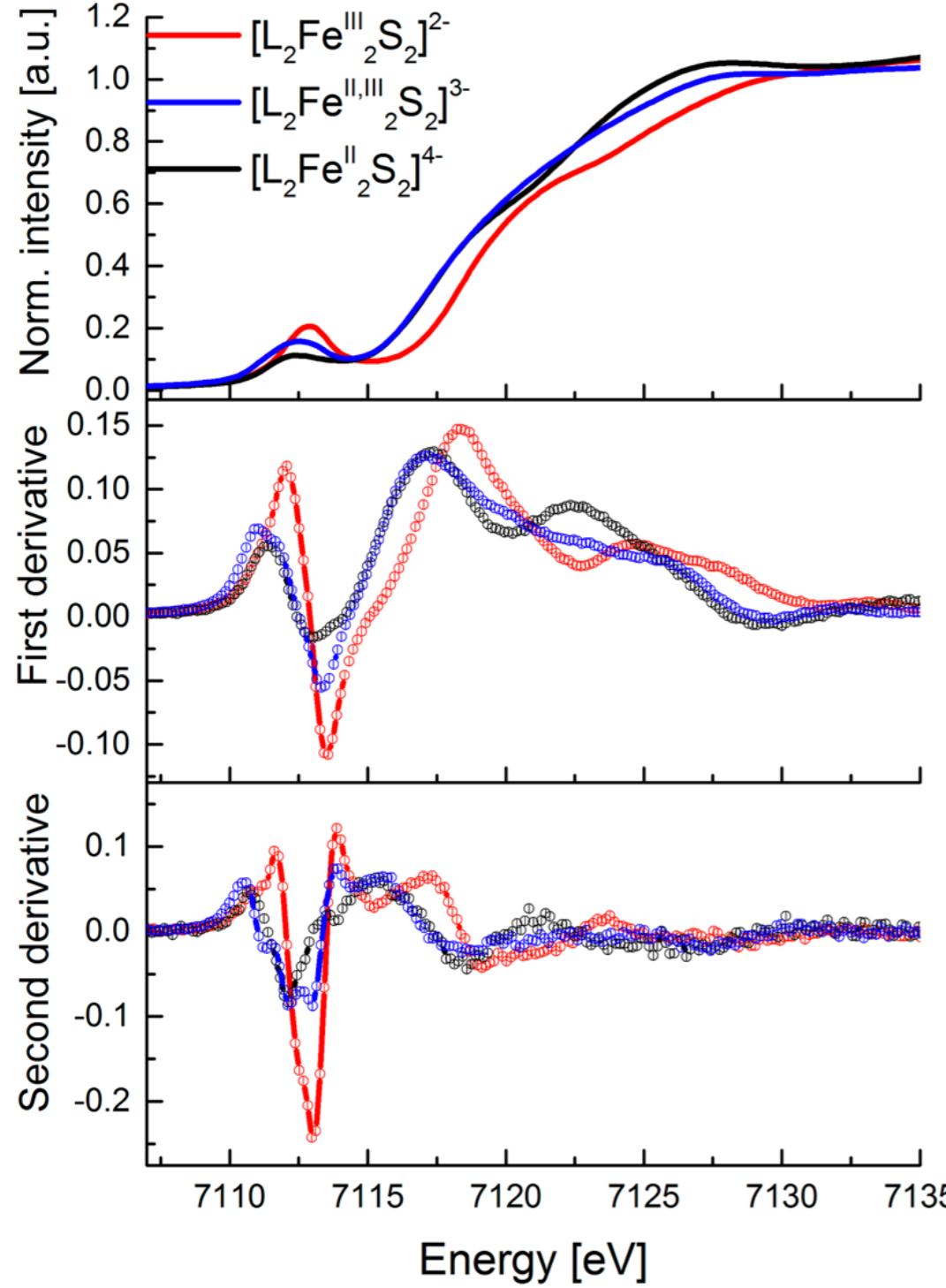
2-3-4

- A. Albers *et al.*, *Angew. Chem. Int. Edit.*, **2011**, *9191* (39)
A. Albers *et al.*, *J. Am. Chem. Soc.*, **135**, *2013*, *1704*.
J. Kowalska *et al.*, *Inorg. Chem.*, **55**, *2016*, *4485*.



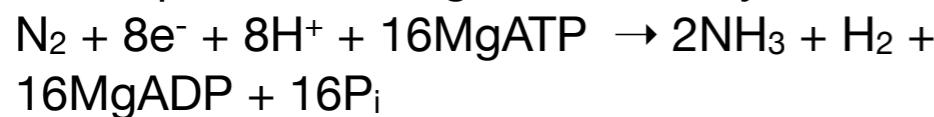
- Contributions due to change in oxidation state and covalency completely cancel!
- A cautionary note on assigning oxidation states from KB mainlines!***

Fe K- and S K-edges clearly different...

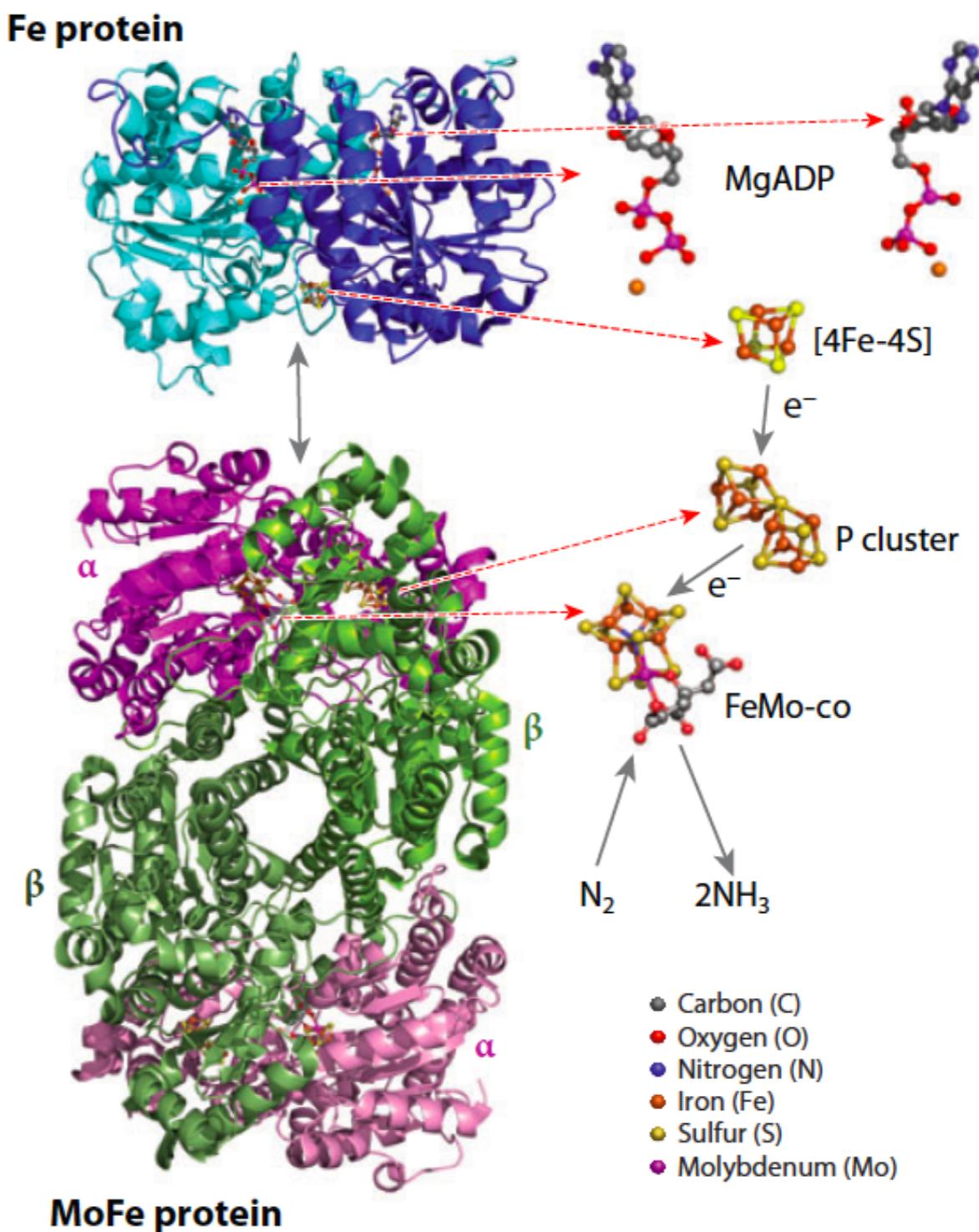


Nature's Machinery for N₂ Activation

Mo-dependent nitrogenases catalyze....



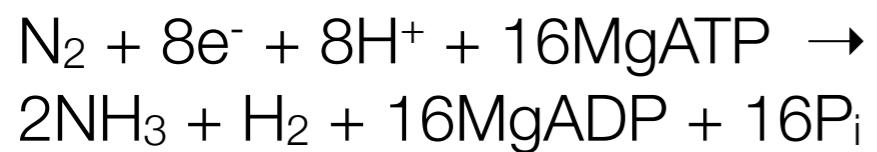
Multicomponent protein (Fe protein (64 kDa) and MoFe protein (250 kDa))



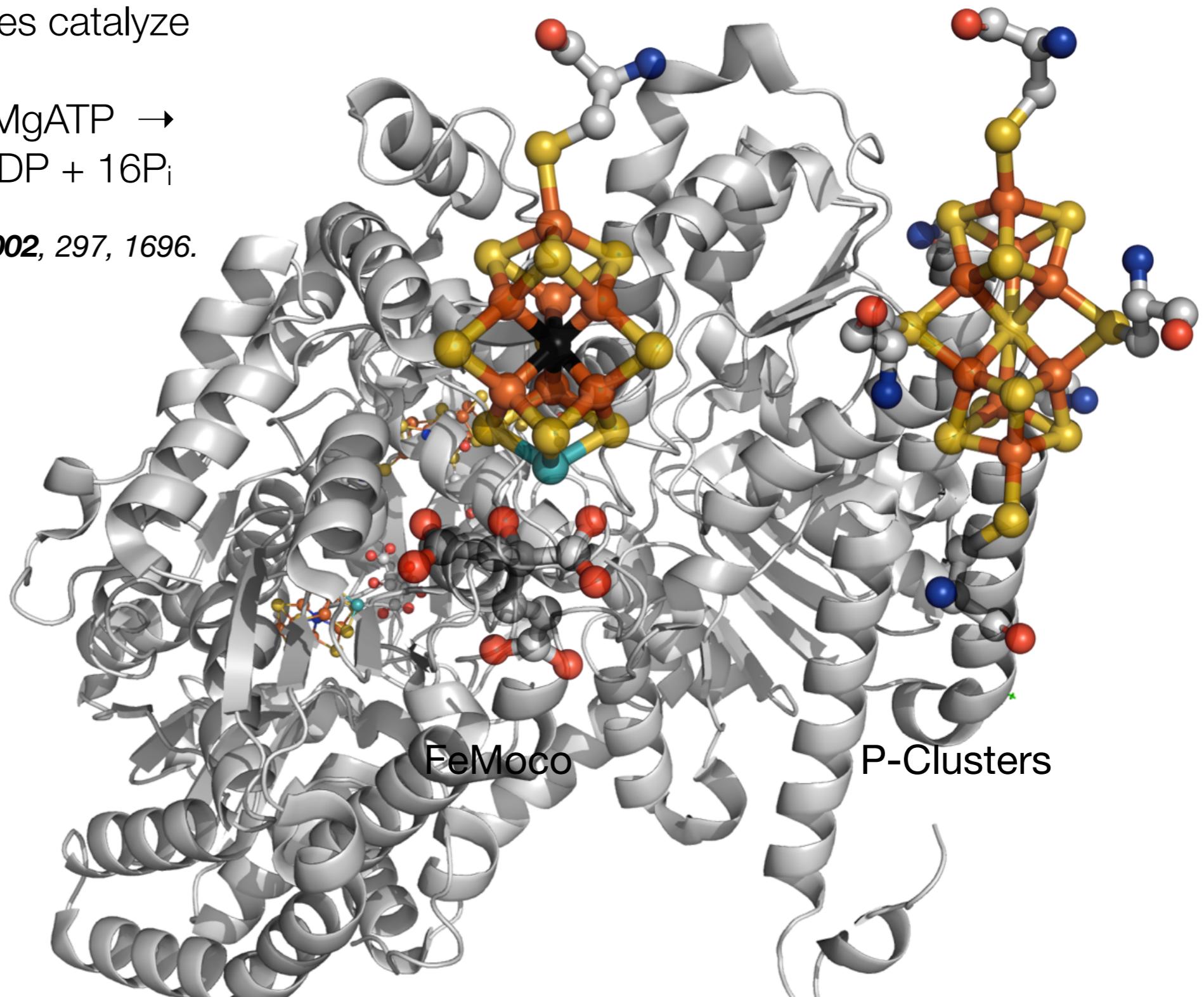
Seefeldt, Hoffman, Dean, Ann. Rev. Biochem. 2009

MoFe Nitrogenase

- Nitrogenase enzymes catalyze the reaction:

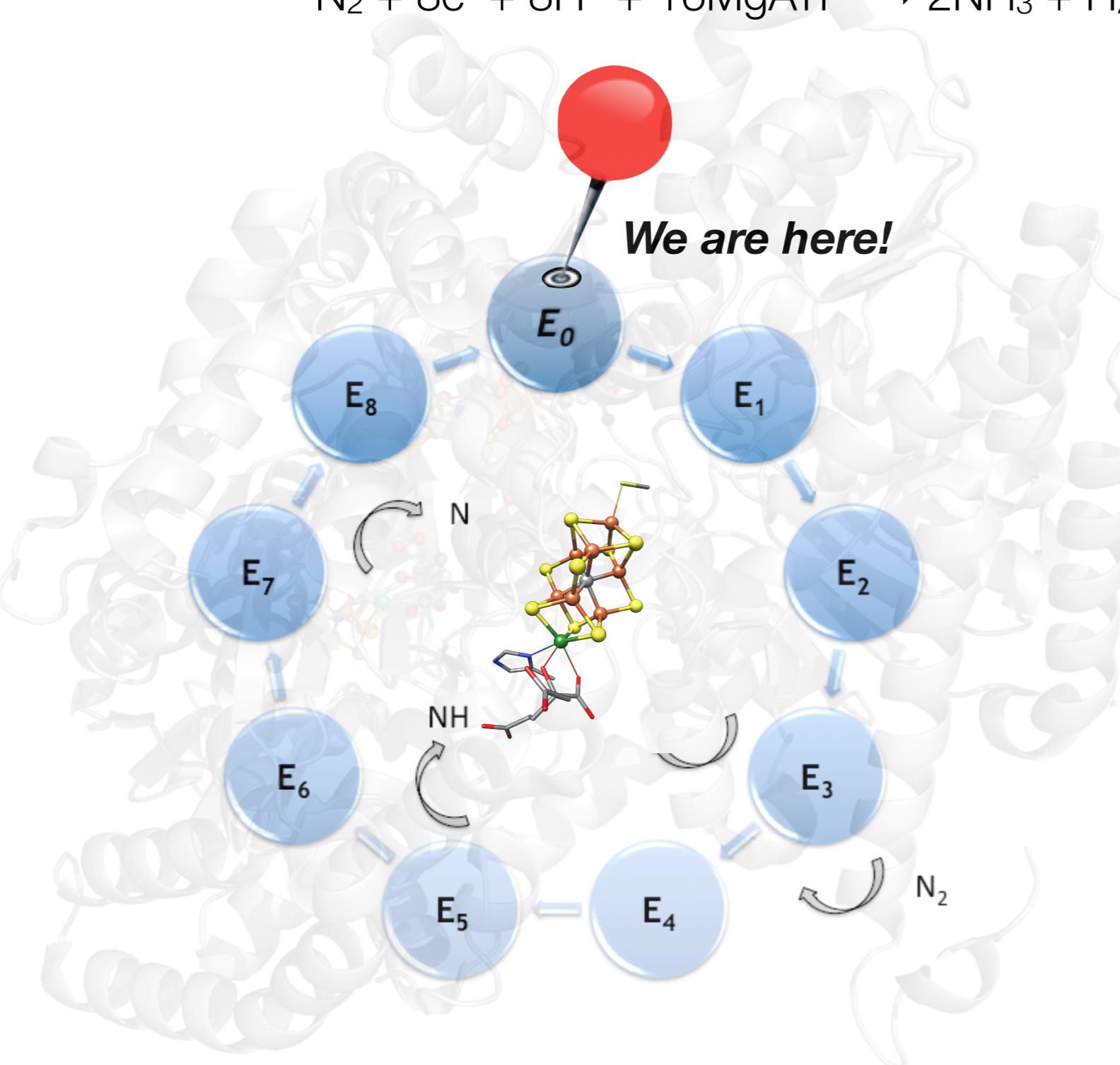


Einsle et al, Science, 2002, 297, 1696.



Biological N₂ Reduction: Open Questions...

- Molybdenum Nitrogenase enzymes catalyze the reaction:
$$\text{N}_2 + 8\text{e}^- + 8\text{H}^+ + 16\text{MgATP} \rightarrow 2\text{NH}_3 + \text{H}_2 + 16\text{MgADP} + 16\text{Pi}$$



An understanding of the molecular level mechanism requires a detailed understanding of the electronic structure of the cofactors at each stage in the cycle.

Requires knowledge of:

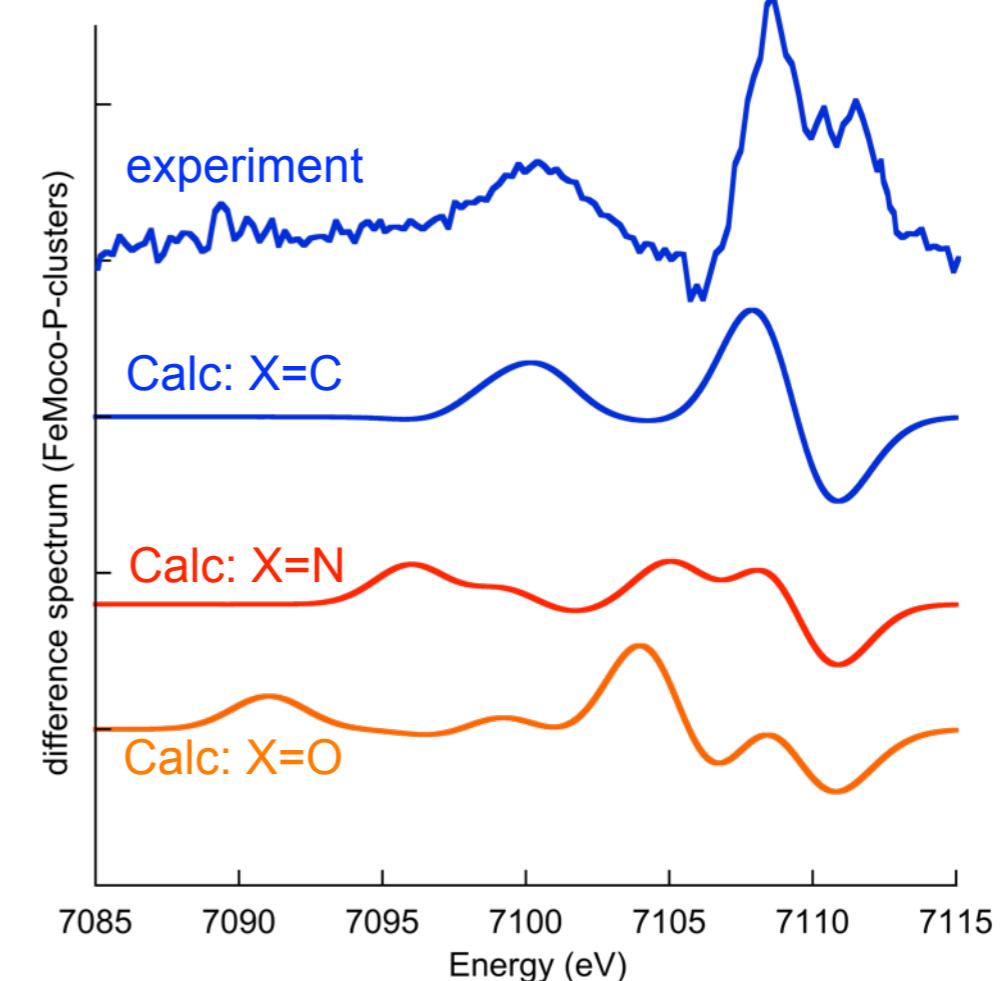
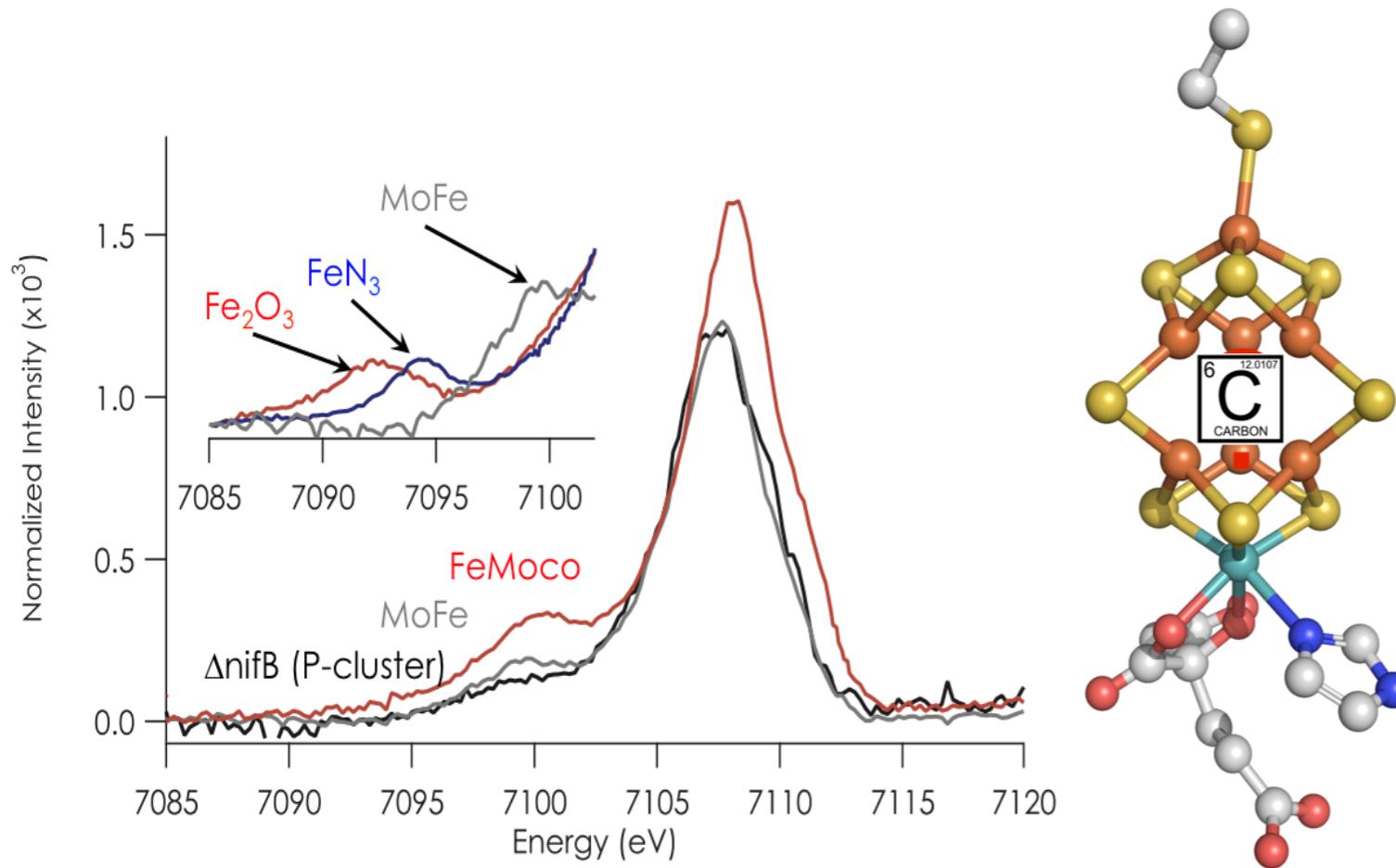
- Atomic composition
- Charge distribution
- Oxidation state distribution
- Magnetic coupling

The role of X-ray Spectroscopy...

Identifying the Central Atom in FeMoco

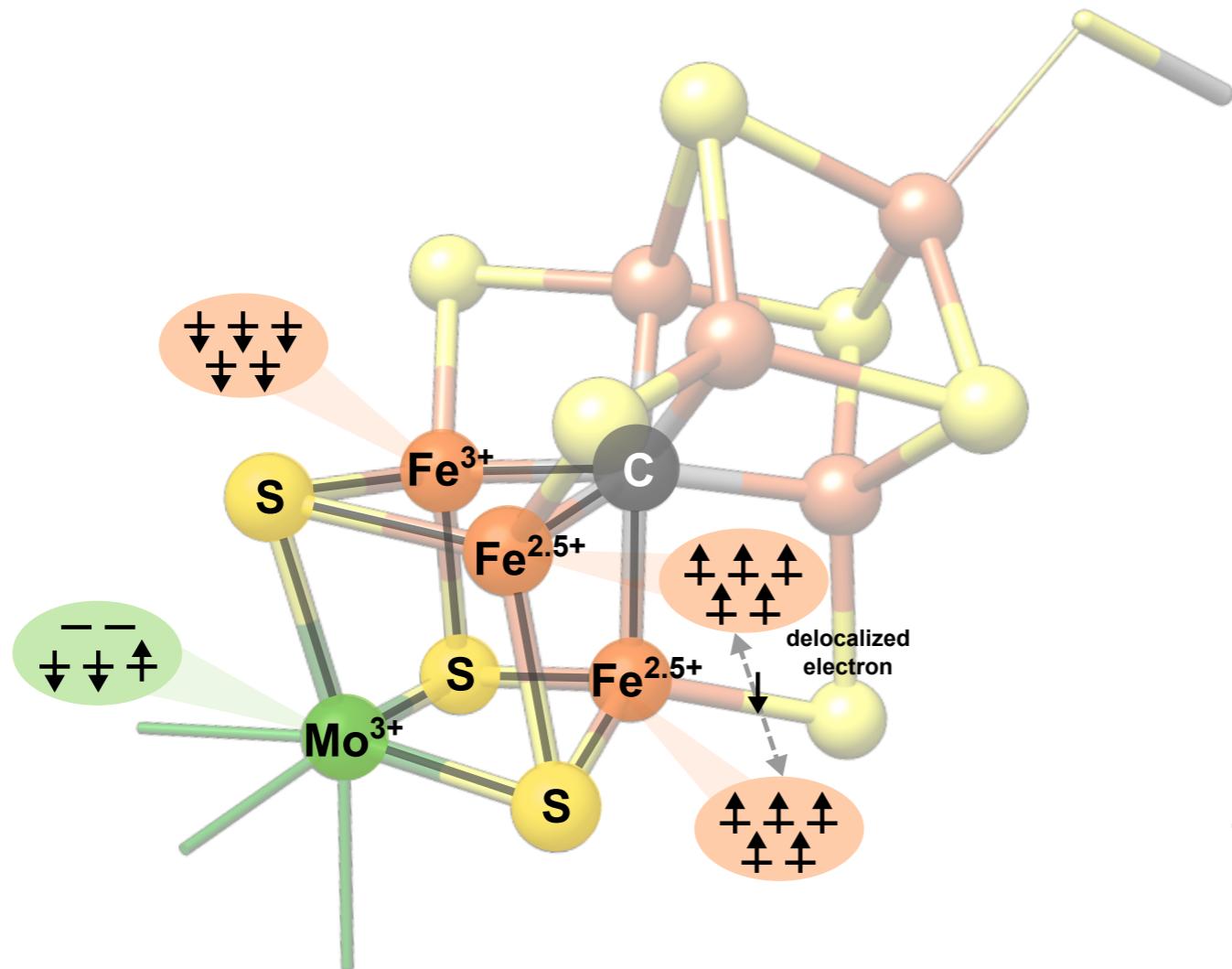
Valence-to-Core:
X-Ray Emission

Quantum
Chemistry



Lancaster, K.M.; Roemelt, M.; Ettenhuber, P.; Hu, Y.; Ribbe, M.W.; Neese, F.; Bergmann, U.; DeBeer, S. *Science* **2011**, 334, 974-977.

Current Picture of the E₀ state of FeMoco



- ✓ Based on:
 - ✓ VtC XES
 - ✓ Mo HERFD XAS
 - ✓ Fe HERFD XAS
 - ✓ Mössbauer
 - ✓ Fe/Mo L-edge XAS + XMCD
 - ✓ DFT correlated to spectroscopy
- ✓ FeMoco: [MoFe₇S₉C]¹⁻: 4Fe(III)3Fe(II)Mo(III) g.s.

Lancaster, K.M.; Roemelt, M.; Ettenhuber, P.; Hu, Y.; Ribbe, M.W.; Neese, F.; Bergmann, U.; DeBeer, S. *Science* **2011**, 334, 974-977.

Lancaster, K. M., Hu, Y., Bergmann, U., Ribbe, M. W., and DeBeer, S. *JACS* **2013** 135, 610-612.

Bjornsson, R., Lima, F. A., Spatzal, T., Weyhermueller, T., Glatzel, P., Bill, E., Einsle, O., Neese, F., and DeBeer, S. *Chemical Science* **2014** 5, 3096-3103.

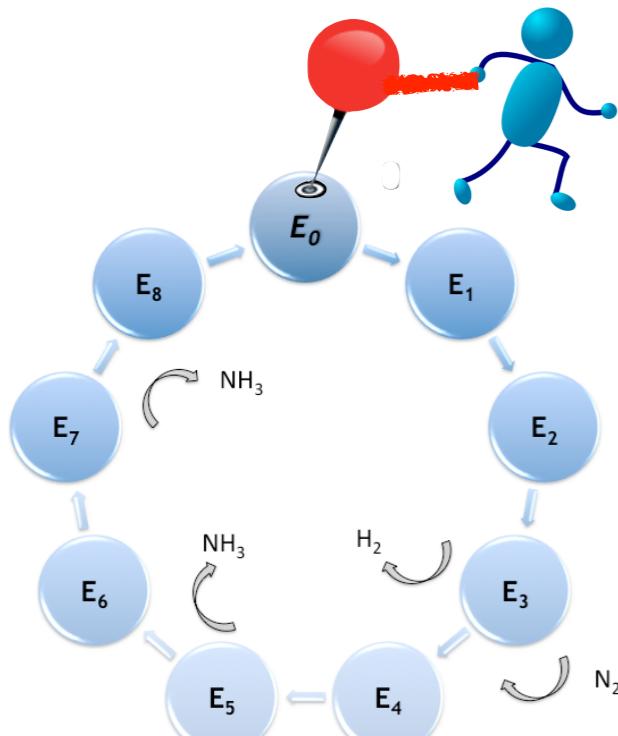
Rees, J. A., Bjornsson, R., Schlesier, J., Sippel, D., Einsle, O., and DeBeer, S. *Angewandte Chemie* **2015**, 54, 13249-13252

Bjornsson, R.; Neese, F.; Schrock, R. R.; Einsle, O.; DeBeer, S. *J. Biol. Inorg. Chem.* **2015**, 20, 447-460.

Bjornsson, R., Neese, F., and DeBeer, S., *Inorganic Chemistry* **2017**, 56, 1470.

Rees, J.A.; Bjornsson, R.; Kowalska, J.K.; Lima, F. A.; Schlesier, J.; Sippel , D.;Weyhermüller, T. Einsle, O.; Kovacs, J.A.;DeBeer, S. *Dalton Transactions*, **2017**, 46, 2445.

Moving Beyond E_0 : The Challenges in Trapping Intermediates

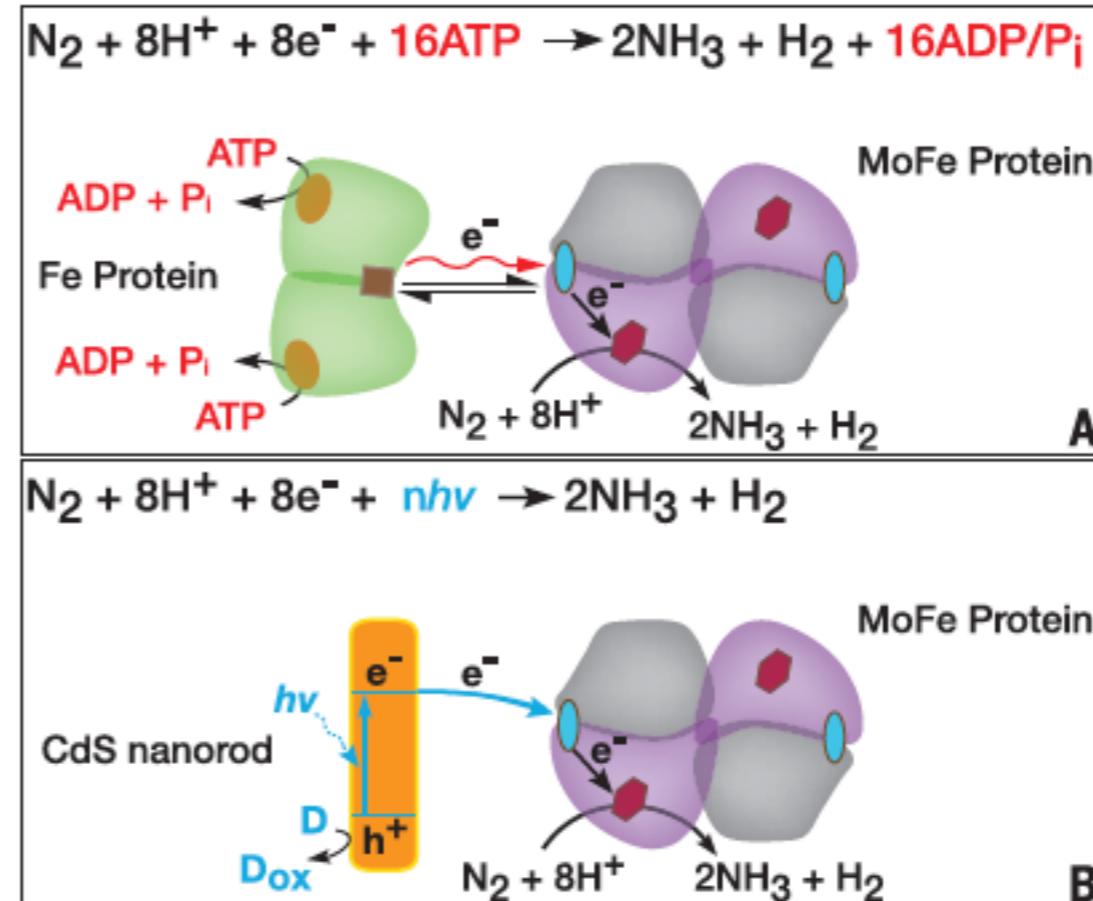


- Depends strongly on electron e-flux conditions (ratio of FeP to MoFe)
- Until very recently MoFe protein had been thought to require native reductase
- CdS produces uncontrolled cascade of electrons....

Light-driven dinitrogen reduction catalyzed by a CdS:nitrogenase MoFe protein biohybrid

Katherine A. Brown,¹ Derek F. Harris,² Molly B. Wilker,^{3*} Andrew Rasmussen,² Nimesh Khadka,² Hayden Hamby,³ Stephen Keable,⁴ Gordana Dukovic,³ John W. Peters,⁴ Lance C. Seefeldt,² Paul W. King^{1†}

Science, 2016

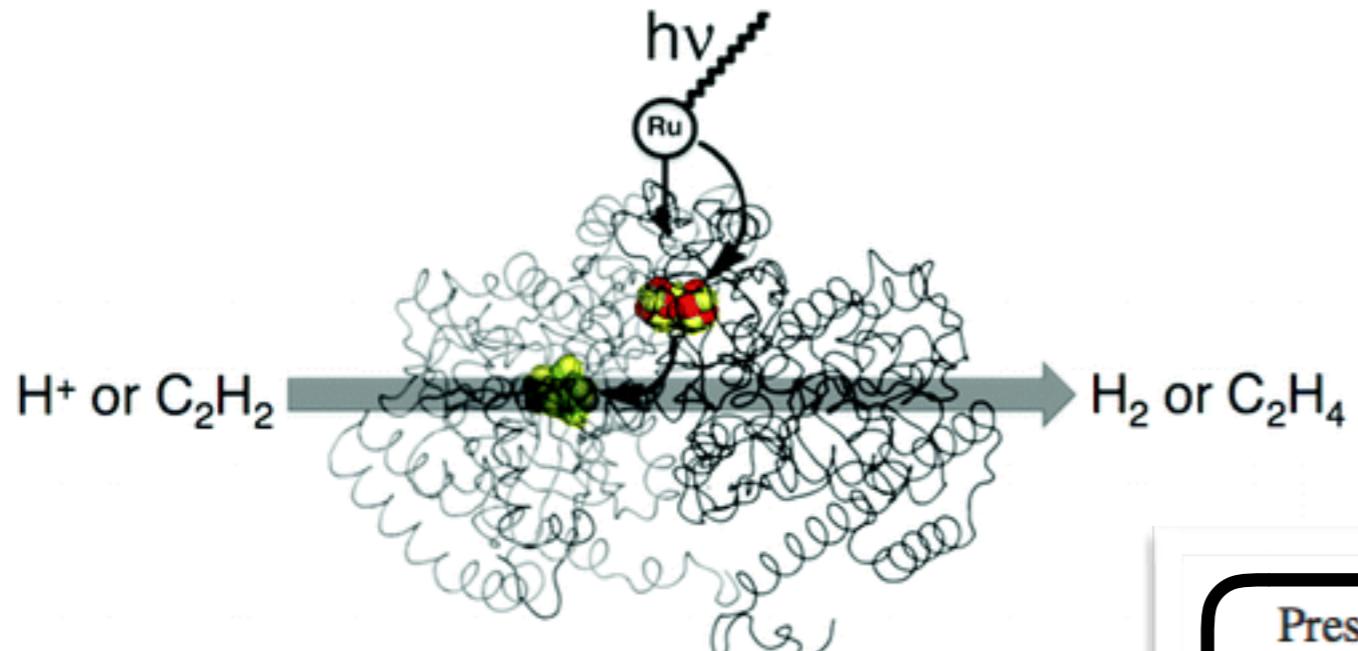


ATP- and Iron–Protein-Independent Activation of Nitrogenase Catalysis by Light

Lauren E. Roth, Joey C. Nguyen, and F. Akif Tezcan*

University of California, San Diego, Department of Chemistry and Biochemistry, La Jolla, California 92093-0356

Received August 10, 2010; E-mail: tezcan@ucsd.edu



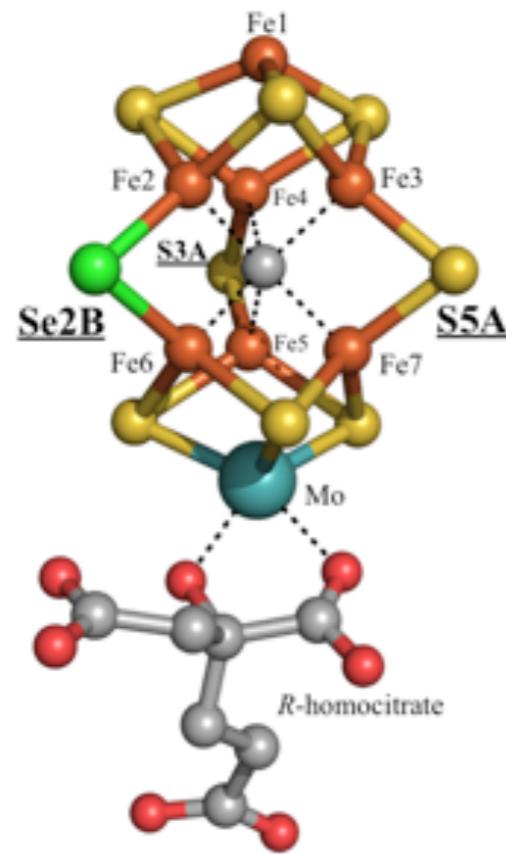
- Ruthenium tagged proteins could provide access to controlled “flashing” to generate intermediates. However....

Presently, the quantum yield (ϕ = catalytically useful electrons/photon) of our light-driven system is <1%, which likely is the reason why Ru-C158 fails to produce significant levels of ammonia from N₂. We are currently pursuing the optimization of our system for increased yields of electron injection to potentially enable this 6-e⁻ process. Ultimately, however, our experiments show that ATP hydrolysis and FeP are not absolutely essential for substrate reduction by nitrogenase as commonly assumed.

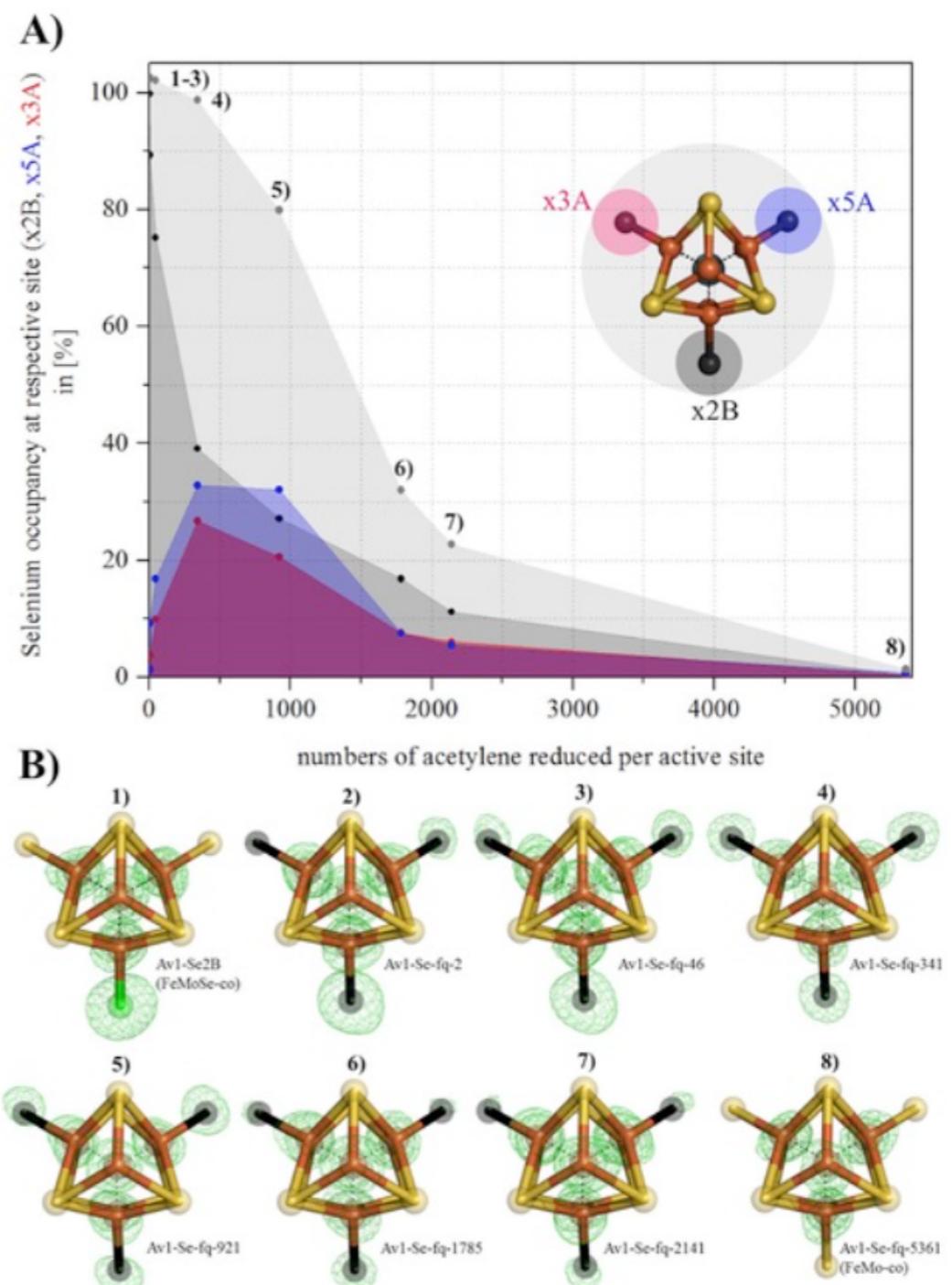
Se HERFD to Interrogate Intermediates

T. Spatzal *et al.*, *ELife*, 2015, 4:e11620

In collaboration
with Renee Arias,
Doug Rees
(Caltech)

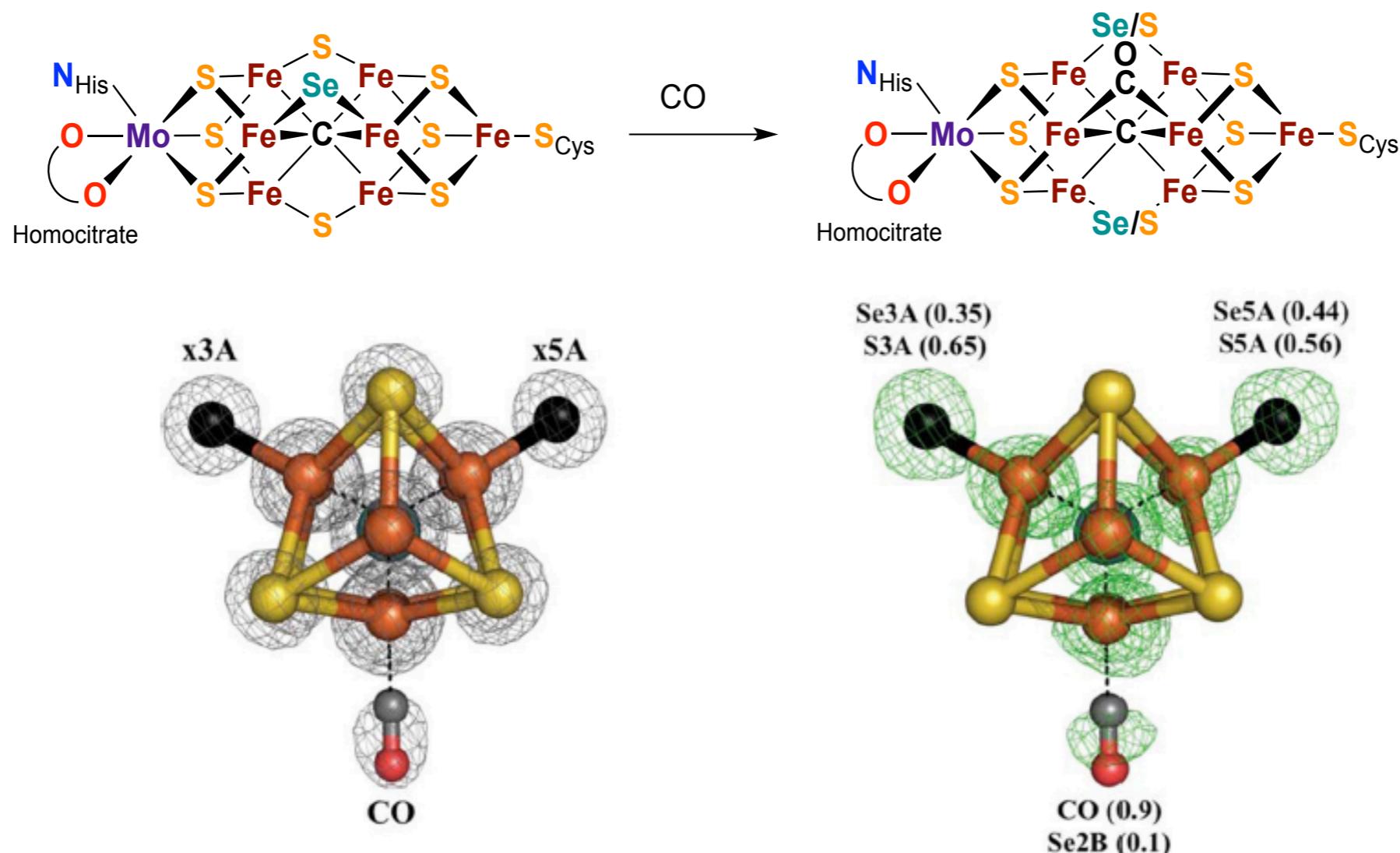


- Se can place the S₂B belt sulfur
- Se migrates to S₃A and S₅A at later stages of catalysis
- FeMoco may be more dynamic than previously thought
- The central carbon, may provide structural stability



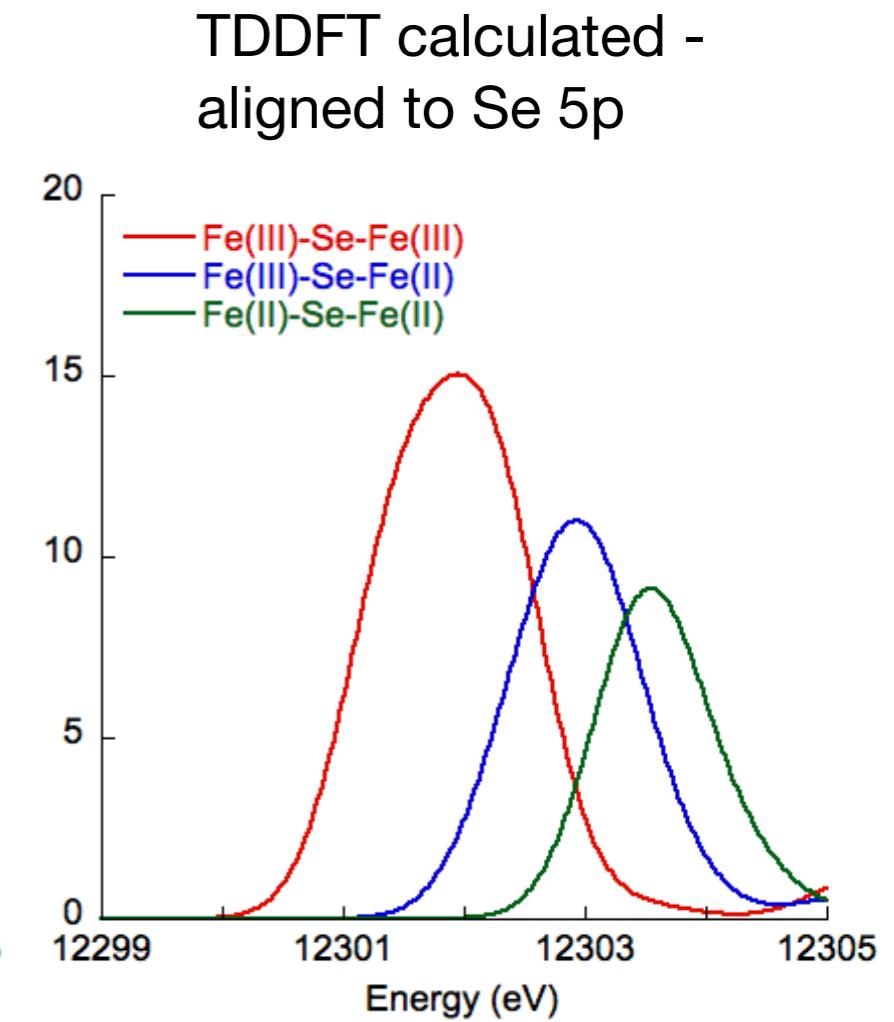
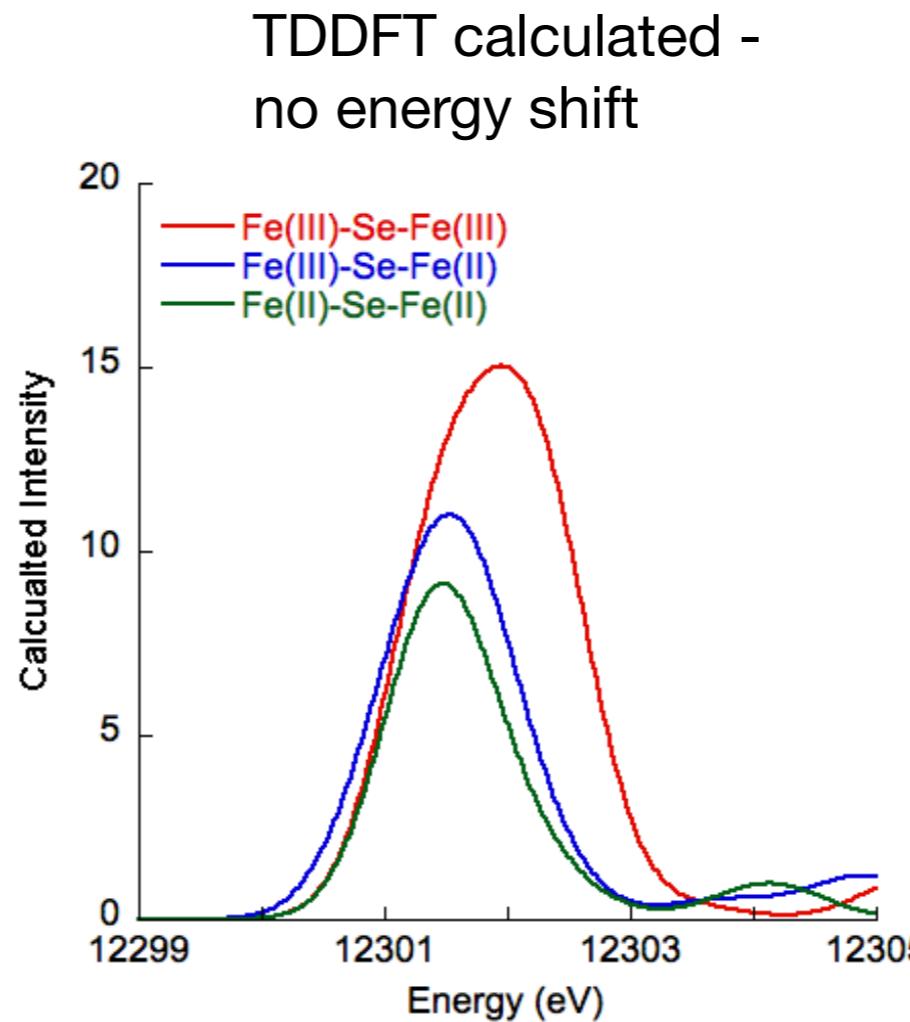
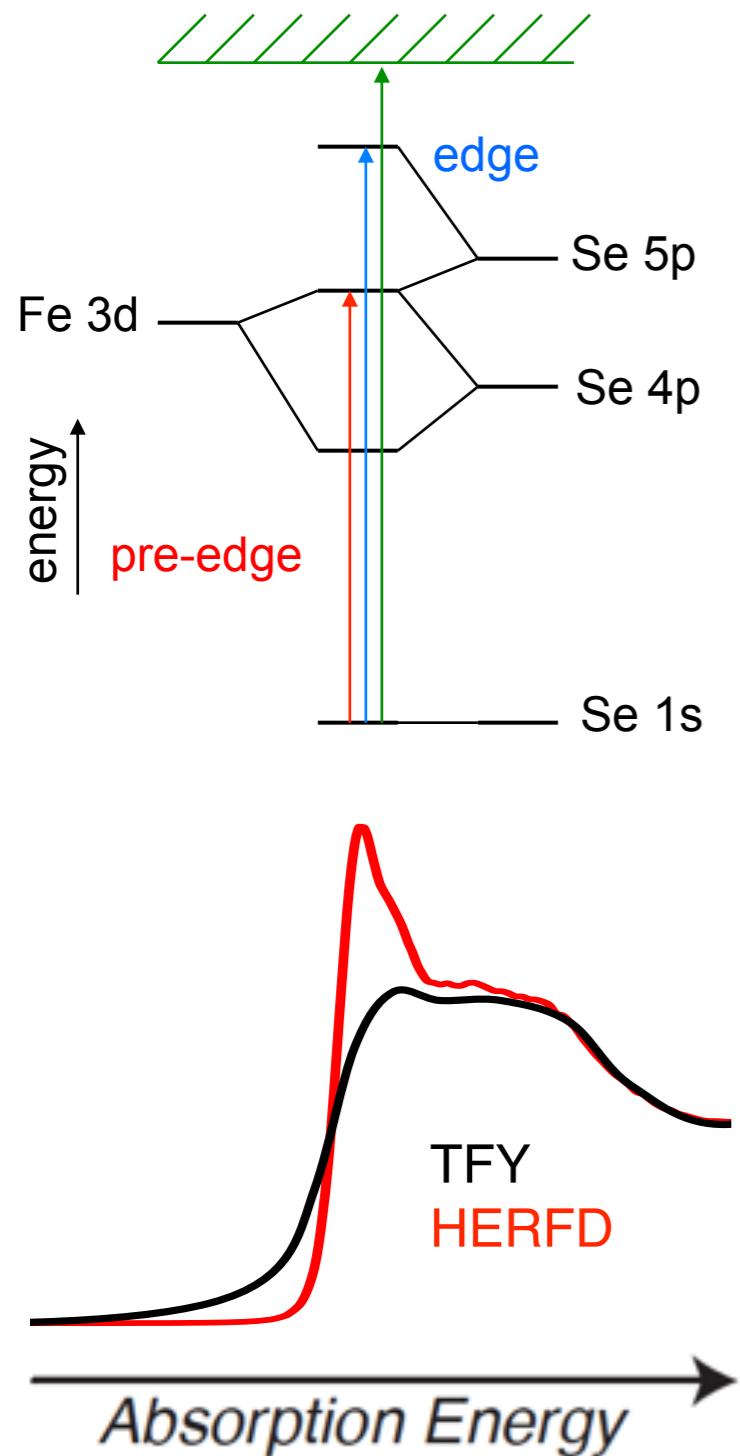
Displacement of Se by CO

T. Spatzal *et al.*, *ELife*, 2015, 4:e11620



- In the presence of CO, CO will displace the S2B Se and Se migrates to S3A and S5A positions
- Electronic structure of CO bound form is unknown, though presumably corresponds to a more reduced FeMoco species. ($S=1/2$ E_2 state, however only ~30% of the EPR signal)
- Should allow for X-ray spectroscopic characterization of a reduced substrate-bound intermediate of FeMoco.**

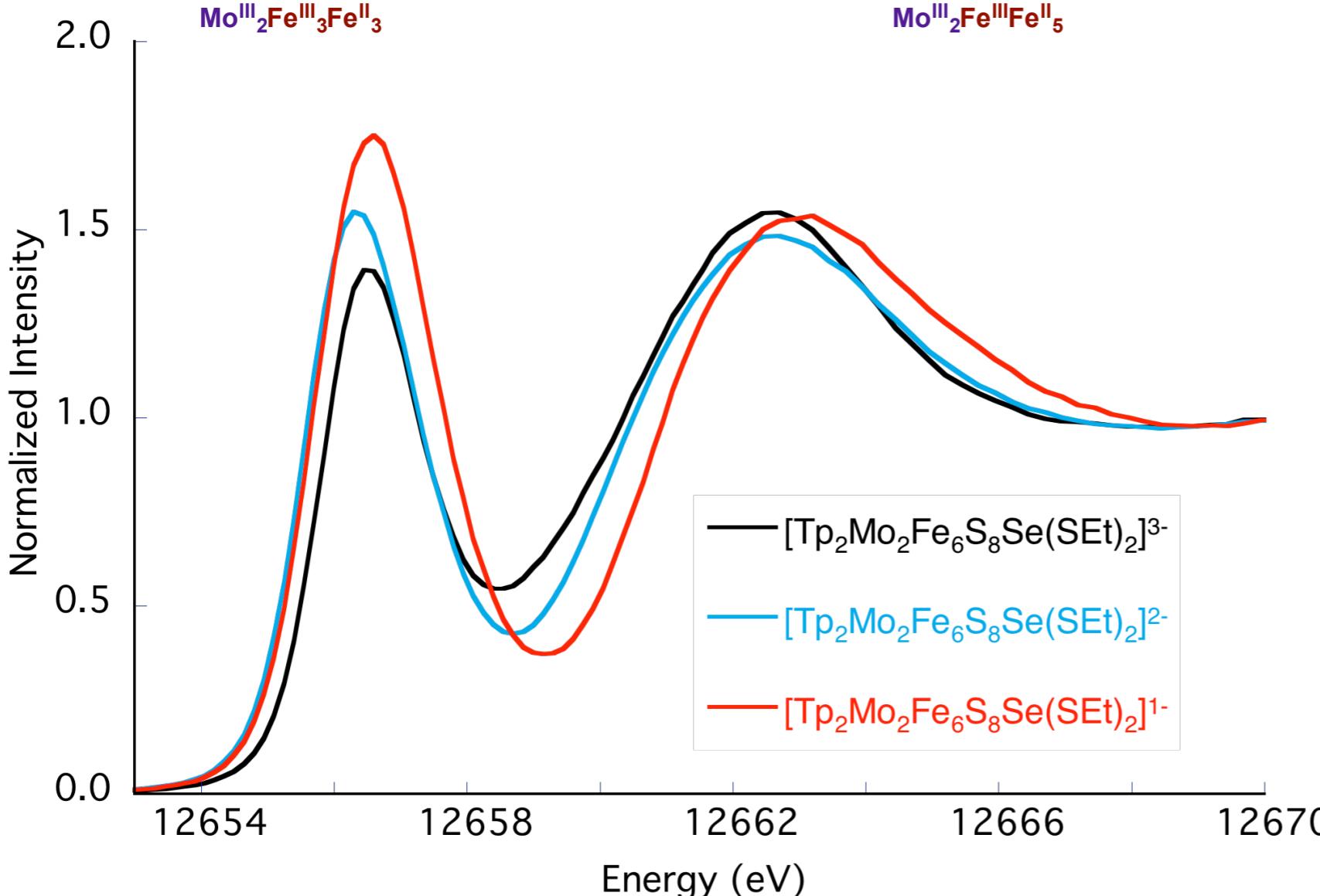
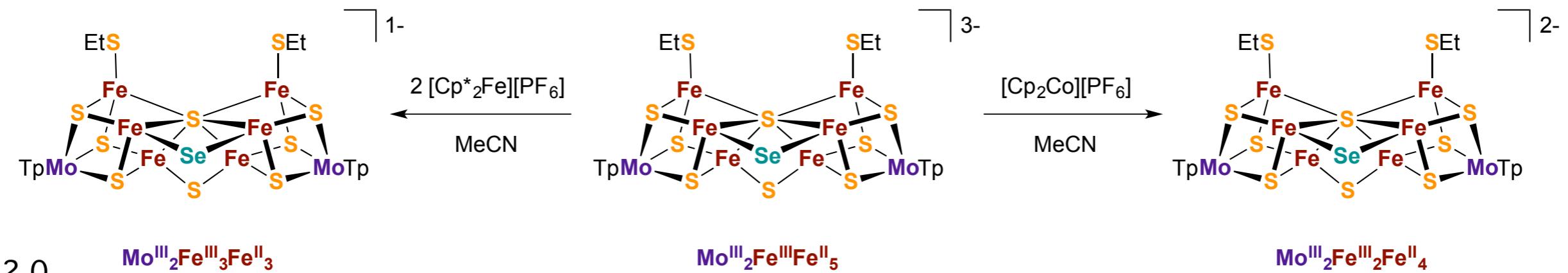
Se HERFD XAS



- Se ligand K-edge XAS applications limited due to large core hole lifetime broadening at Se.
- Increased resolution of a HERFD XAS measurement allow for well resolved pre-edge to be observed.
- Se pre-edge provides a probe of Fe-Se covalency
- Utilize Se as a spectator probe of Fe oxidation state.

Se HERFD of Se-bridged N2ase Models

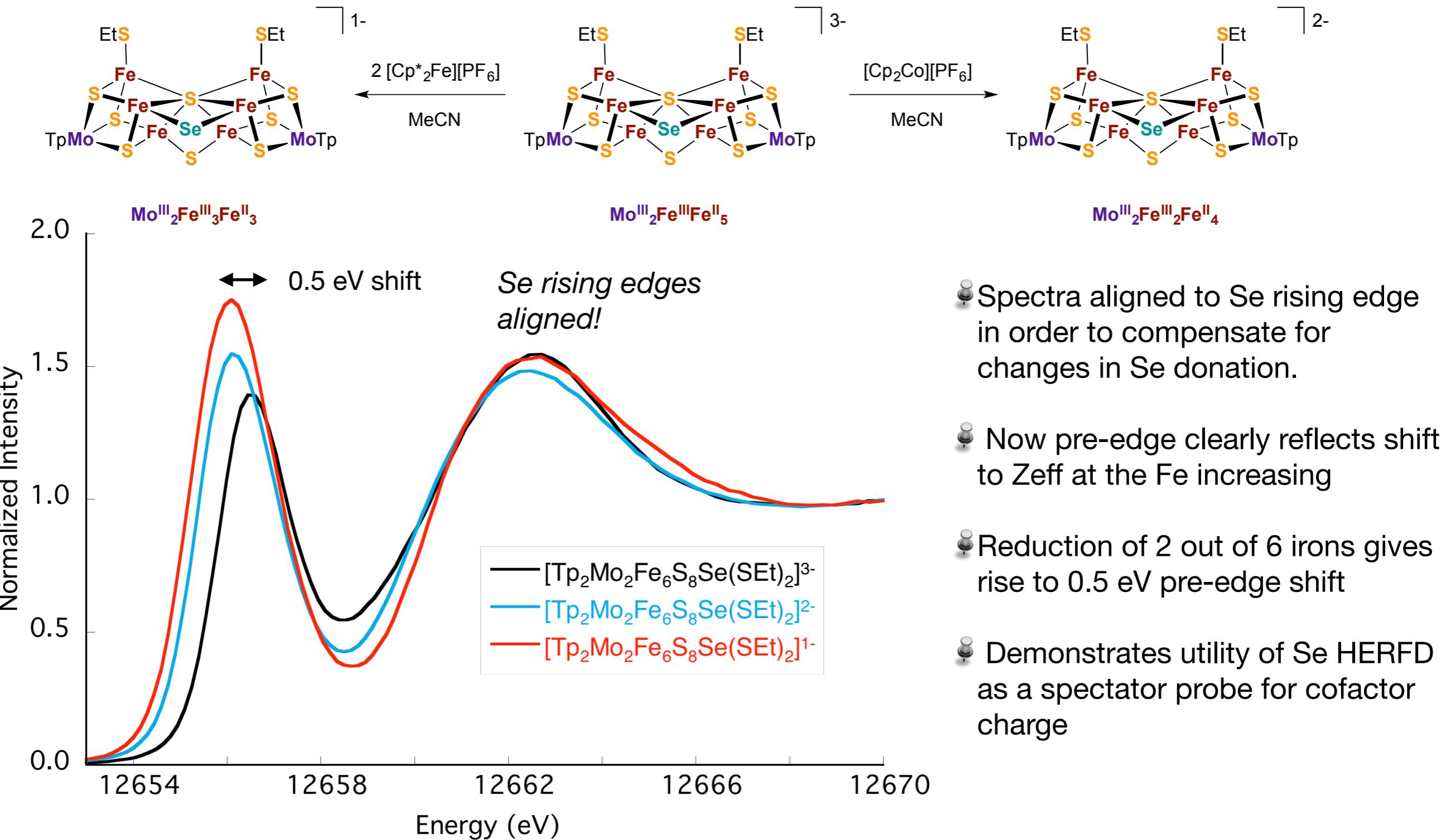
Berlinguette, C.P; Holm, R.H. *J. Am. Chem. Soc.*, 2006, 128, 11993–12000.



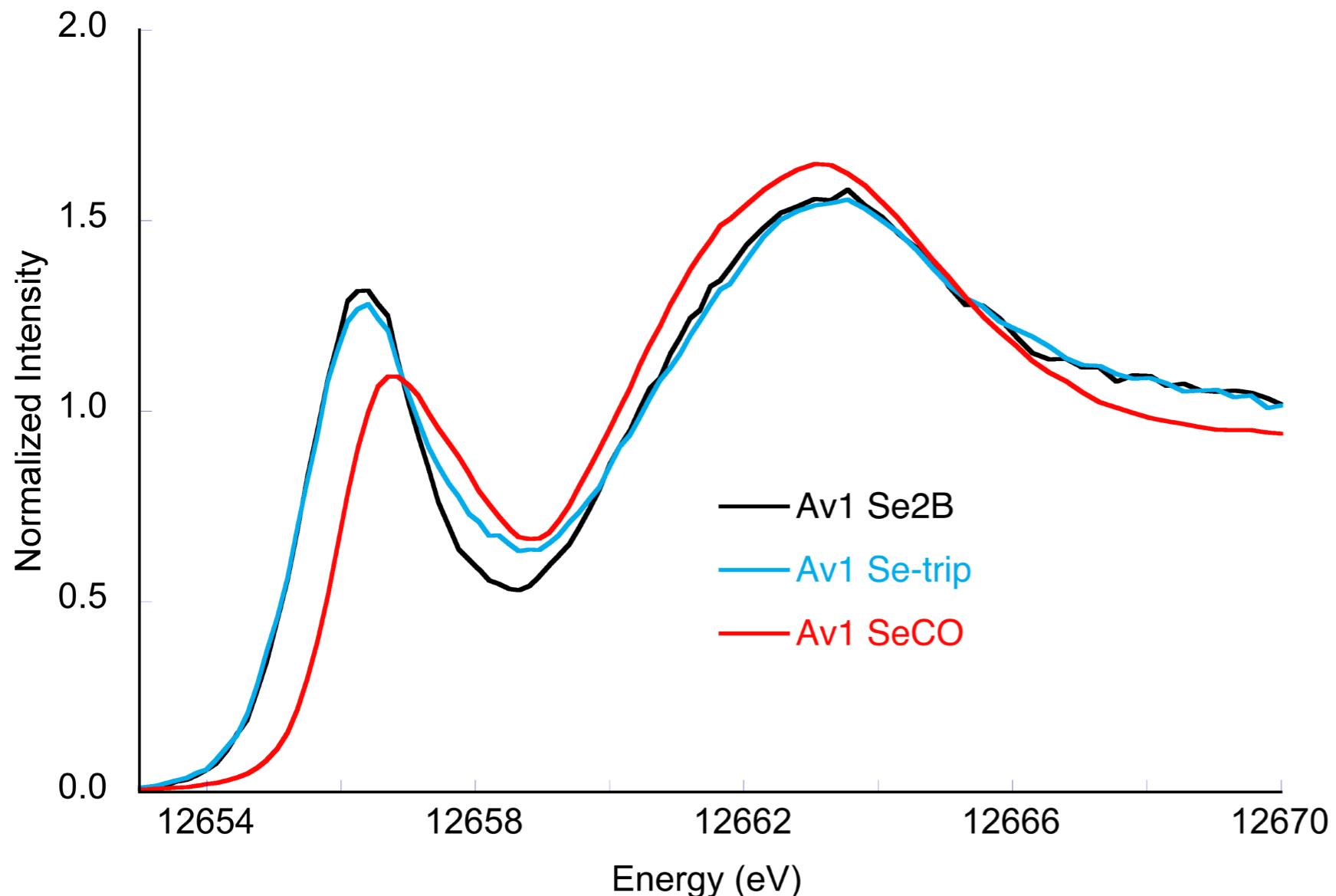
Justin Henthorn

Se HERFD of Se-bridged N2ase Models

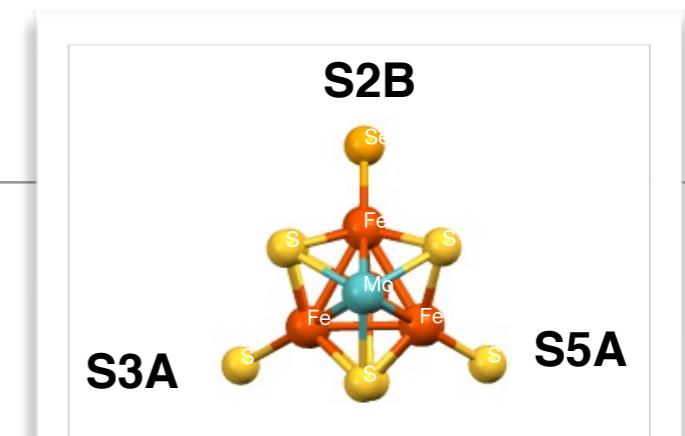
Berlinguette, C.P; Holm, R.H. *J. Am. Chem. Soc.*, 2006, 128, 11993–12000.



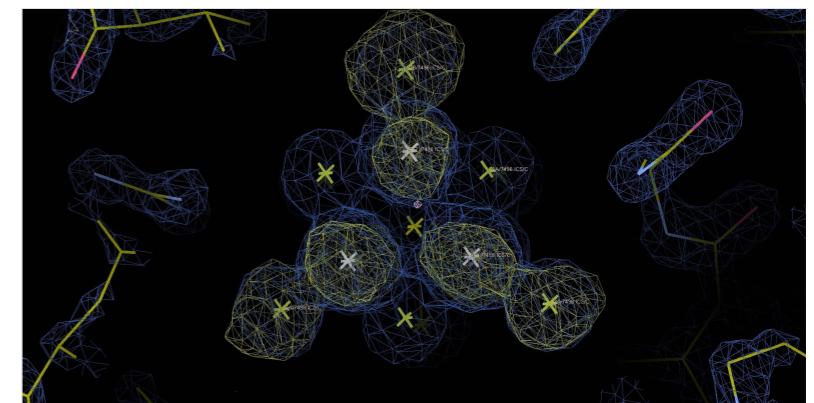
Se HERFD of MoFe



- Changes upon CO binding clearly more pronounced
- Reflects iron oxidation states interacting with S3A and S5A
- Indicates overall more reduced cluster

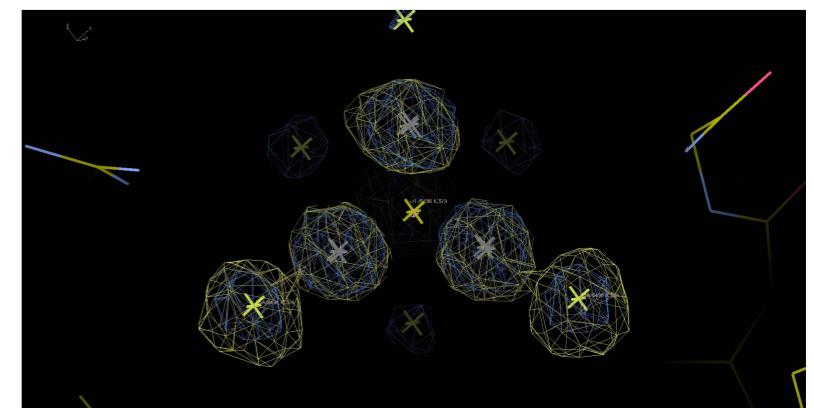


Av1 Se2B



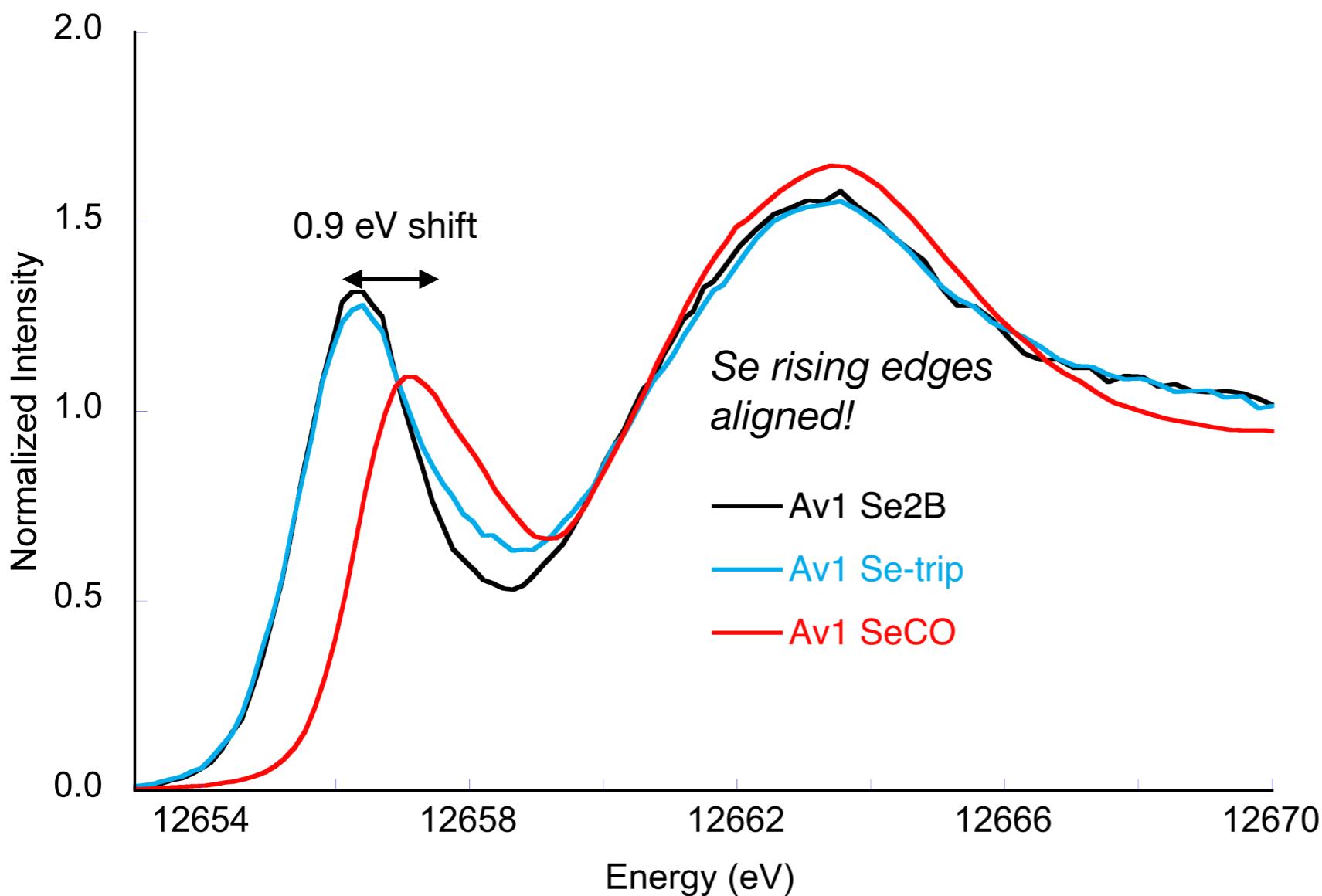
S2B: 87.7% S3A: 49.2% S5A: 48.5%

Av1 CO

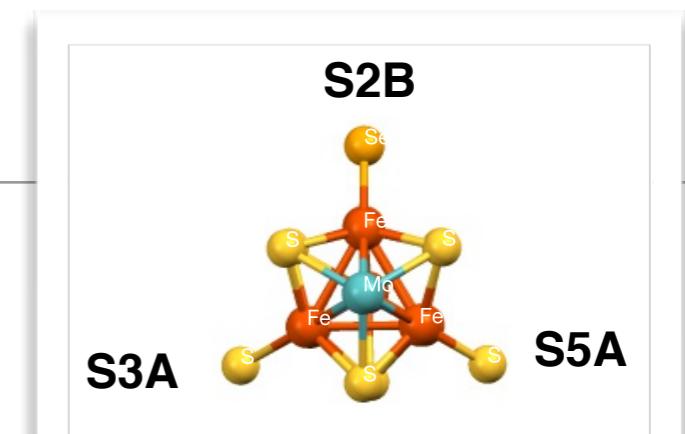


S2B: 12.5% S3A: 49.4% S5A: 48.5%

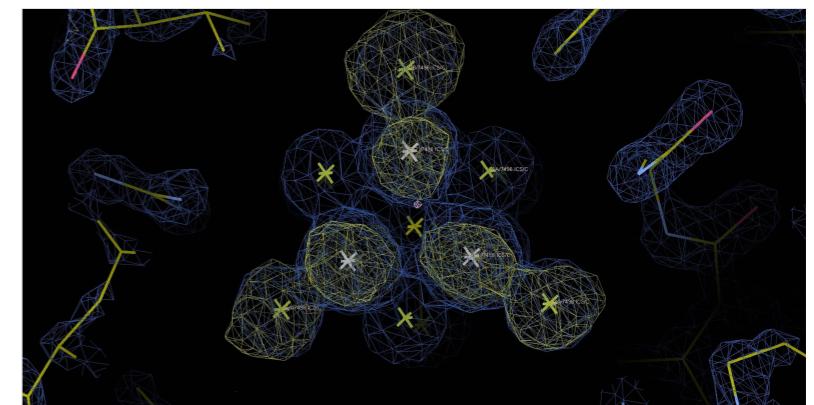
Se HERFD of MoFe



- Alignment of rising edge shows average d-manifold position shifts by almost 1 eV in CO bound protein
- Typically assumed that CO binds at two-electron reduced E2 state, with S=1/2 EPR (but only 30% by EPR spin quantitation)
- Is CO binding to other integer spin reduced states?
- Correlations to theory in progress...

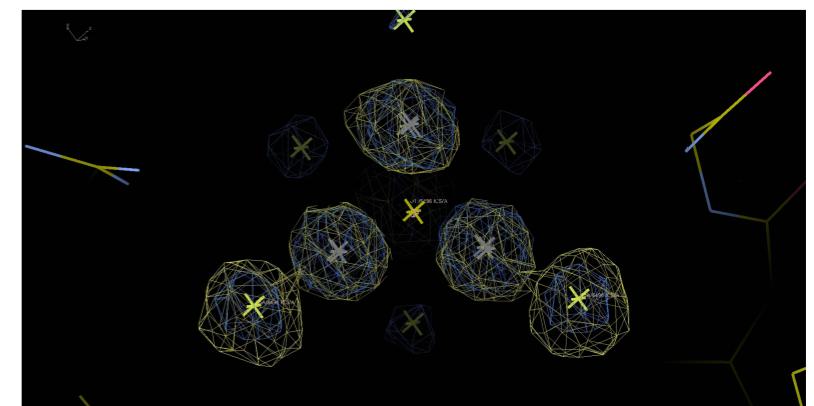


Av1 Se2B



S2B: 87.7% S3A: 49.2% S5A: 48.5%

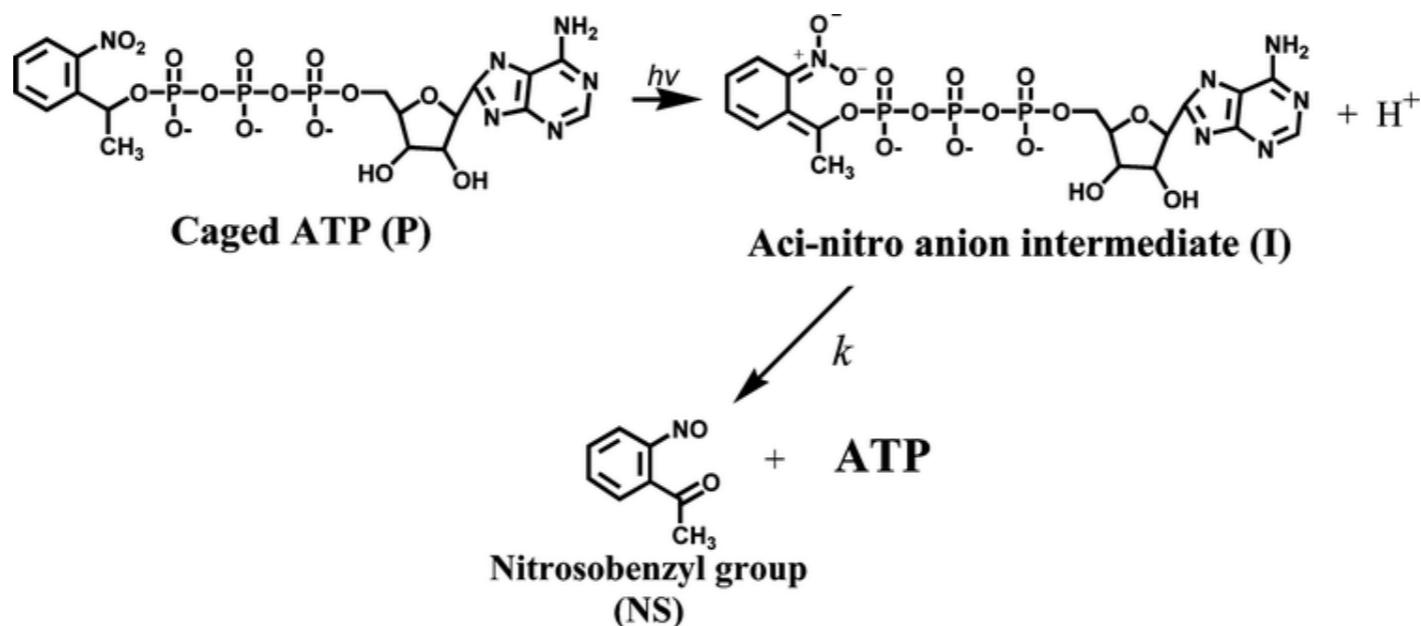
Av1 CO



S2B: 12.5% S3A: 49.4% S5A: 48.5%

Time-resolved studies of Nitrogenase - Outlook

- ⌚ Further developments are needed to enable controlled access to reaction intermediates in nitrogenase
- ⌚ The presence of the iron protein, P-cluster, and FeMoco makes evaluation of iron spectra more challenging. Selective tools are needed!
- ⌚ Selenated nitrogenase provides access to a more selective probe. Se HERFD in progress...
- ⌚ Additional photo triggers should be explored, such as caged ATP...



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Mustafa al Samarai

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Vijay Chilkuri

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Stefan Hugenbruch

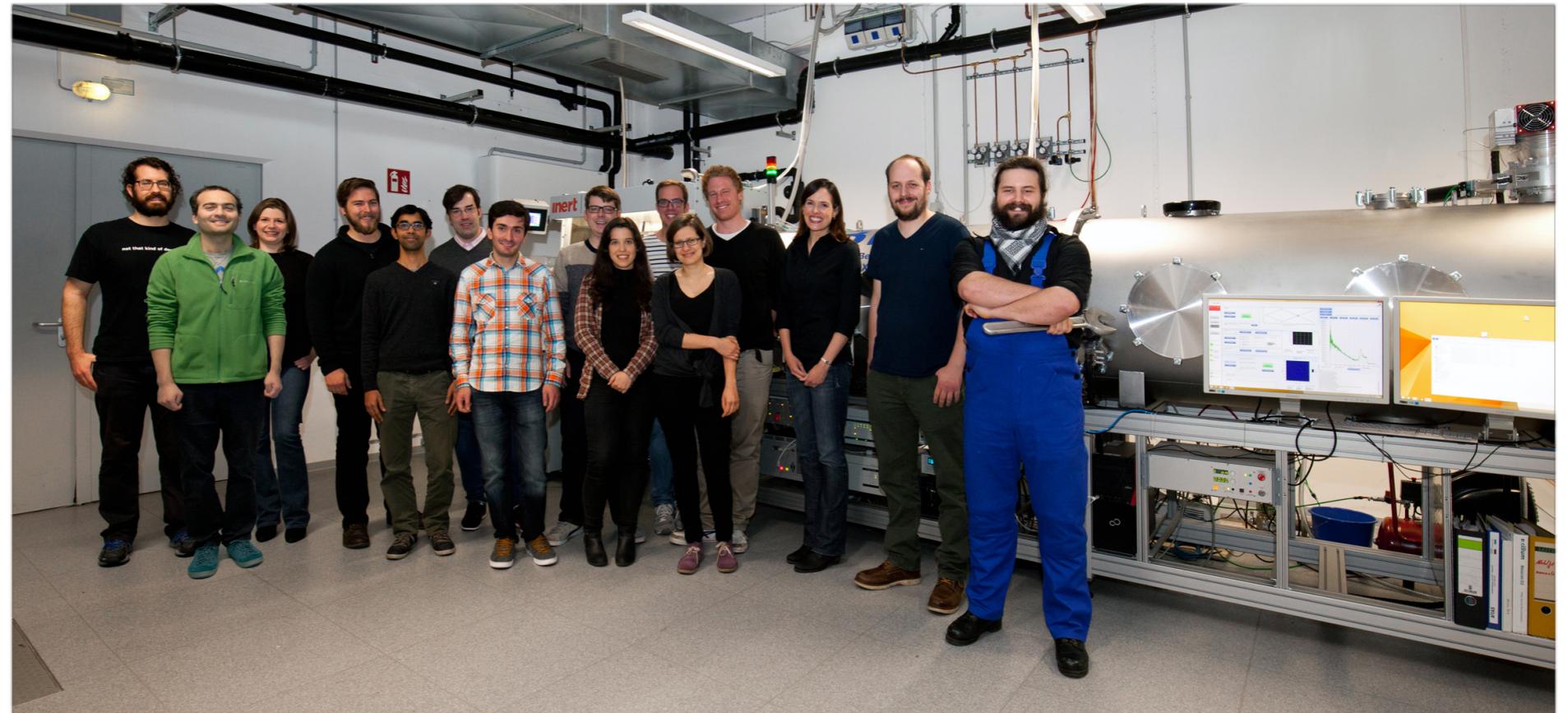
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Thomas Weyhermüller

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Doug Rees (Caltech)

Oliver Einsle (Freiburg)

Uwe Bergmann (SLAC)

Pieter Glatzel (ESRF)

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\$ DFG

\$ BMBF

\$ AvH (GC and JH)

\$ IMPRS (RGC and CV)

Beam time: ESRF, CHESS, SOLEIL, SSRL

Thank you for your attention!

