



HAXPES 2011

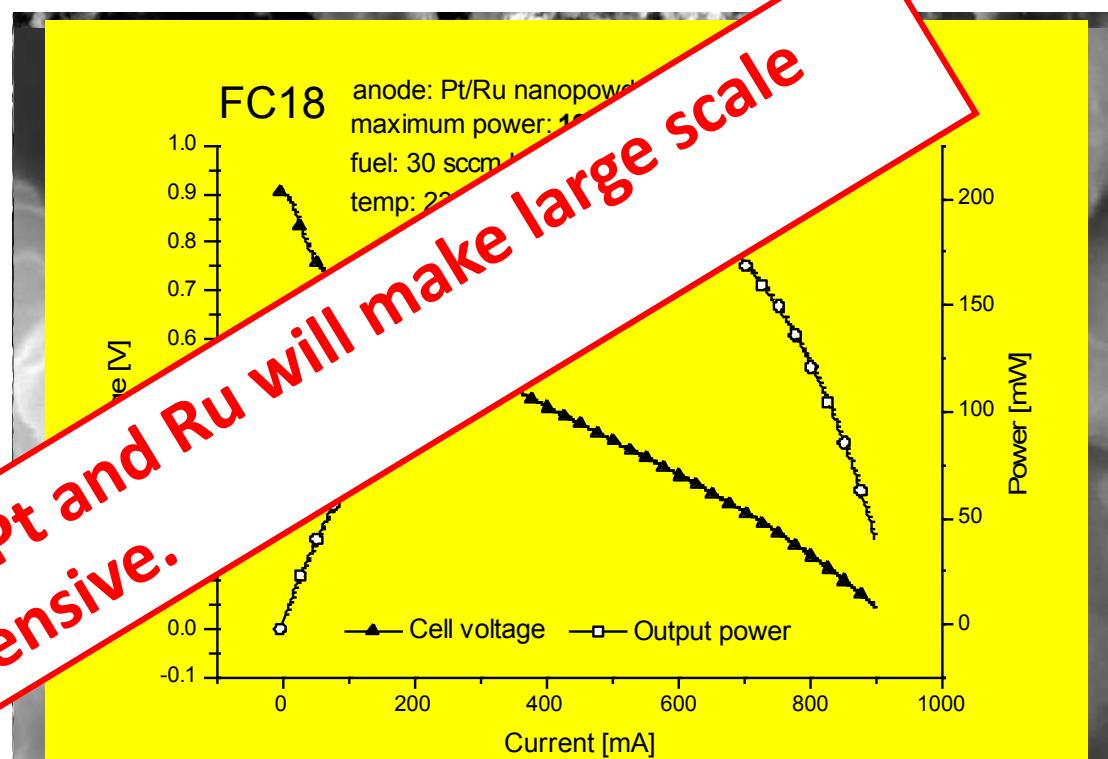
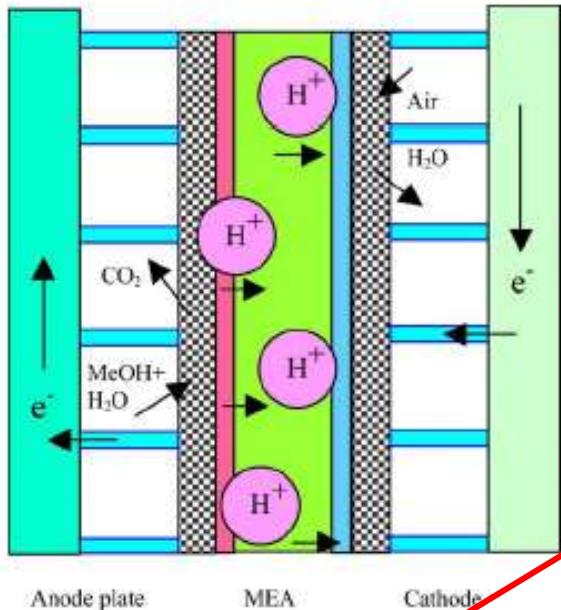


HAXPES investigation of nanostructured proton exchange membrane fuel cell catalysts

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Kobayashi, B. Detlefs**

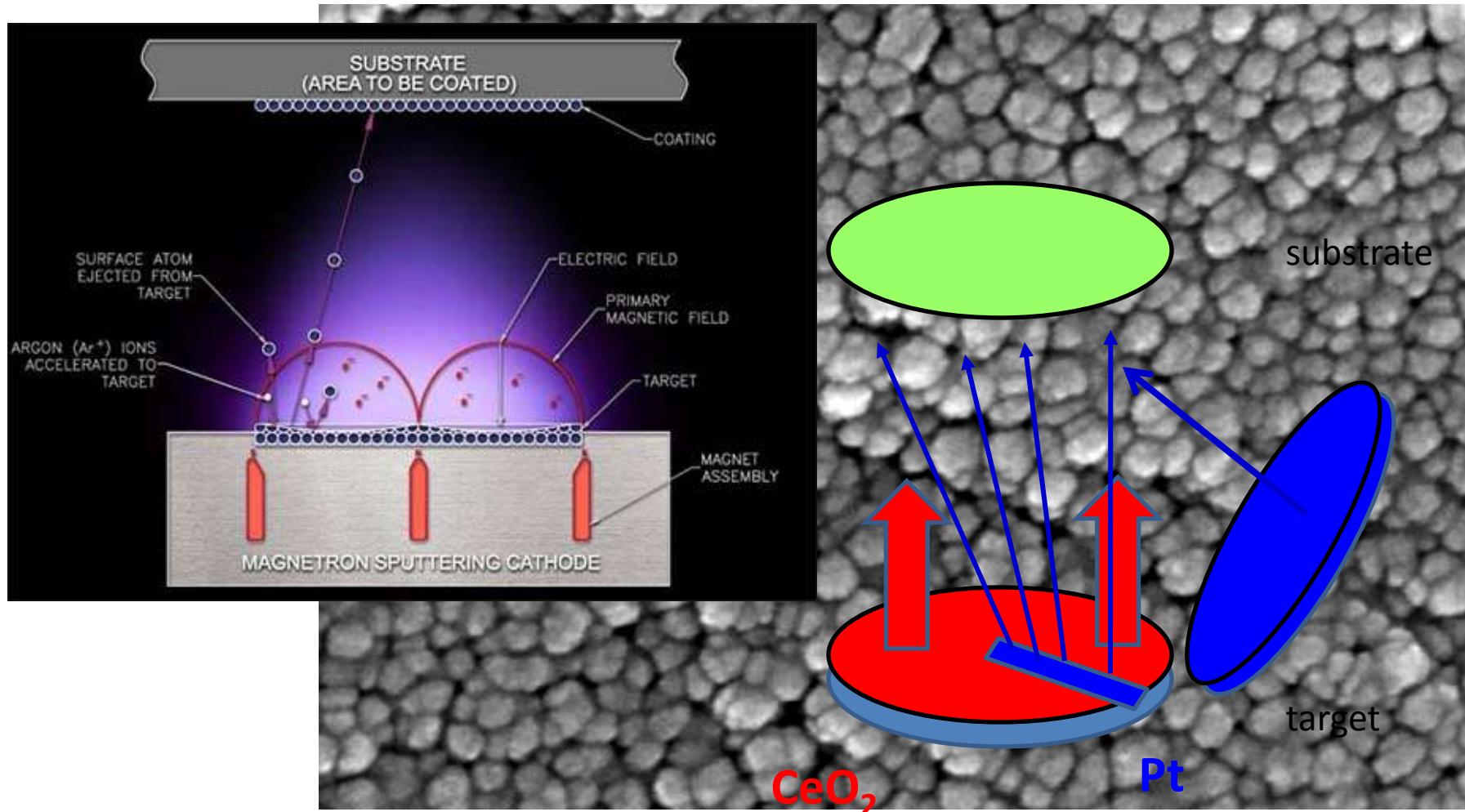
Charles University in Prague, Czech Republic
BL15XU, Spring-8, Japan
ID 32, ESRF, France

Polymer membrane fuel cell: Pt-Ru anode



Specific Power
 $SP = 0.1 \text{ W/mg(Pt)}$
Power Density
 $PD = 200 \text{ mW/cm}^2$

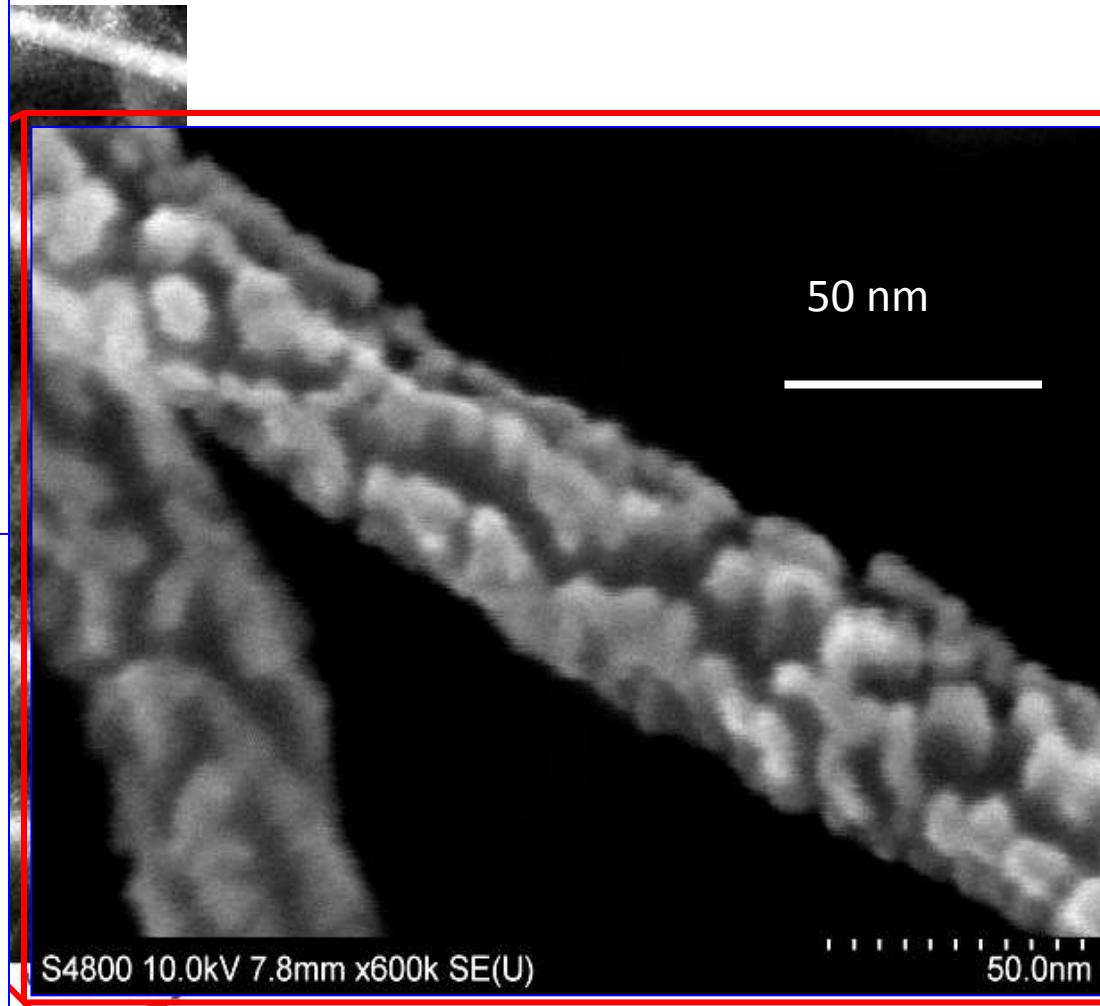
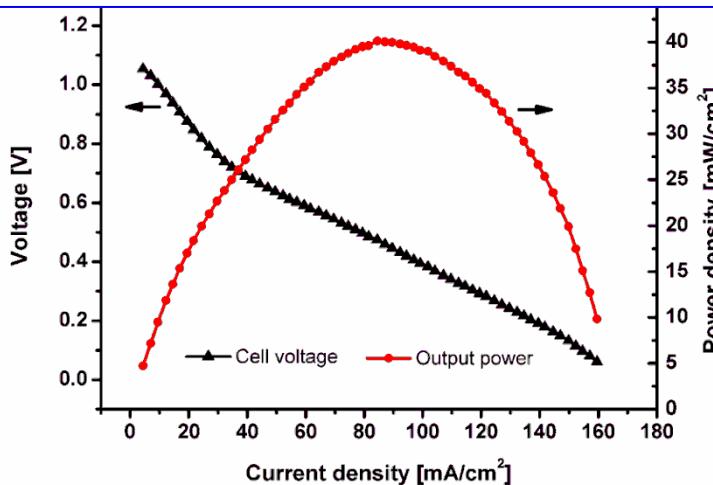
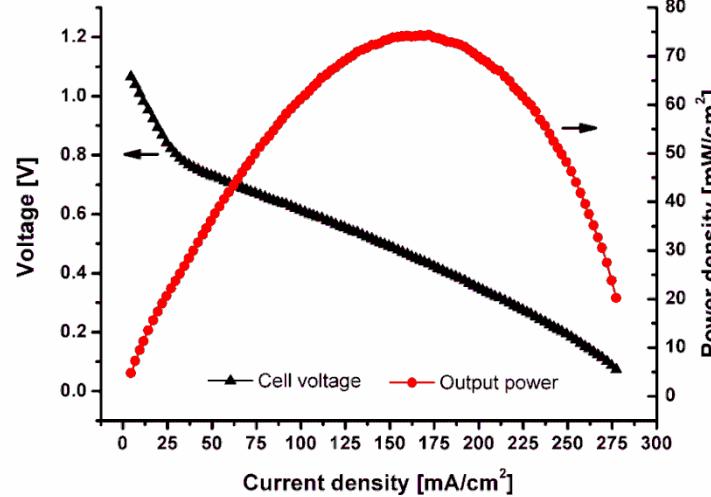
Preparation of Pt-CeO₂ catalyst by rf co-sputtering



Pt-CeO_x interaction: New catalytic properties

G. Vayssilov, ...V. M. et al., Support nanostructure boosts oxygen transfer to catalytically active platinum nanoparticles, *Nature Materials*, 10 (2011) 310

Pt-ceria/CNTs on GDL



Date(Entry): 07/03/09 Ivan

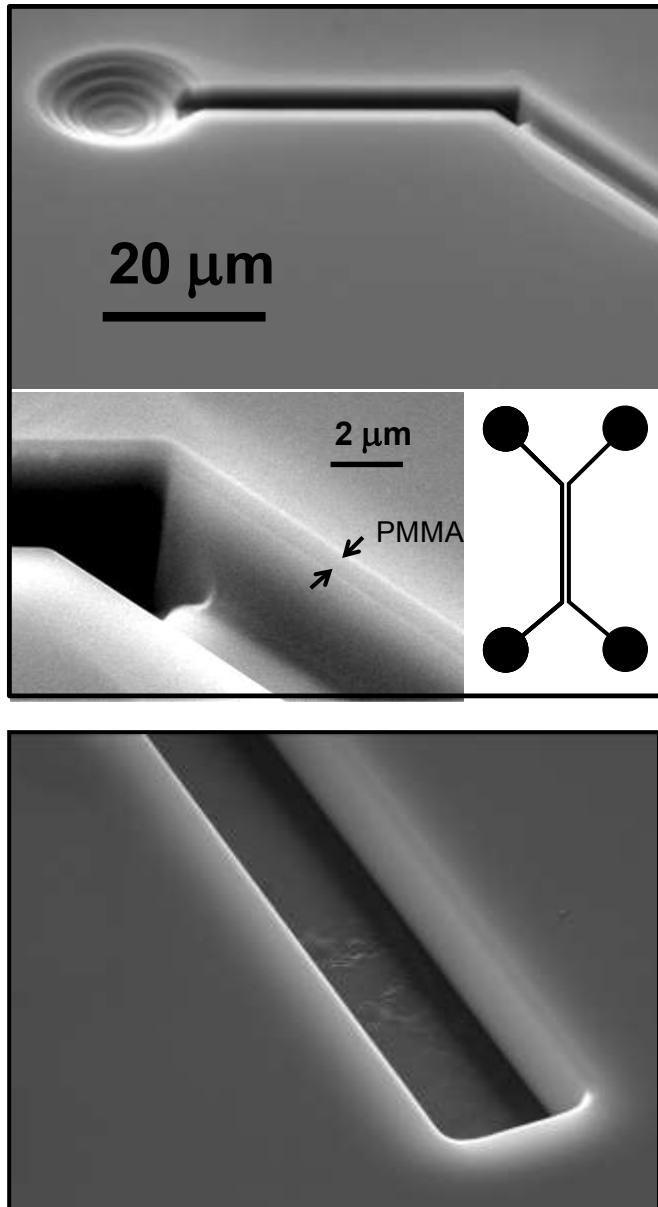
Digital Micro

PD = 40 – 80 mW

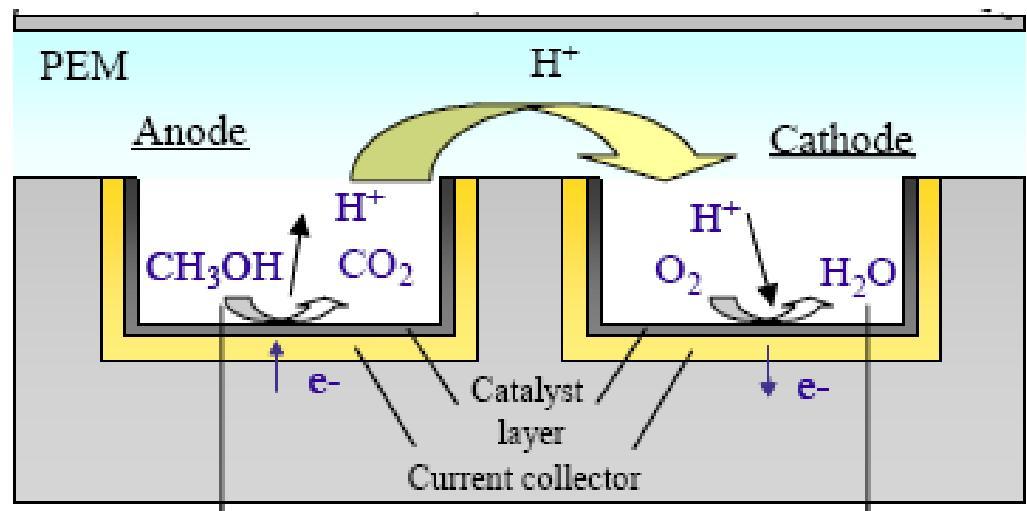
SP = 40 – 80 W/mgPt

R. Fiala,.., V.M. J Nanosci. Nanotechnol. 11 (2011)
5065
V.M. et al. Langmuir 26 (2010) 12824

Planar μ -PEMFCs, powering of integrated circuits

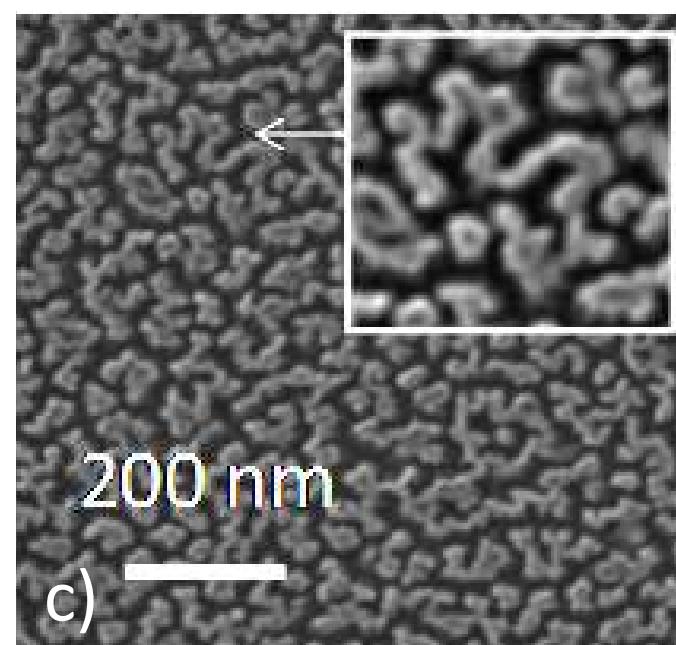
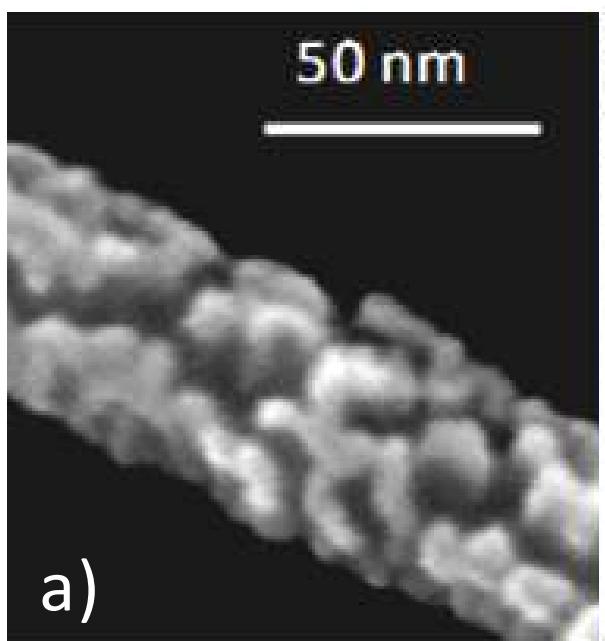
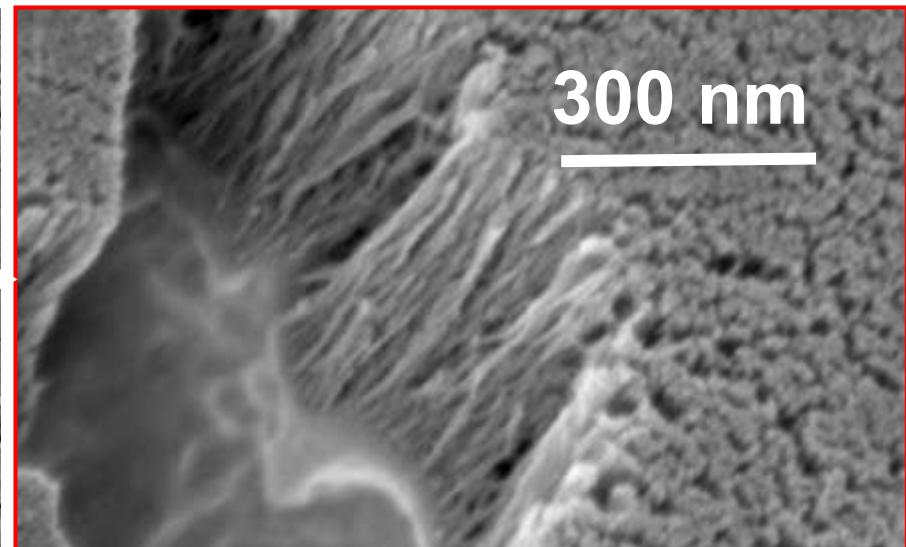
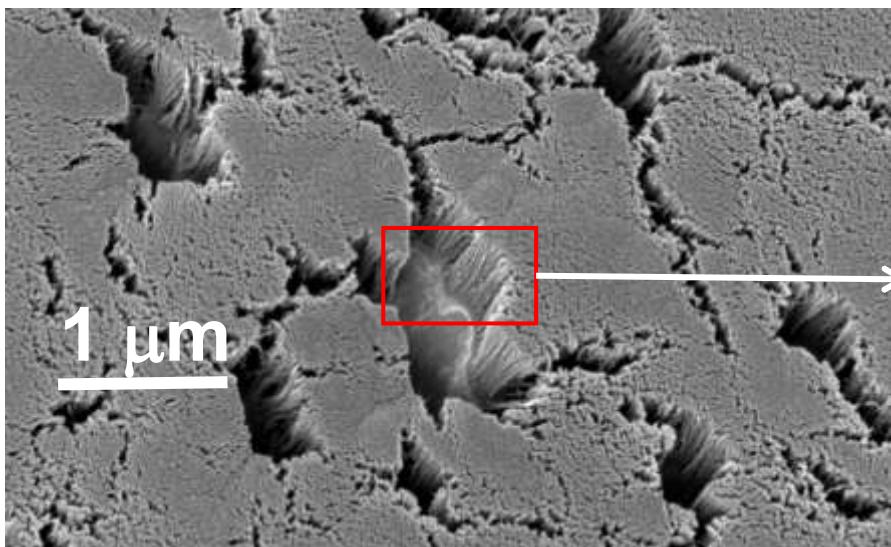


Focus Ion Beam (FIB) milling, Ion Beam Lithography, Catalyst deposition by PVD

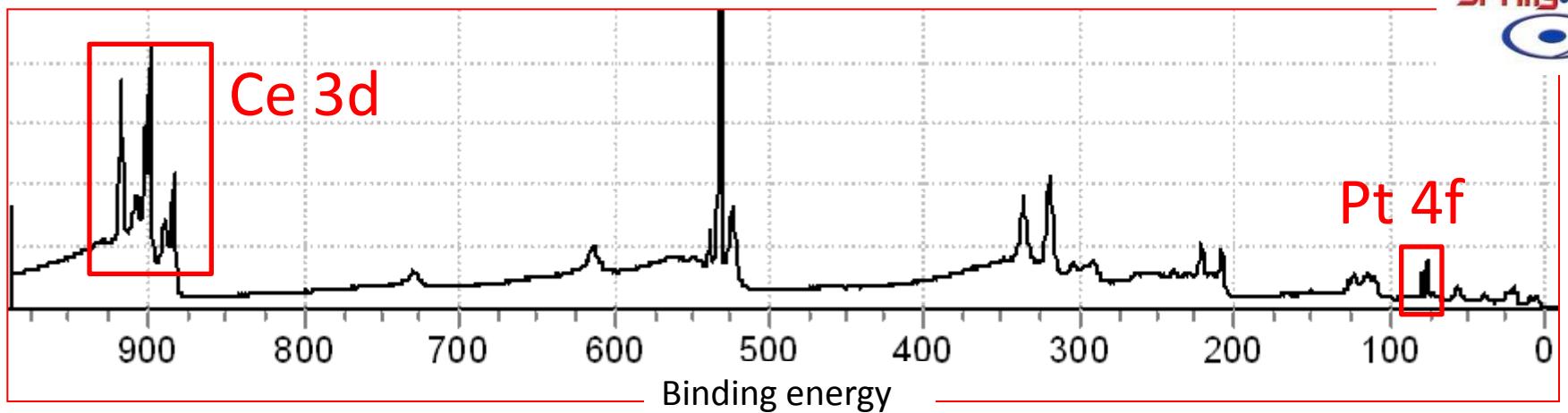


Methanol,
hydrogen

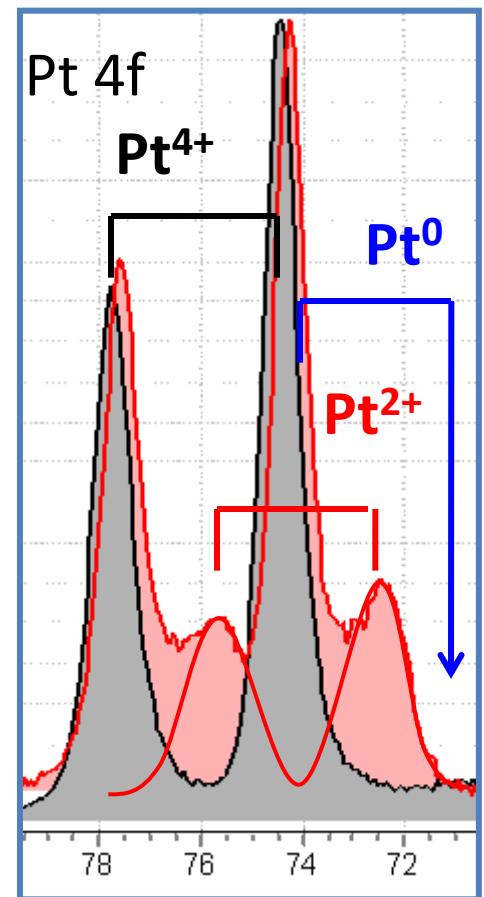
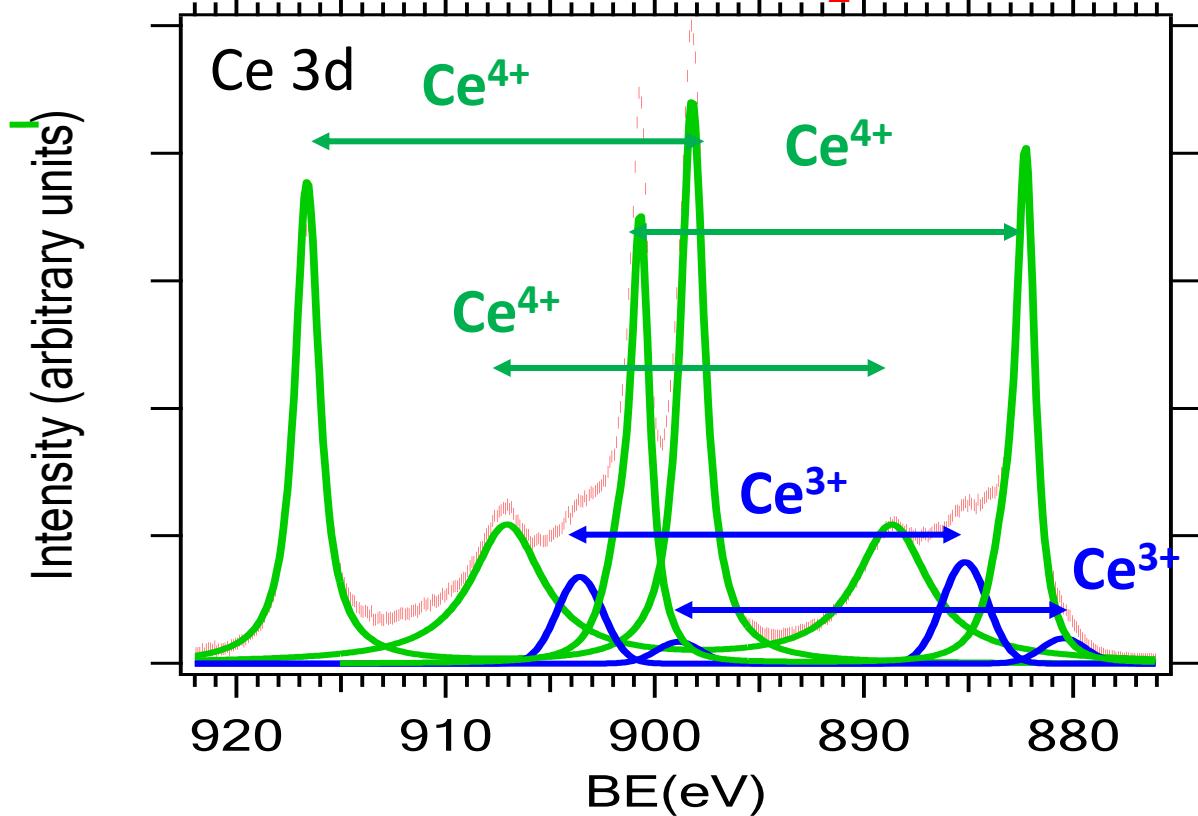
Air



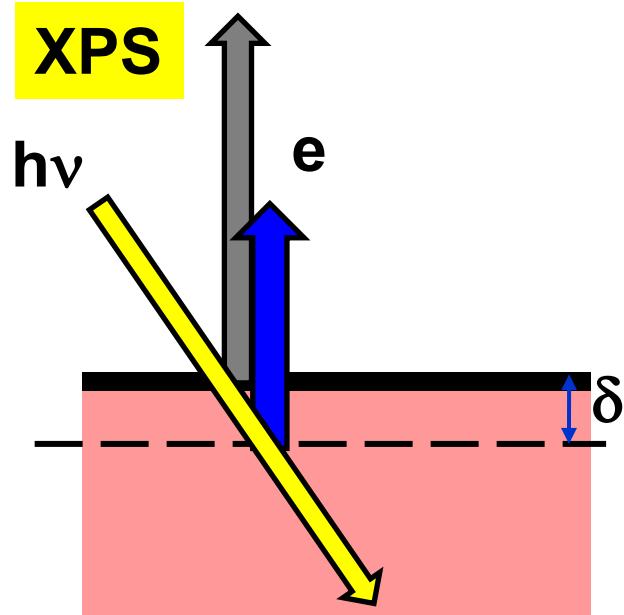
Carbon foil substrate



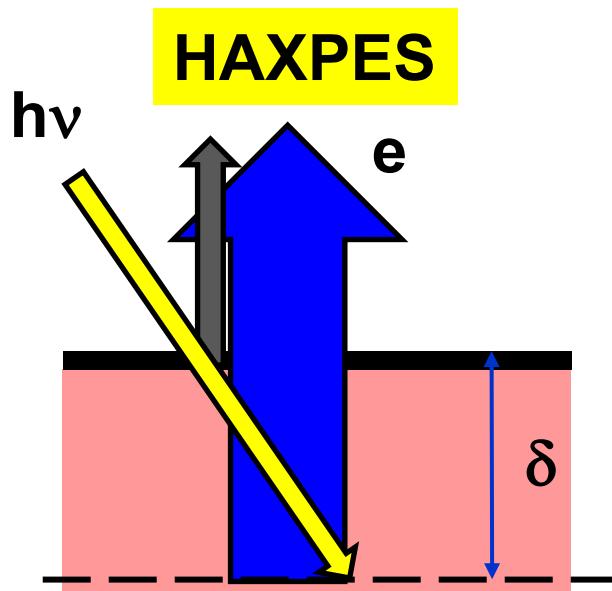
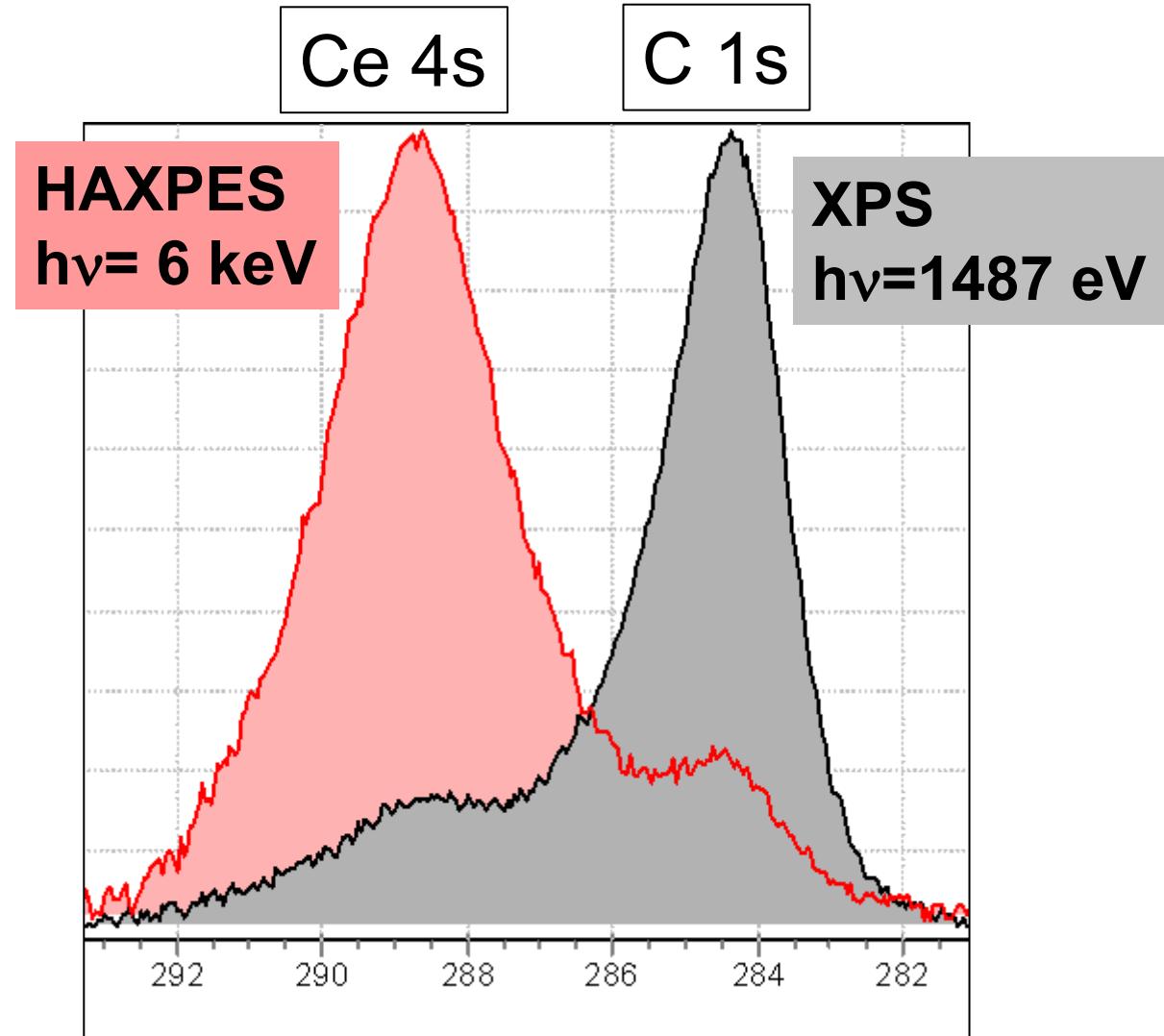
PES spectra of the Pt-CeO₂ catalyst



Relative Surface Sensitivity

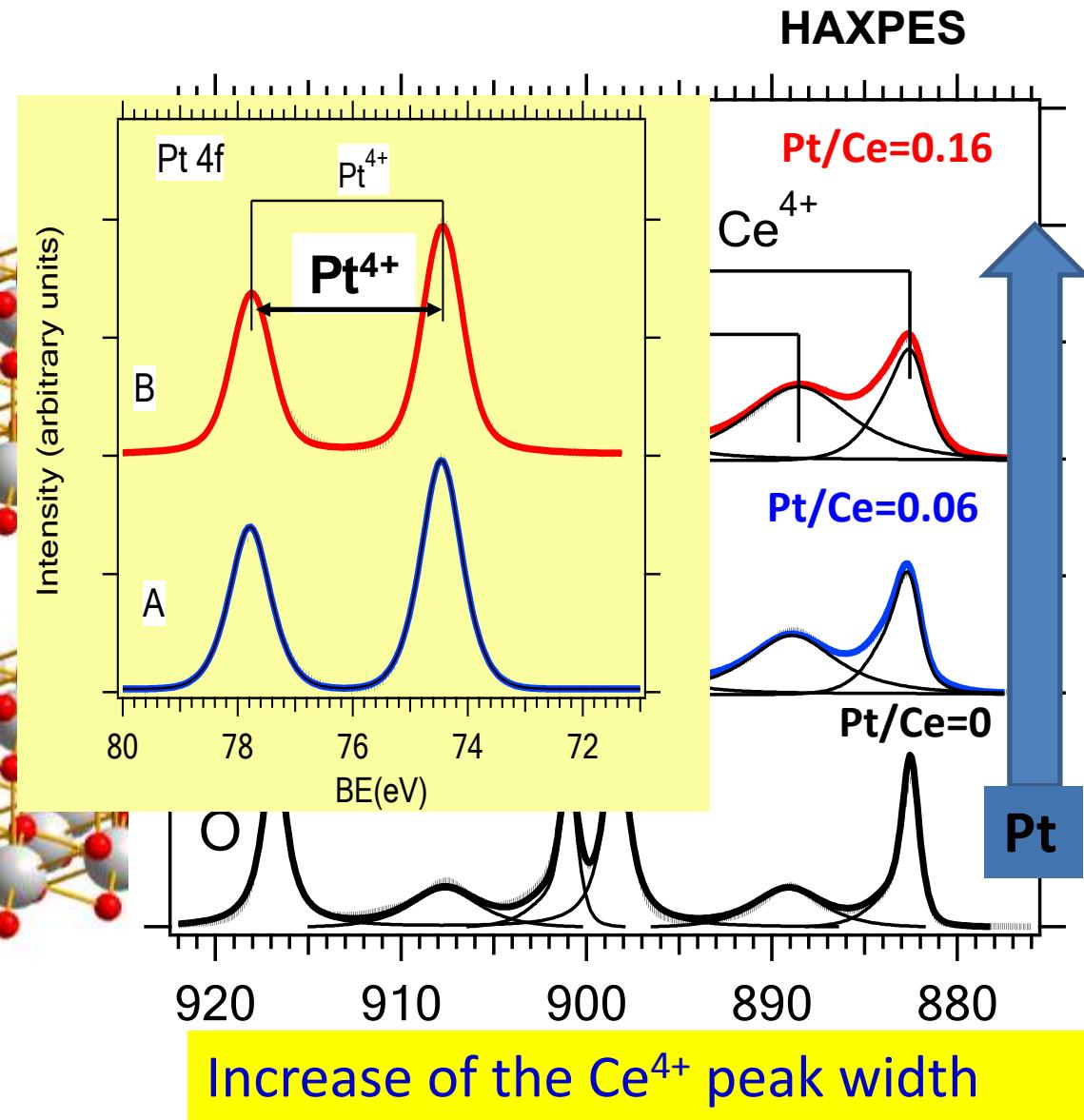
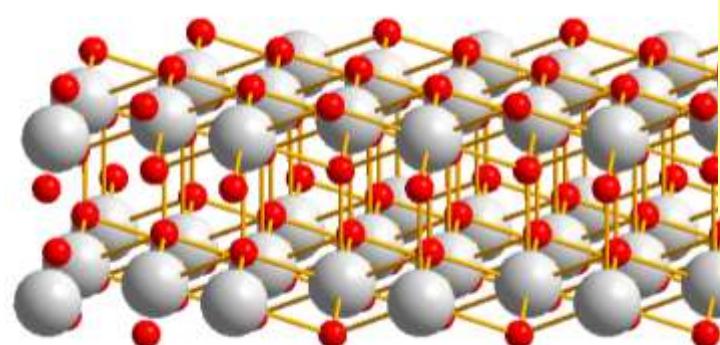
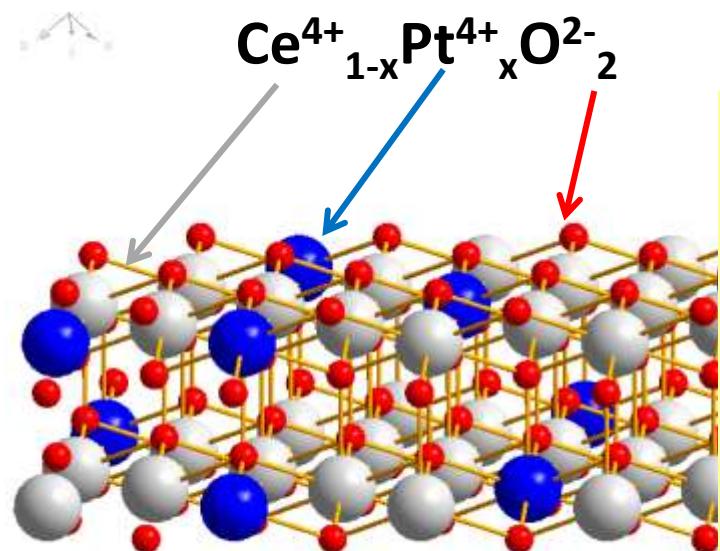


Air exposed Cerium oxide – C contamination
(normalized to peak maximum)



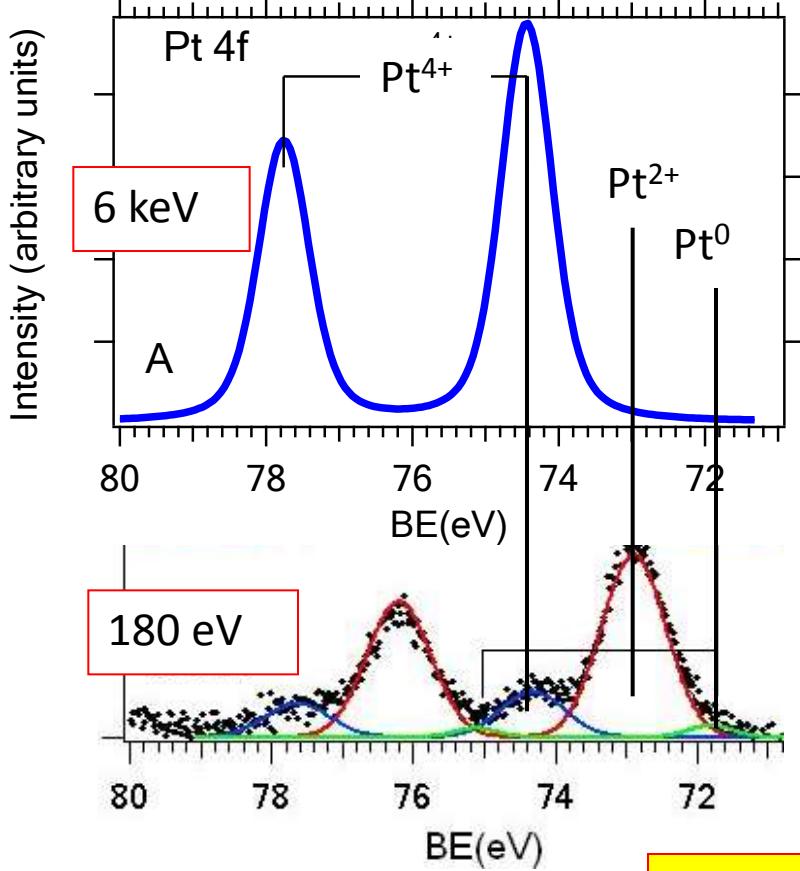
Co-sputtering – growth of mixed oxides (solid solution)

Substrate: SiO_2 , normal deposition

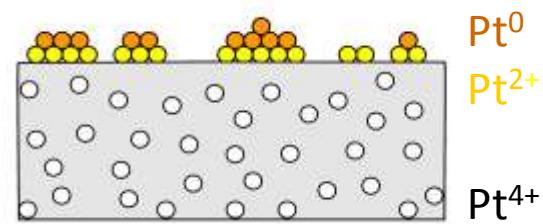


PES study - rf sputtered thin films: s-x→H-x

Pt-CeO₂/SiO₂/Si

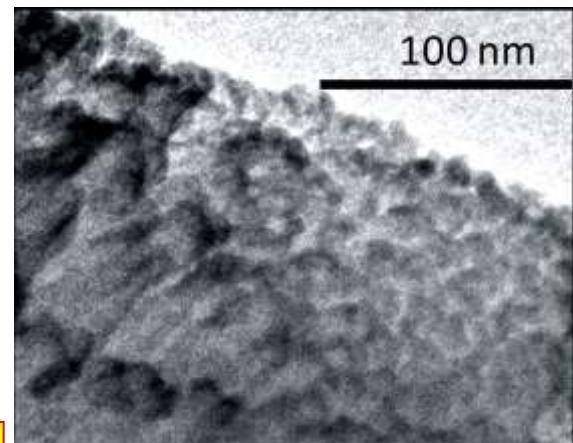


HAXPES
10 nm



SXPES

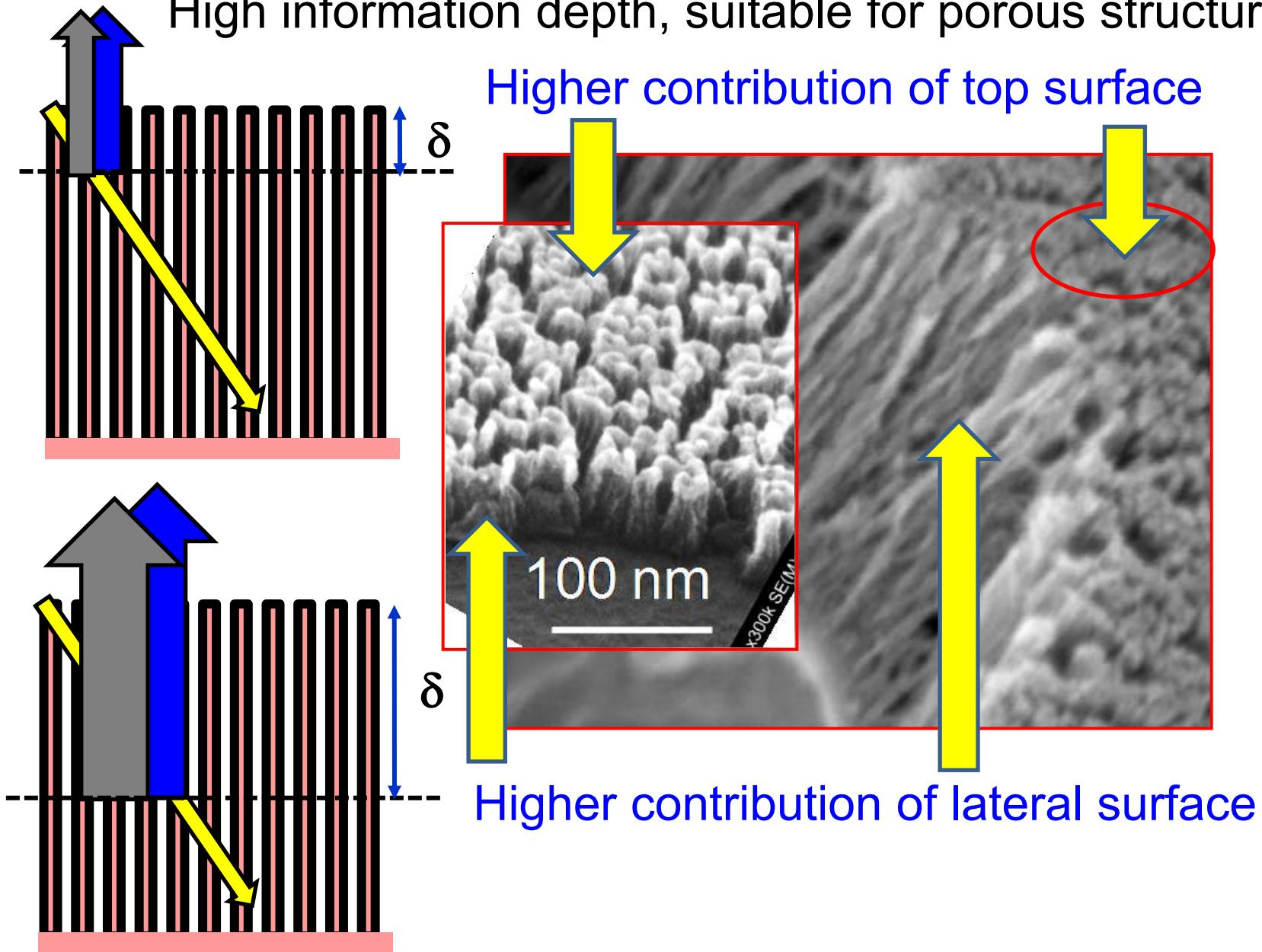
0.4 nm

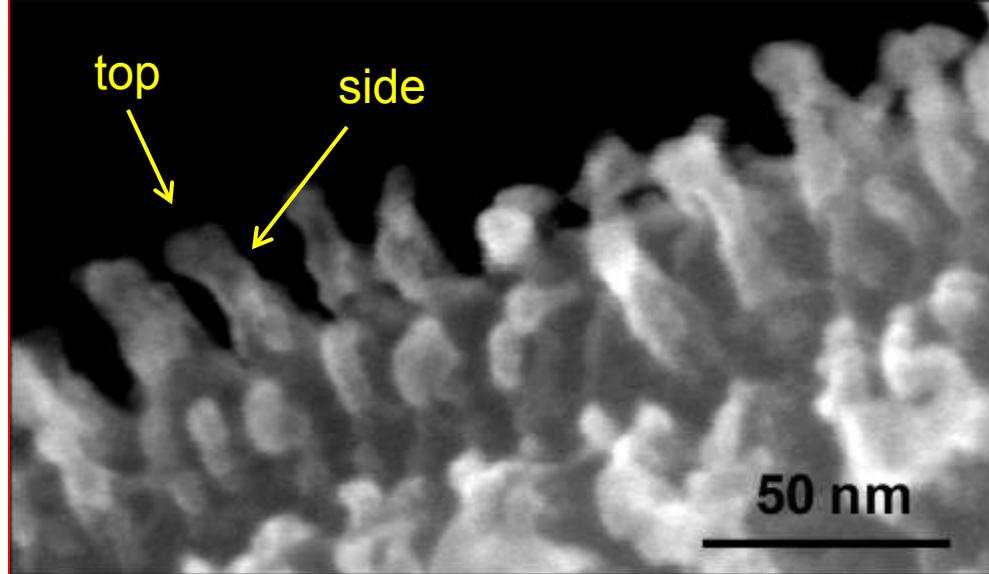
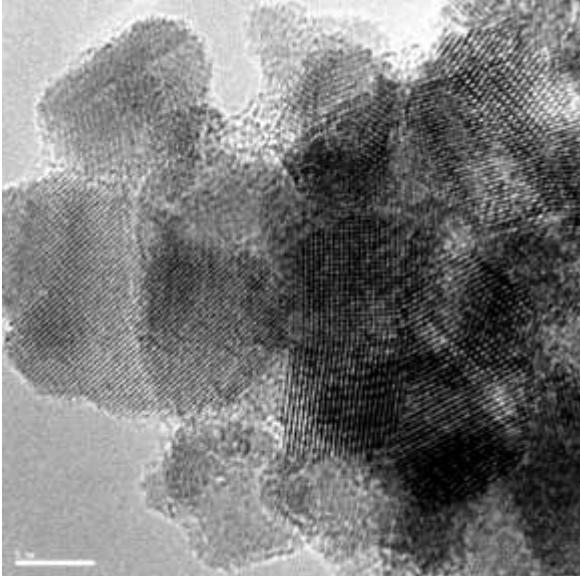


Surface is Pt²⁺ rich

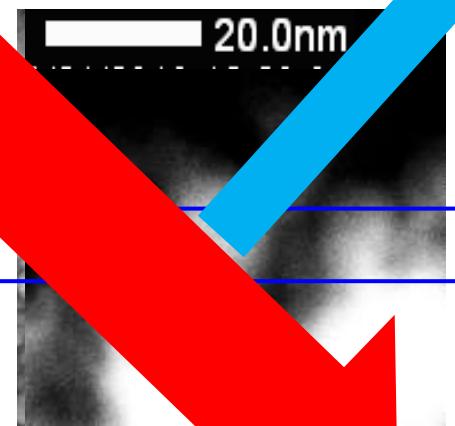
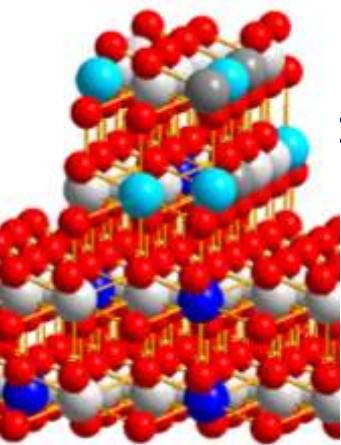
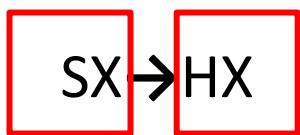
Hard X-ray Photoelectron Spectroscopy

High information depth, suitable for porous structures





nanopowder



el.

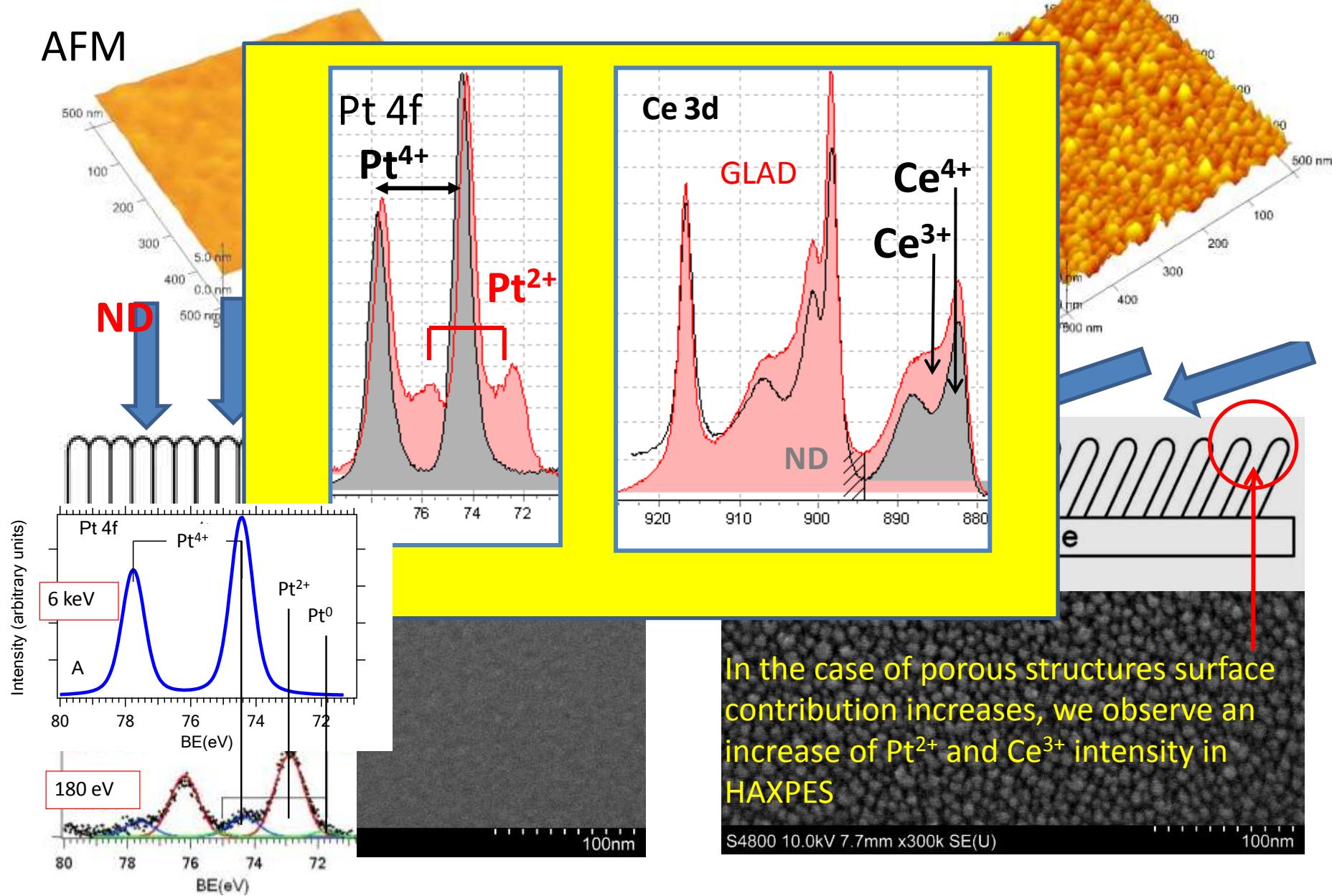
nanorods

Catalytic activity of top and side sites might be different due to growth mechanism

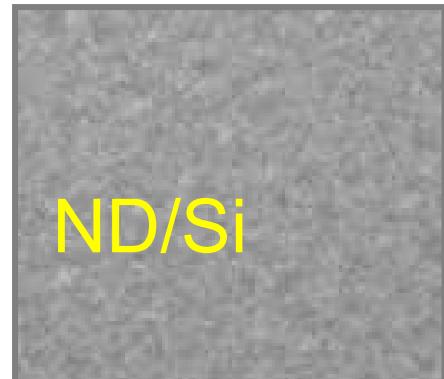
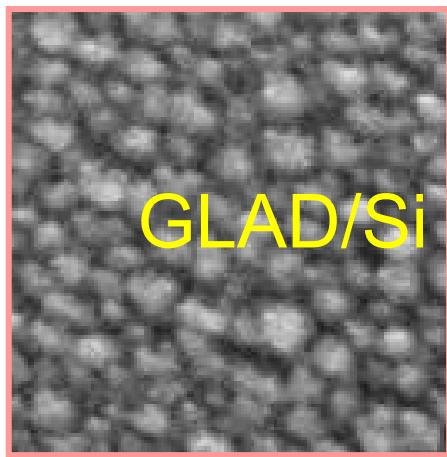
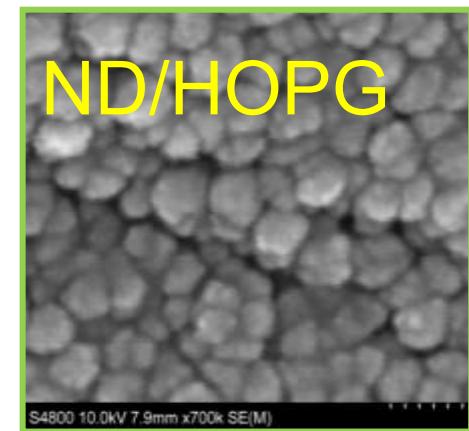
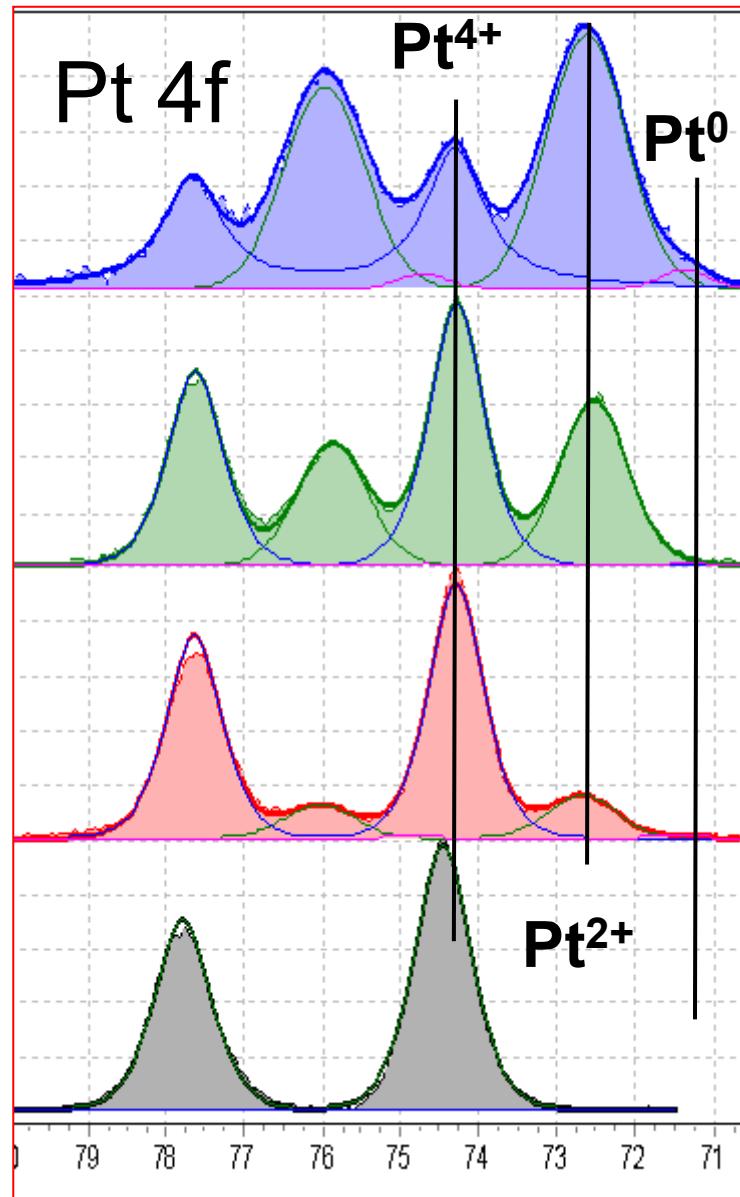
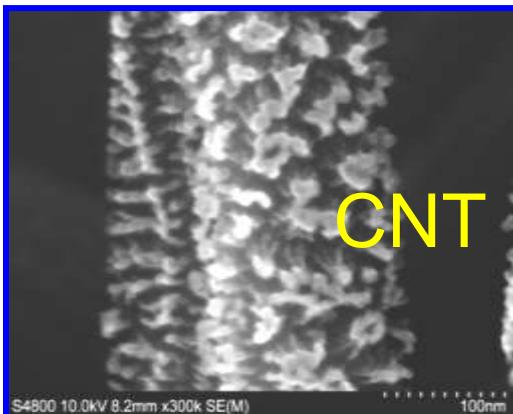
HAXPES makes possible to investigate catal. properties of in-pore catalytic sites

Co-sputtering – growth of mixed oxides (solid solution)

Substrate: SiO_2 , ND x GLAD conditions,

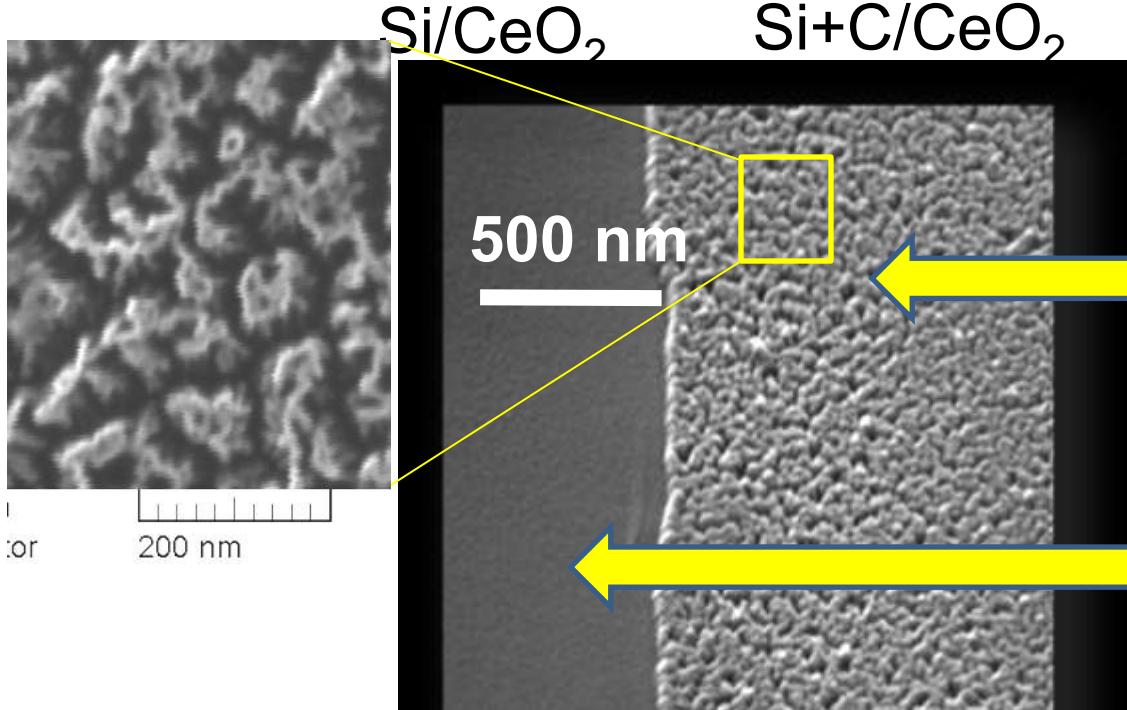


Morphology-substrate determined chemical state of Pt

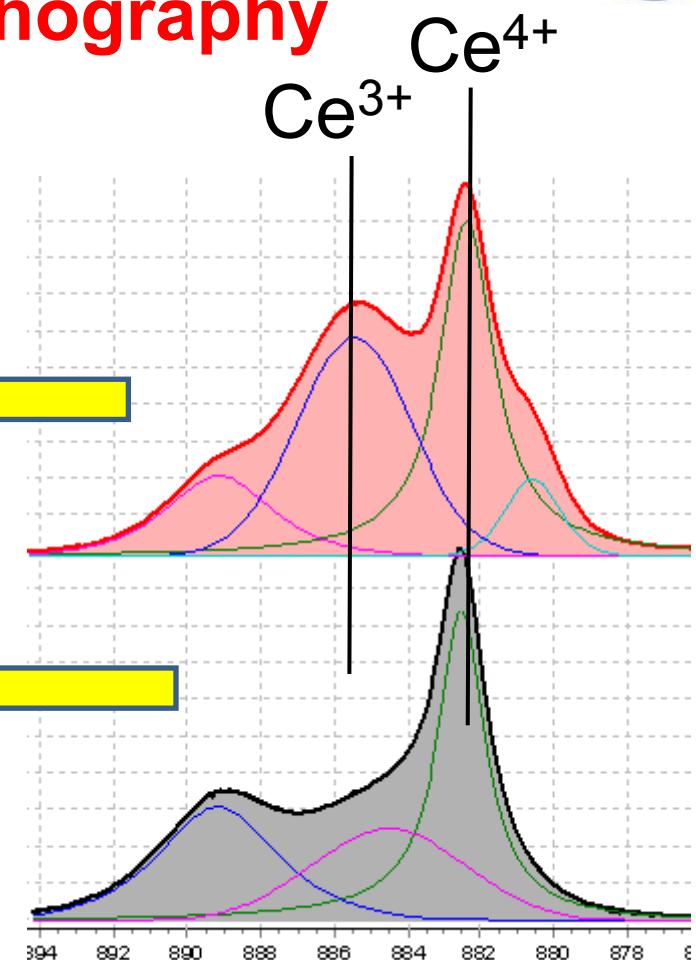
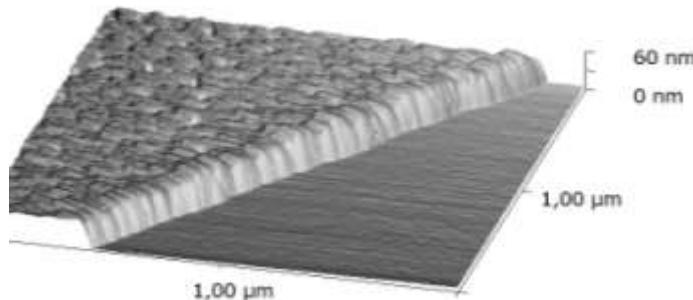


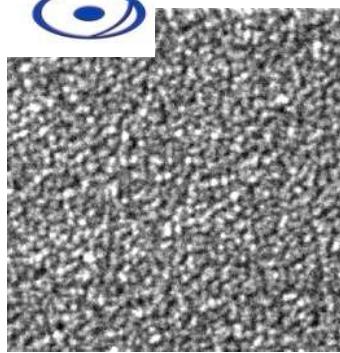
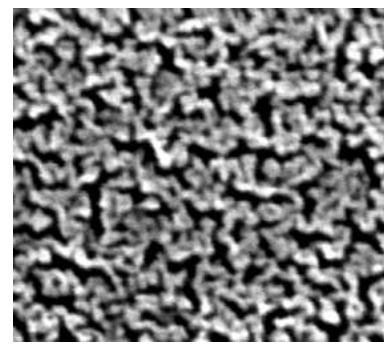
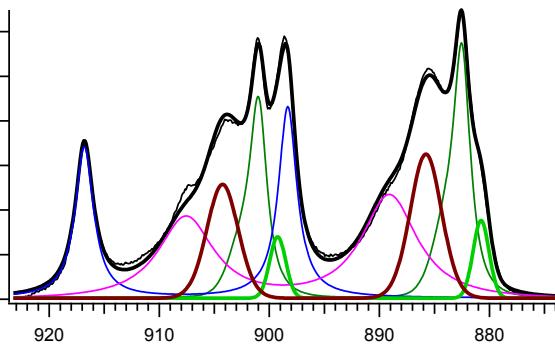
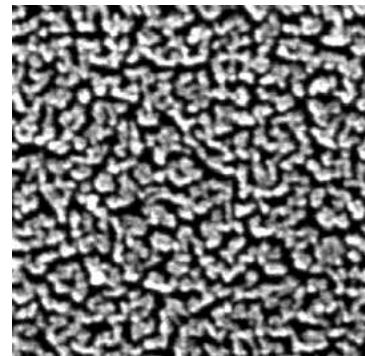
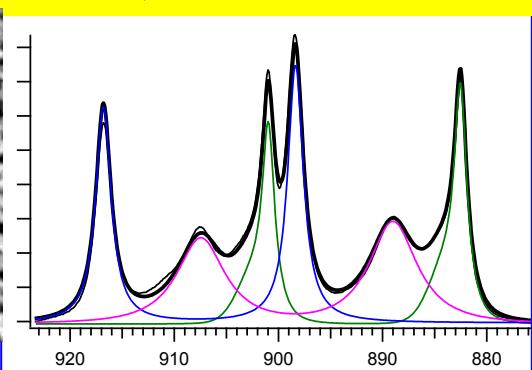
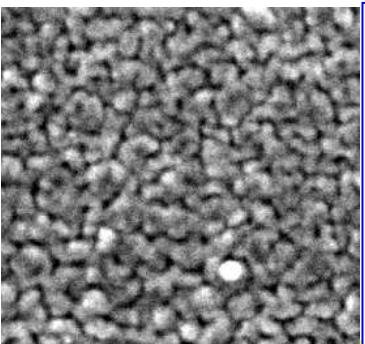
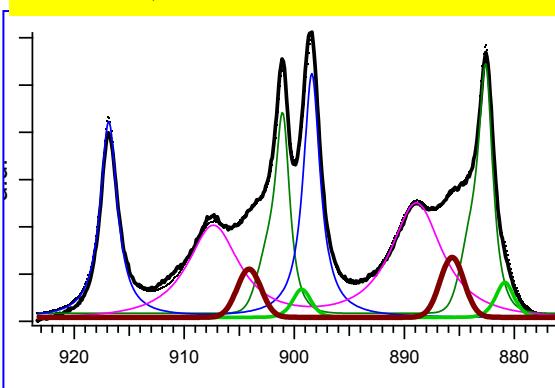
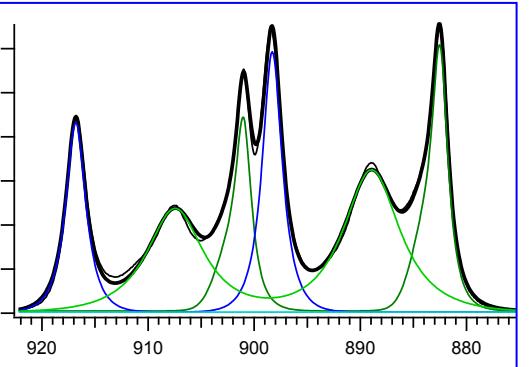
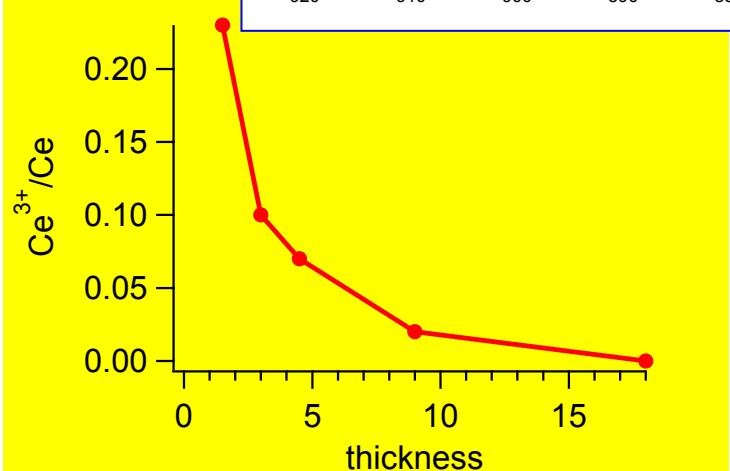
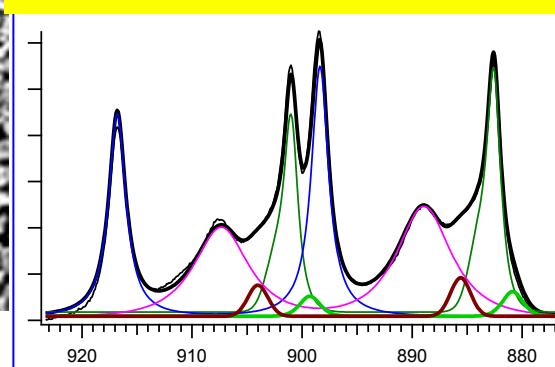
WHY POROUS FILMS ARE GROWN ON C???

CeO₂ growth on 5 nm sputtered carbon film prepared by electron beam lithography

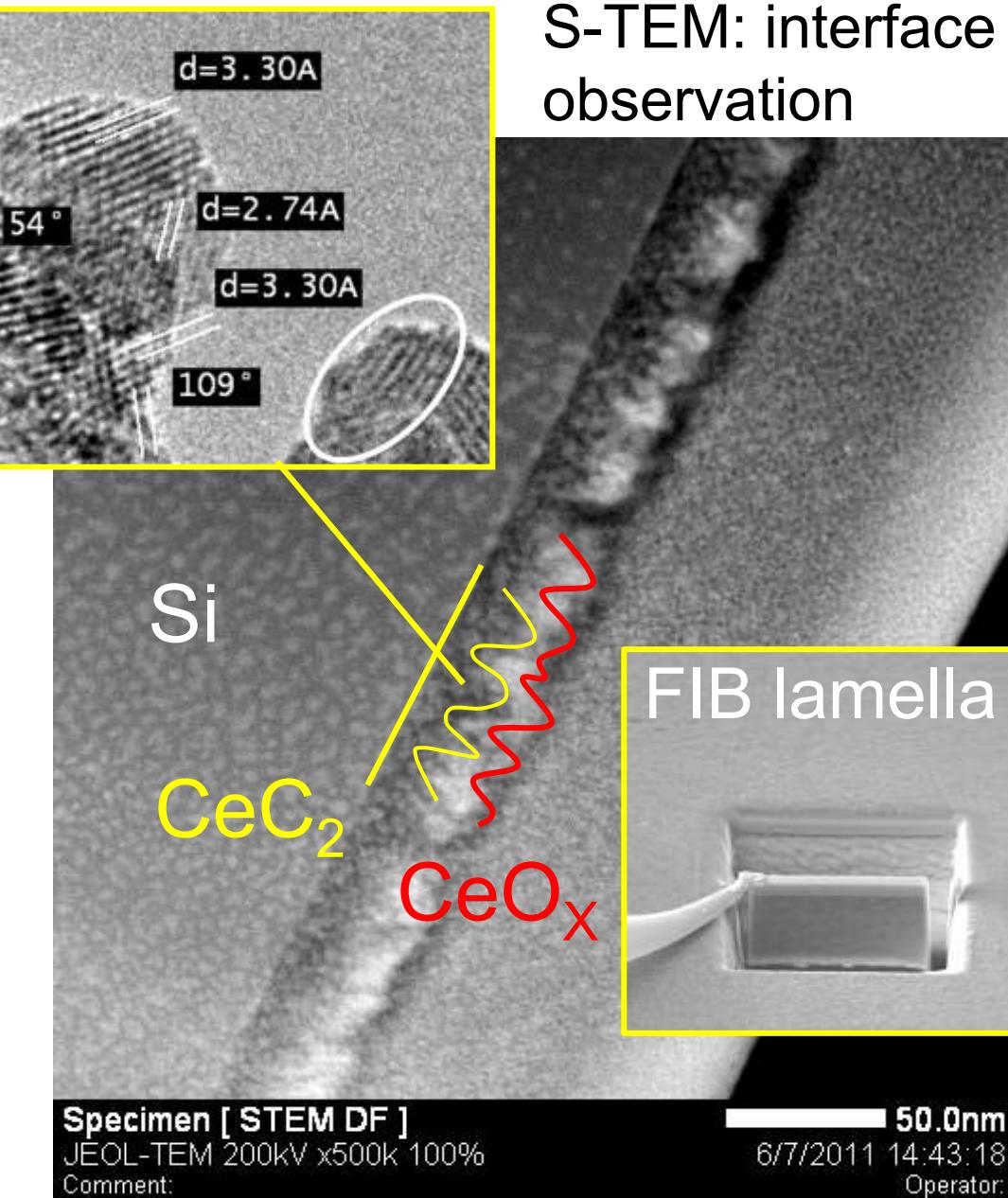


C (EBL)
Si



1.5 nm, $\text{Ce}^{3+}/\text{Ce} = 0.23$ 9 nm, $\text{Ce}^{3+}/\text{Ce} = 0.02$ 3 nm, $\text{Ce}^{3+}/\text{Ce} = 0.10$ 18 nm, $\text{Ce}^{3+}/\text{Ce} = 0$ 4.5 nm, $\text{Ce}^{3+}/\text{Ce} = 0.07$ 

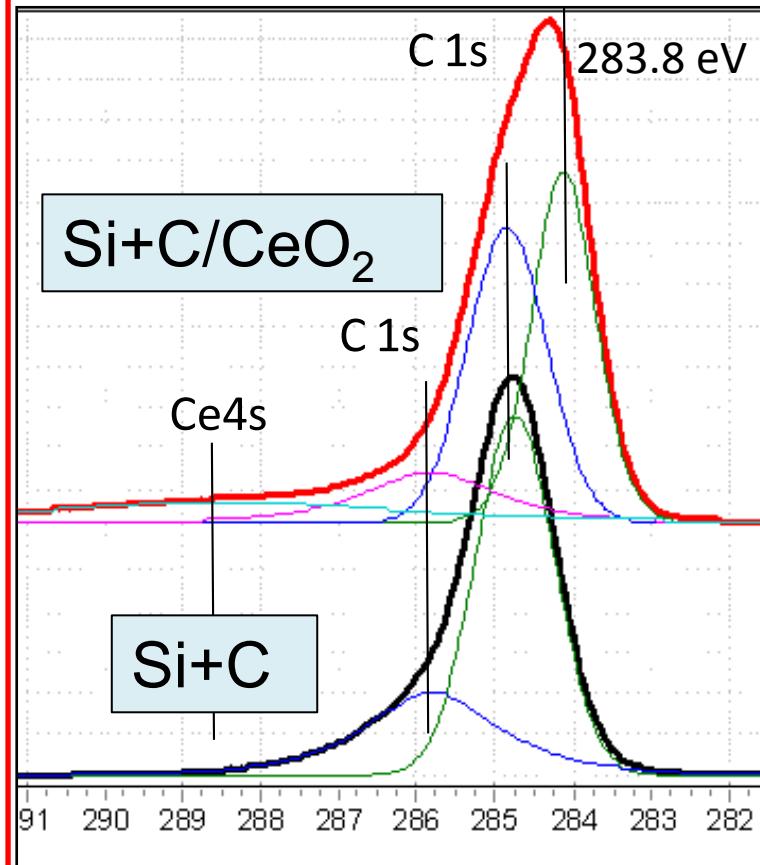
HRTEM: CeC₂ zone axes

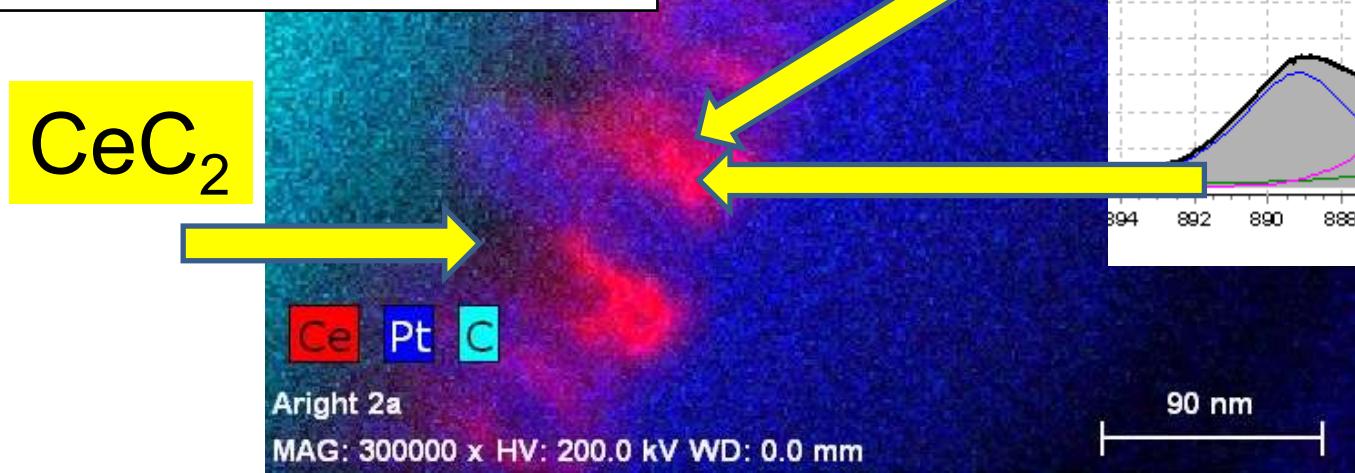
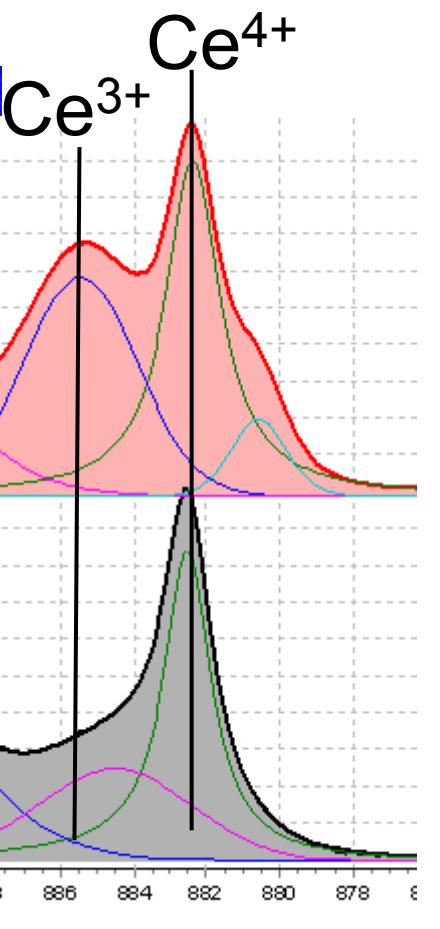
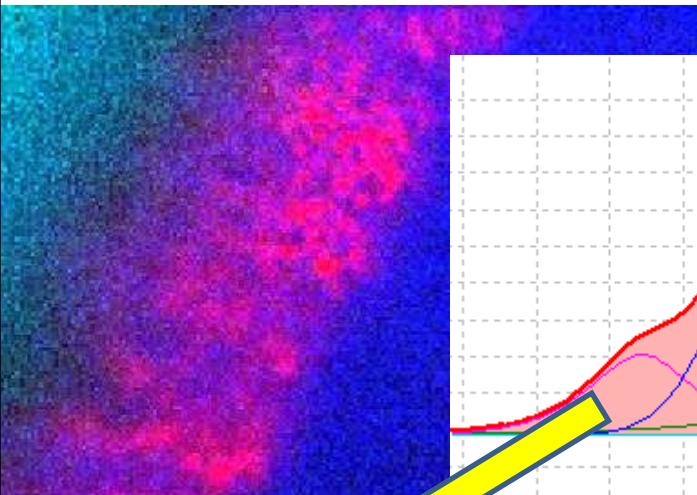
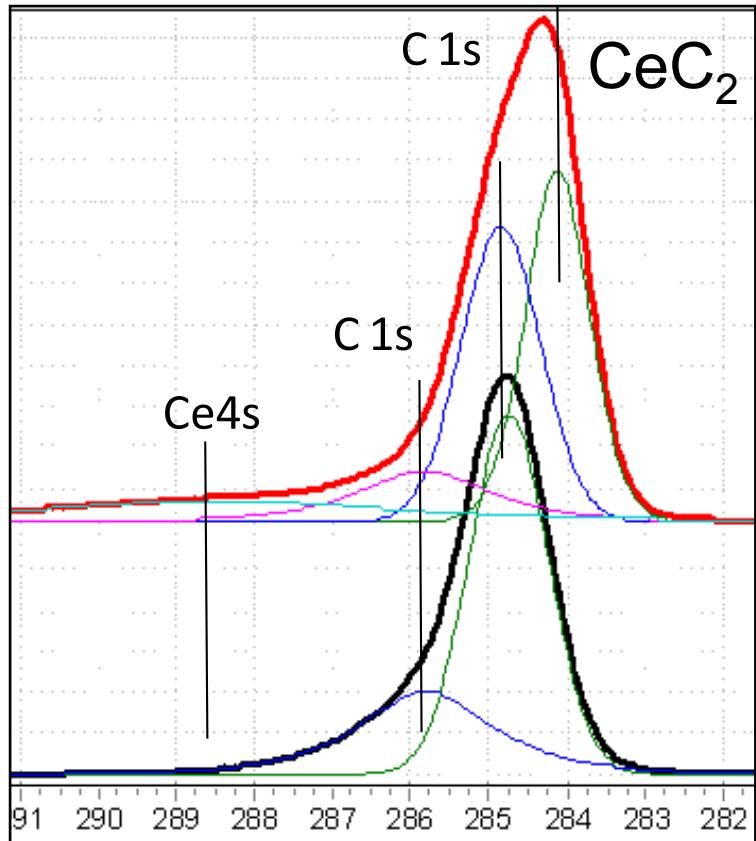


S-TEM: interface observation

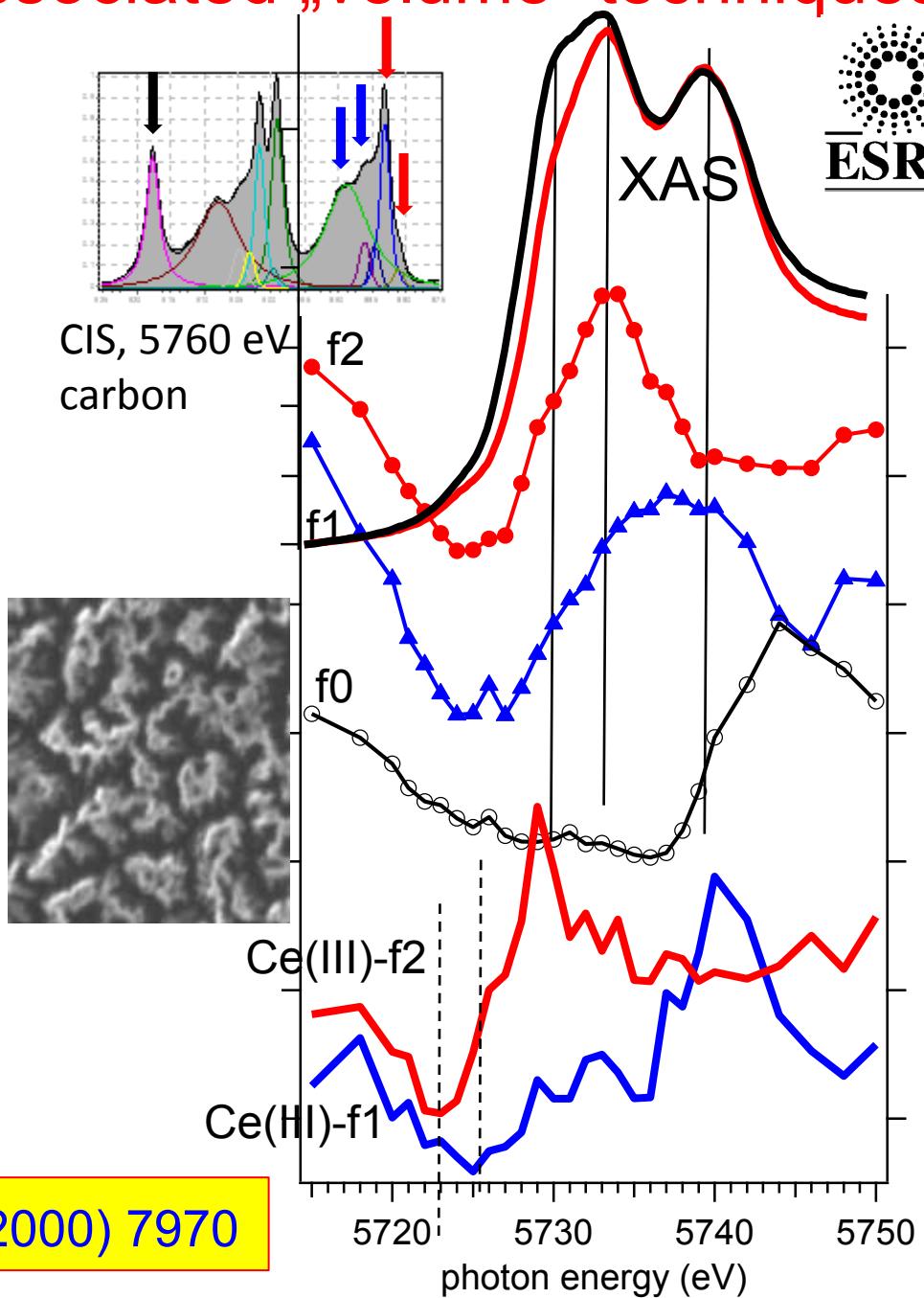
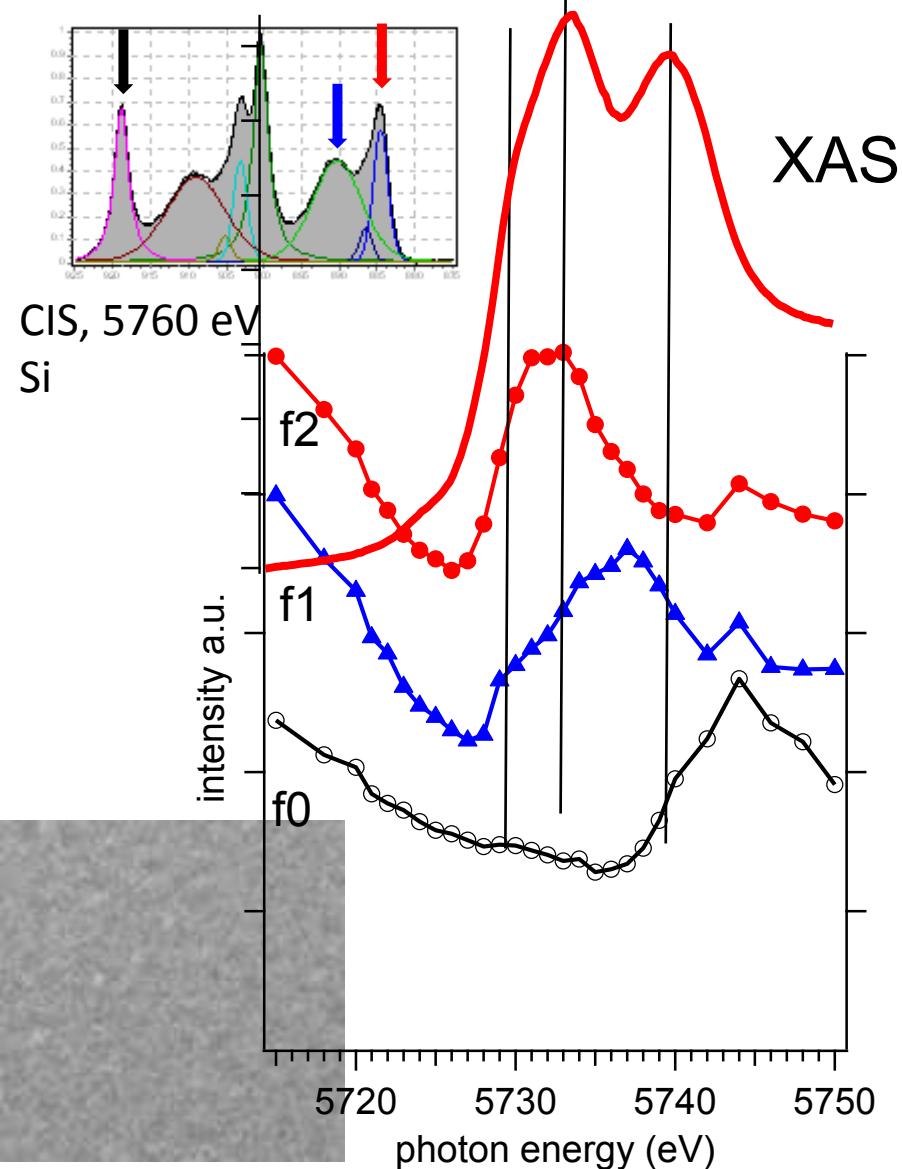
HAXPES interface analysis: C 1s

CeC₂





Future prospects: use of associated „volume“ techniques



XAS and Ce L₃ RPES, PRB 62 (2000) 7970

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