

# Phenomena at High X-ray Intensity: Part 2

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DESY Summer Student Program 2023

August 11, 2023

Hamburg, Germany





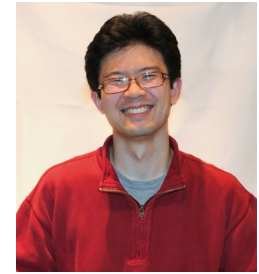
Yajiang Hao



Ludger Inhester



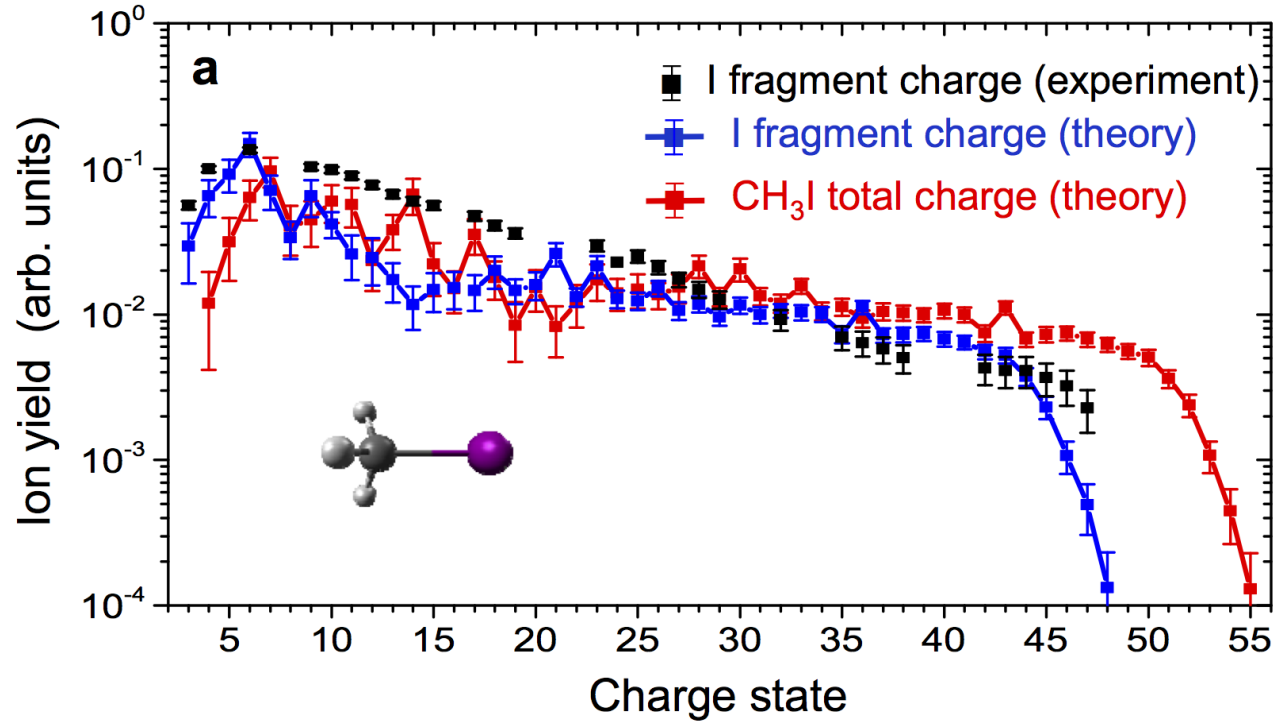
Kota Hanasaki



Sang-Kil Son

- > An ab-initio electronic-structure approach dedicated to ionization dynamics of molecules
- > Self-consistent-field calculation for every electronic configuration formed during interaction with intense XFEL pulse
- > Demonstration of a new ionization enhancement mechanism

# The highest charge states ever produced using light!



Photon energy: 8.3 keV

X-ray peak intensity:  
>  $10^{19}$  W/cm<sup>2</sup>

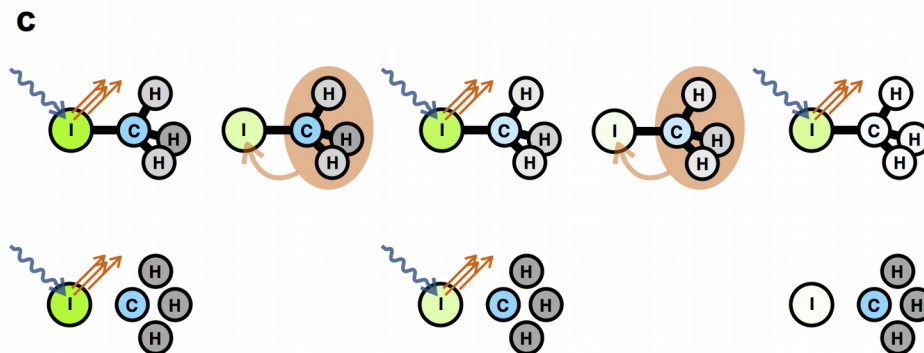
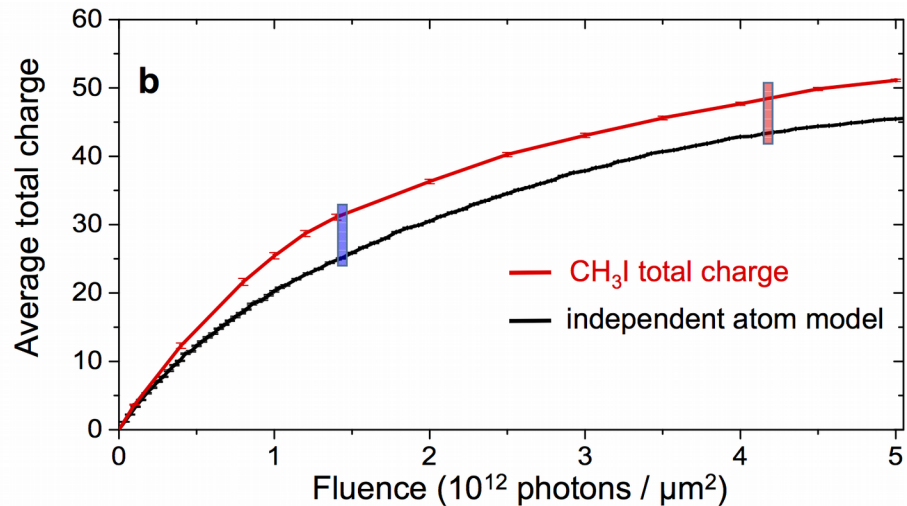
Experimental data taken by Artem Rudenko, Daniel Rolles, and collaborators



A. Rudenko *et al.*, Nature **546**, 129 (2017).



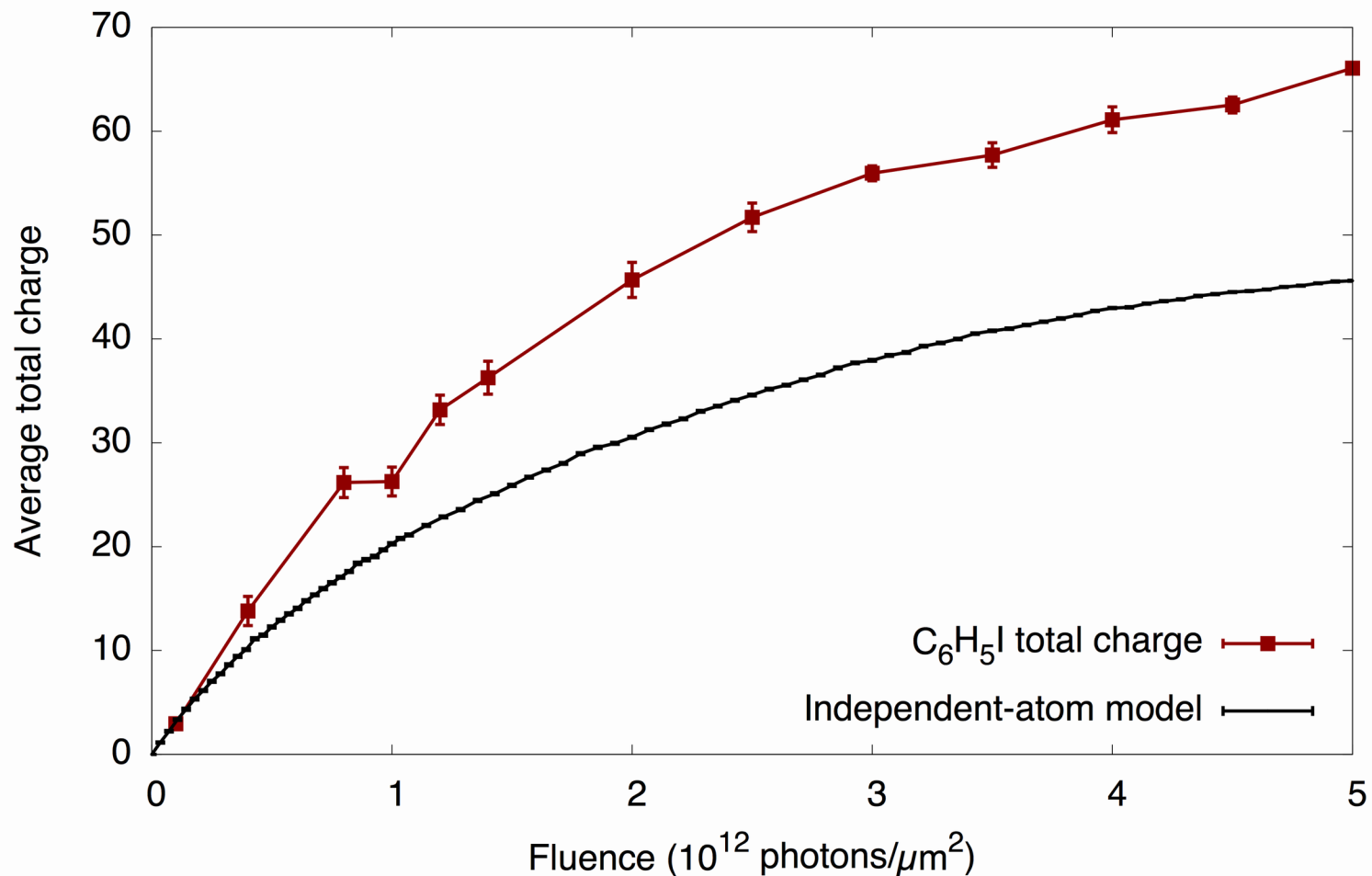
# New ionization enhancement mechanism (molecular effect!)



A. Rudenko *et al.*, Nature **546**, 129 (2017).



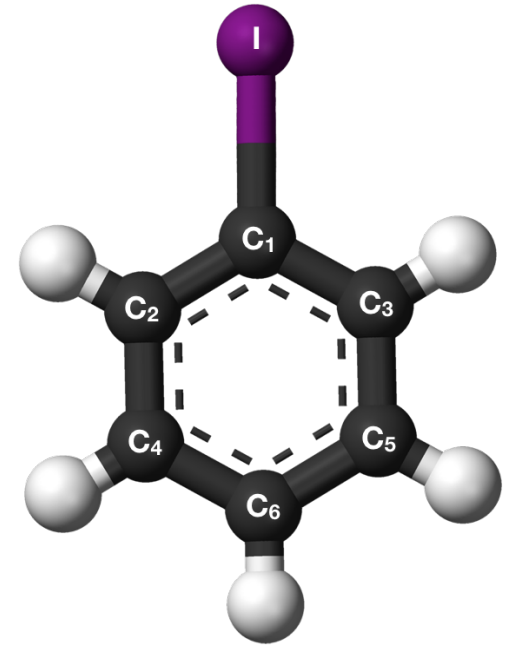
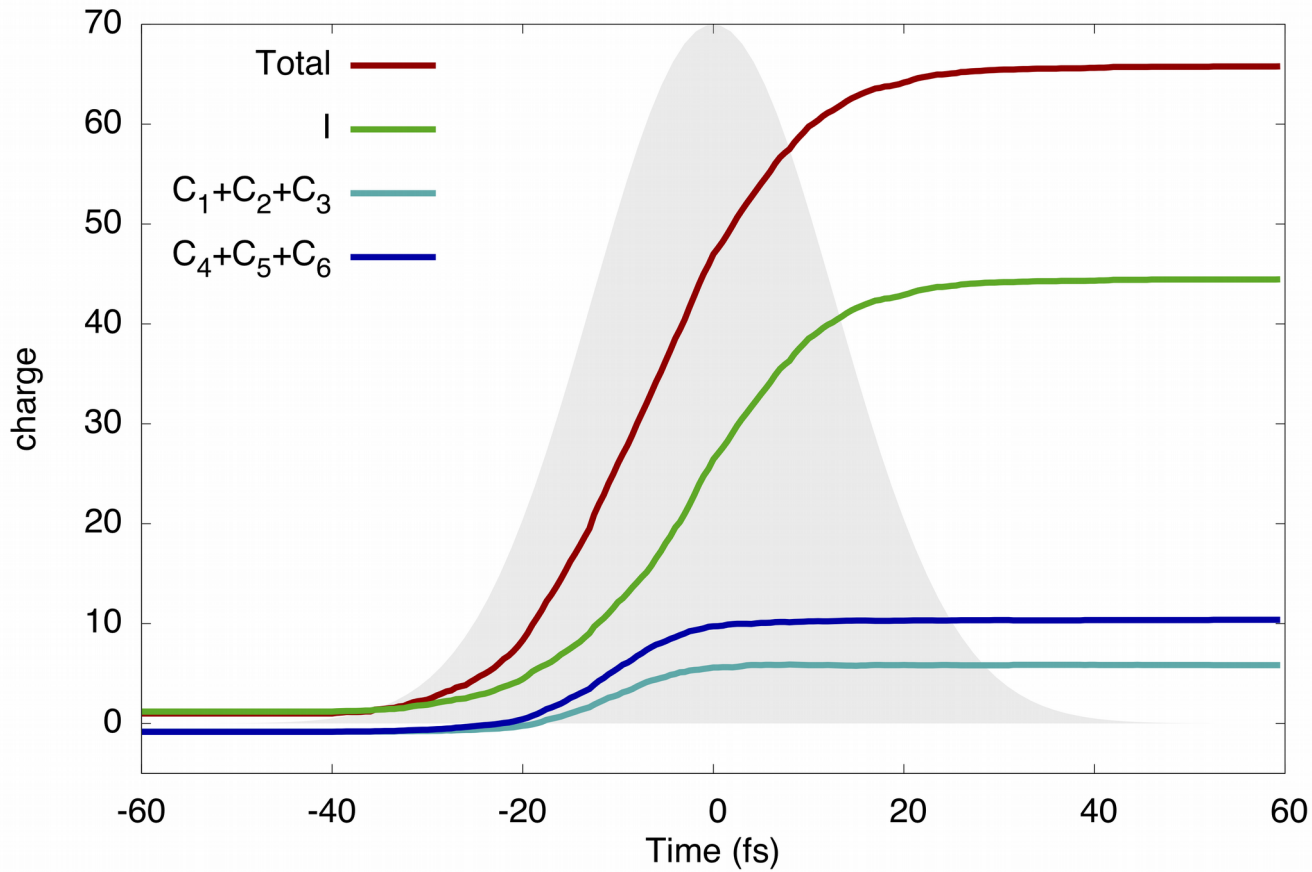
# Iodobenzene (photon energy 8.3 keV)



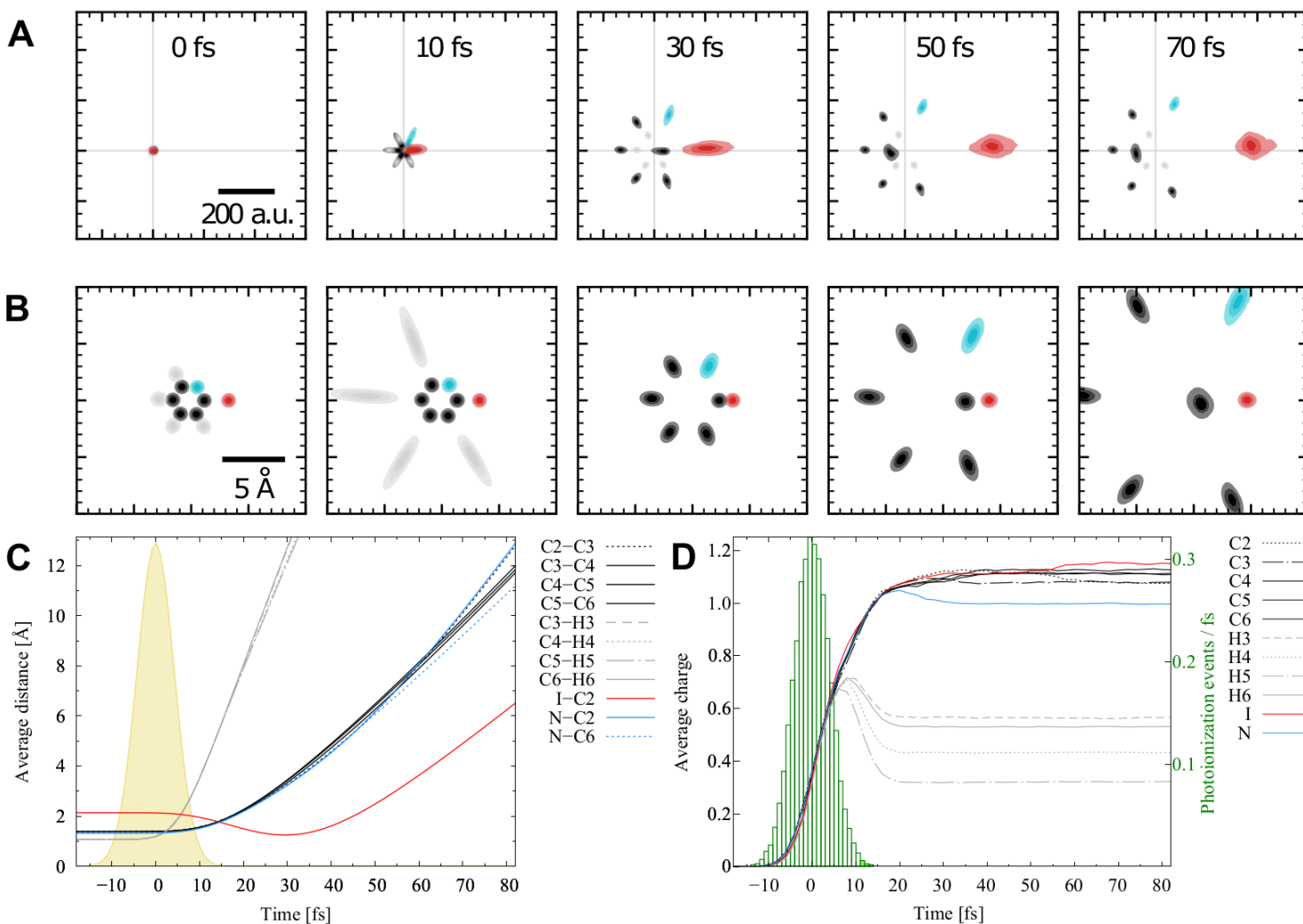
Y. Hao *et al.*, Phys. Rev. A **100**, 013402 (2019).



# Ionization dynamics in iodobenzene (photon energy 8.3 keV, fluence $5 \times 10^{12}$ photons/ $\mu\text{m}^2$ )

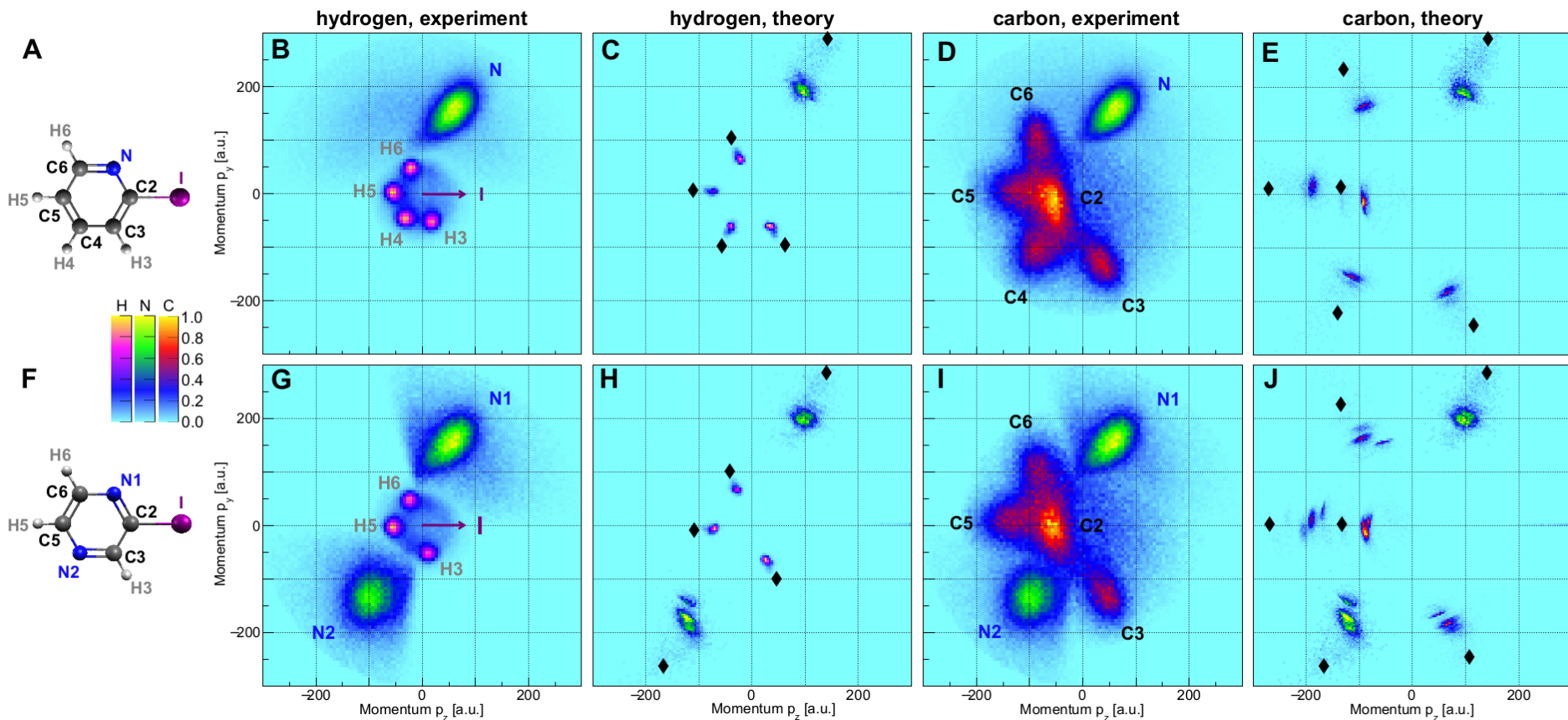


# Single-molecule imaging via XFEL-driven Coulomb explosion: theory for 2-iodopyridine



R. Boll *et al.*, Nature Phys. **18**, 423 (2022).

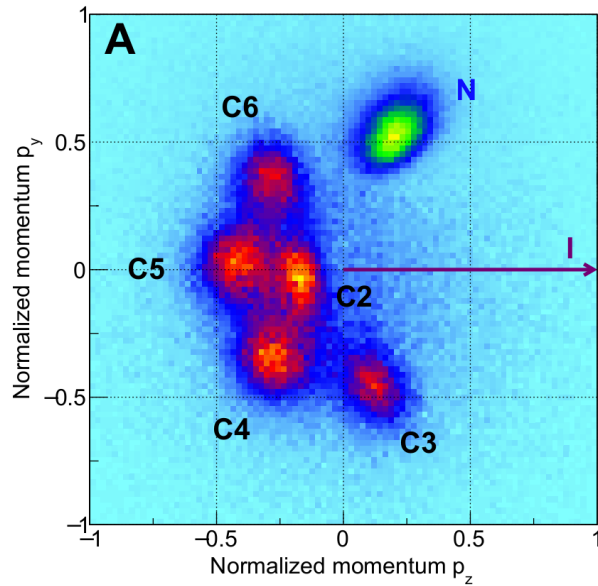
# High-quality Coulomb explosion imaging at the European XFEL



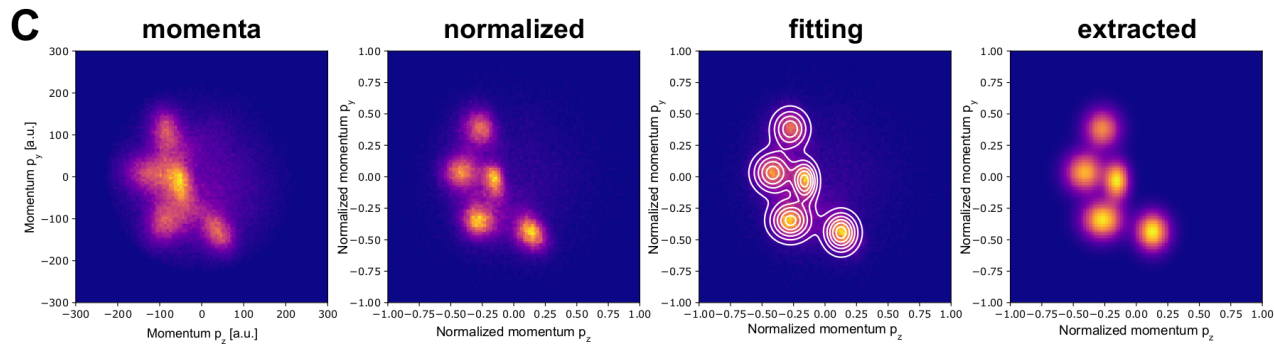
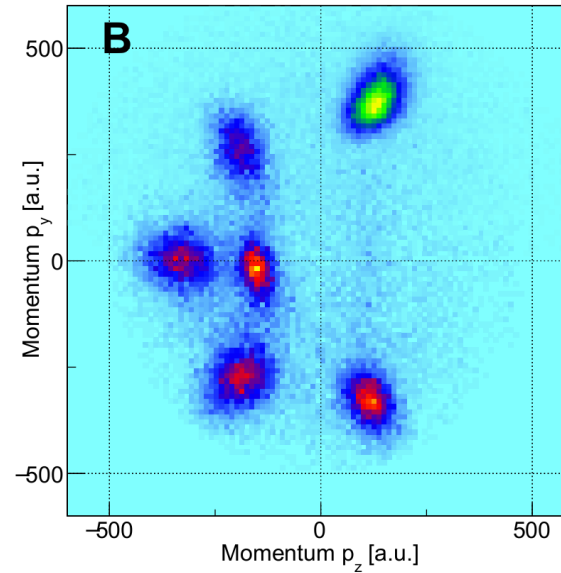
R. Boll *et al.*, *Nature Phys.* **18**, 423 (2022).

# Momentum-space normalization and extraction of site-specific charge abundances

$I^+$ ,  $N^+$ ,  $C^+$  coincidences

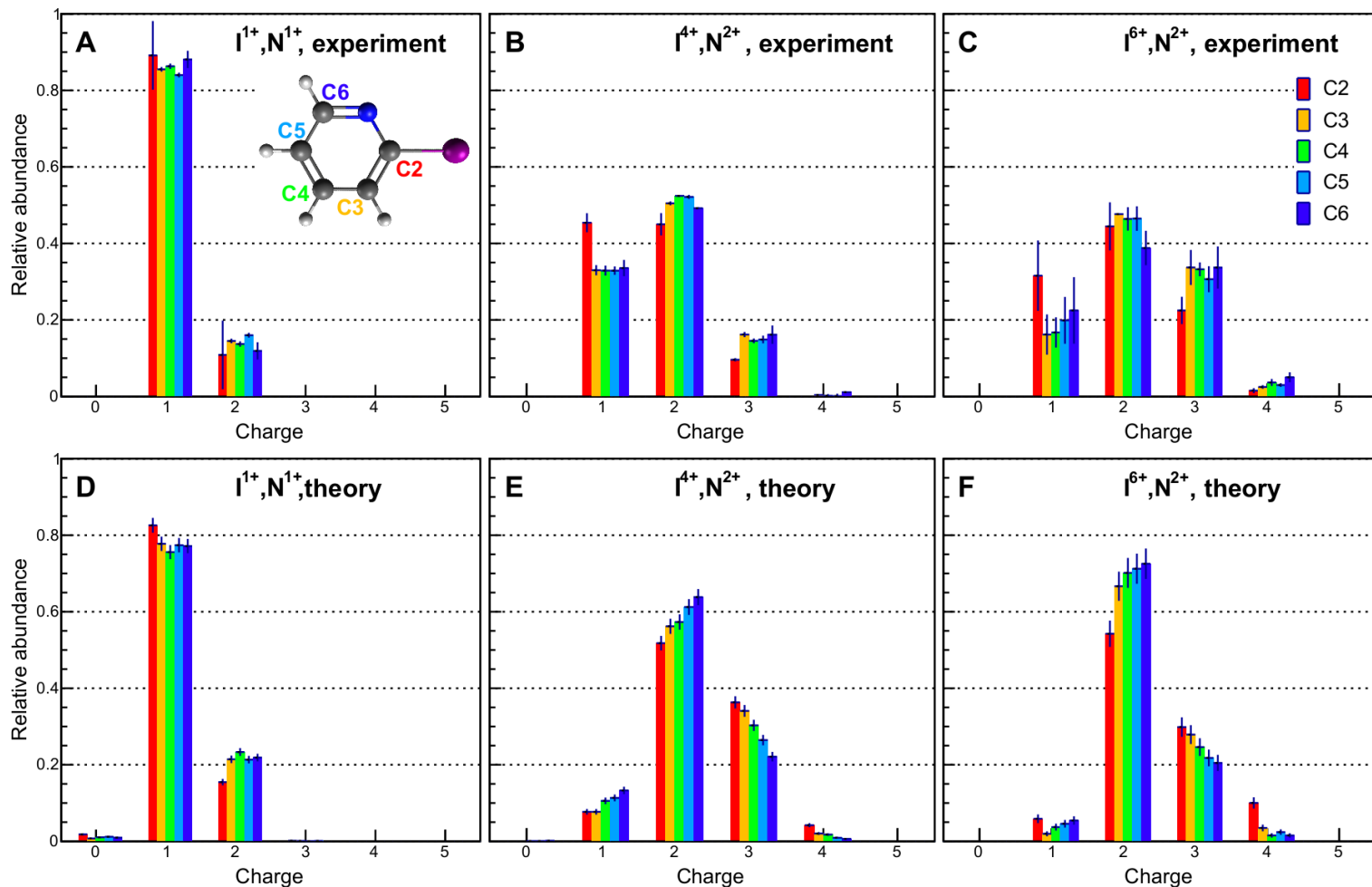


$I^{4+}$ ,  $N^{2+}$ ,  $C^{2+}$  coincidences



R. Boll *et al.*, Nature Phys. **18**, 423 (2022).

# Charge state distributions of carbon atoms



R. Boll *et al.*, Nature Phys. **18**, 423 (2022).



# Acknowledgment



**Rebecca Boll 1, Julia M. Schäfer 2,3**, Benoit Richard 2,4,5, Kilian Fehre 6, Gregor Kastirke 6, Zoltan Jurek 2, Markus S. Schöffler 6, Malik M. Abdullah 5, Nils Anders 6, Thomas M. Baumann 1, Sebastian Eckart 6, Benjamin Erk 7, Alberto De Fanis 1, Reinhard Dörner 6, Sven Grundmann 6, Patrik Grychtol 1, Alexander Hartung 6, Max Hofmann 6, Markus Ilchen 1,8, Ludger Inhester 2, Christian Janke 6, Rui Jin 2,9, Max Kircher 6, Katharina Kubicek 1,5, Maksim Kunitski 6, Xiang Li 10, Tommaso Mazza 1, Severin Meister 11, Niklas Melzer 6, Jacobo Montano 1, Valerija Music 1,8, Giammarco Nalin 6, Yevheniy Ovcharenko 1, Christopher Passow 7, Andreas Pier 6, Nils Rennhack 1, Jonas Rist 6, Daniel E. Rivas 1, Daniel Rolles 10, Ilme Schlichting 12, Lothar Ph. H. Schmidt 6, Philipp Schmidt 1,8, Juliane Siebert 6, Nico Strenger 6, Daniel Trabert 6, Florian Trinter 6,7,13, Isabel Vela-Perez 6, Rene Wagner 1, Peter Walter 14, Miriam Weller 6, Pawel Ziolkowski 1, Sang-Kil Son 2, Artem Rudenko 10, Michael Meyer 1, **Till Jahnke 1**

1: European XFEL

2: CFEL, DESY

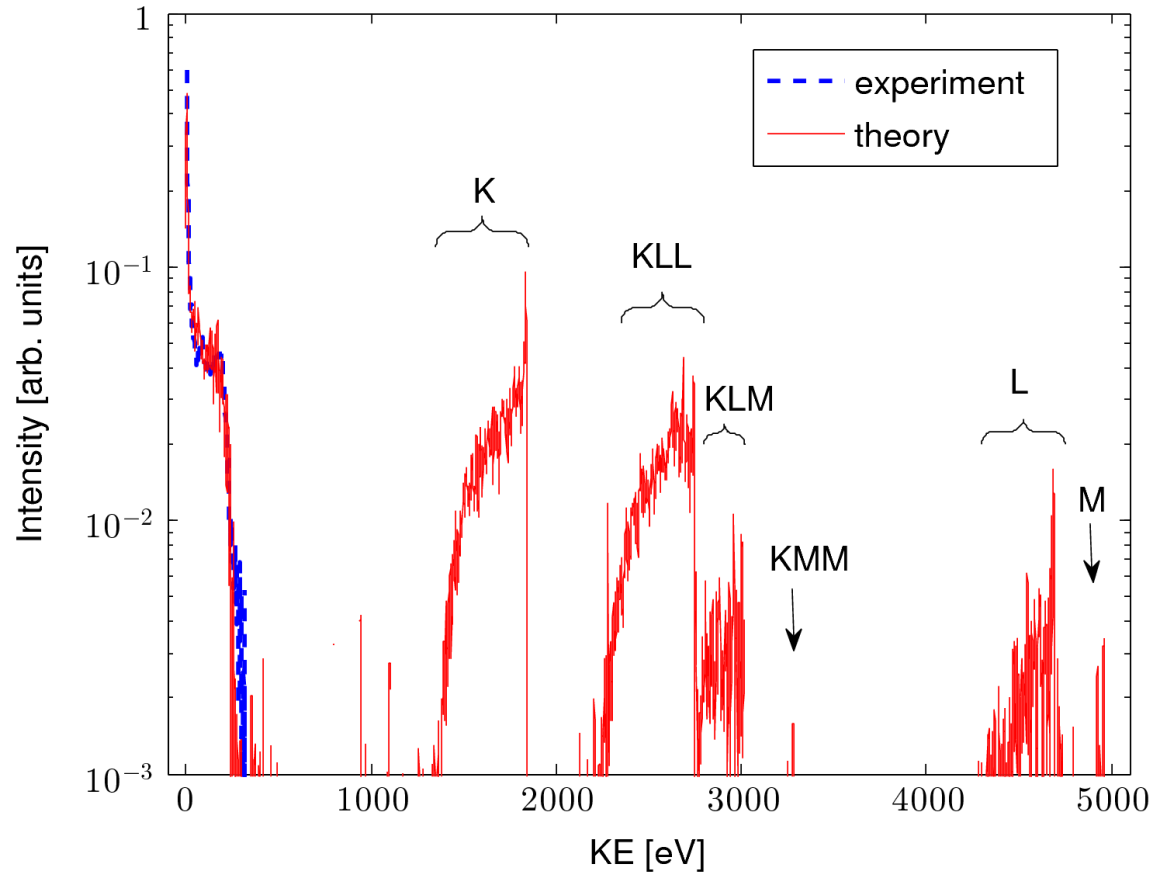


# Argon clusters @ SACLA (Kiyoshi Ueda *et al.*)

> Theoretical and experimental electron kinetic energy spectra,

5 keV, 30 fs

N=1000 atoms



– Experiment

– Theory

**no parameter fitting!**



T. Tachibana *et al.*, Scientific Reports **5**, 10977 (2015).



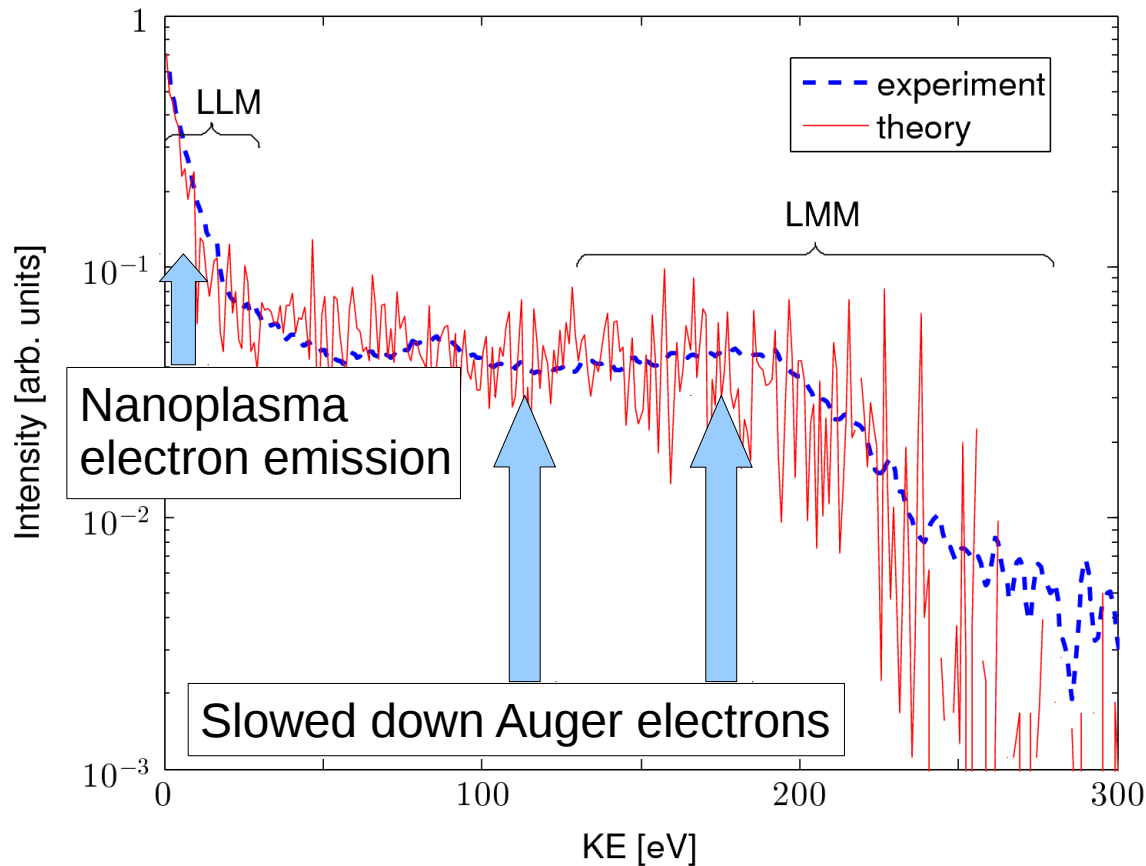


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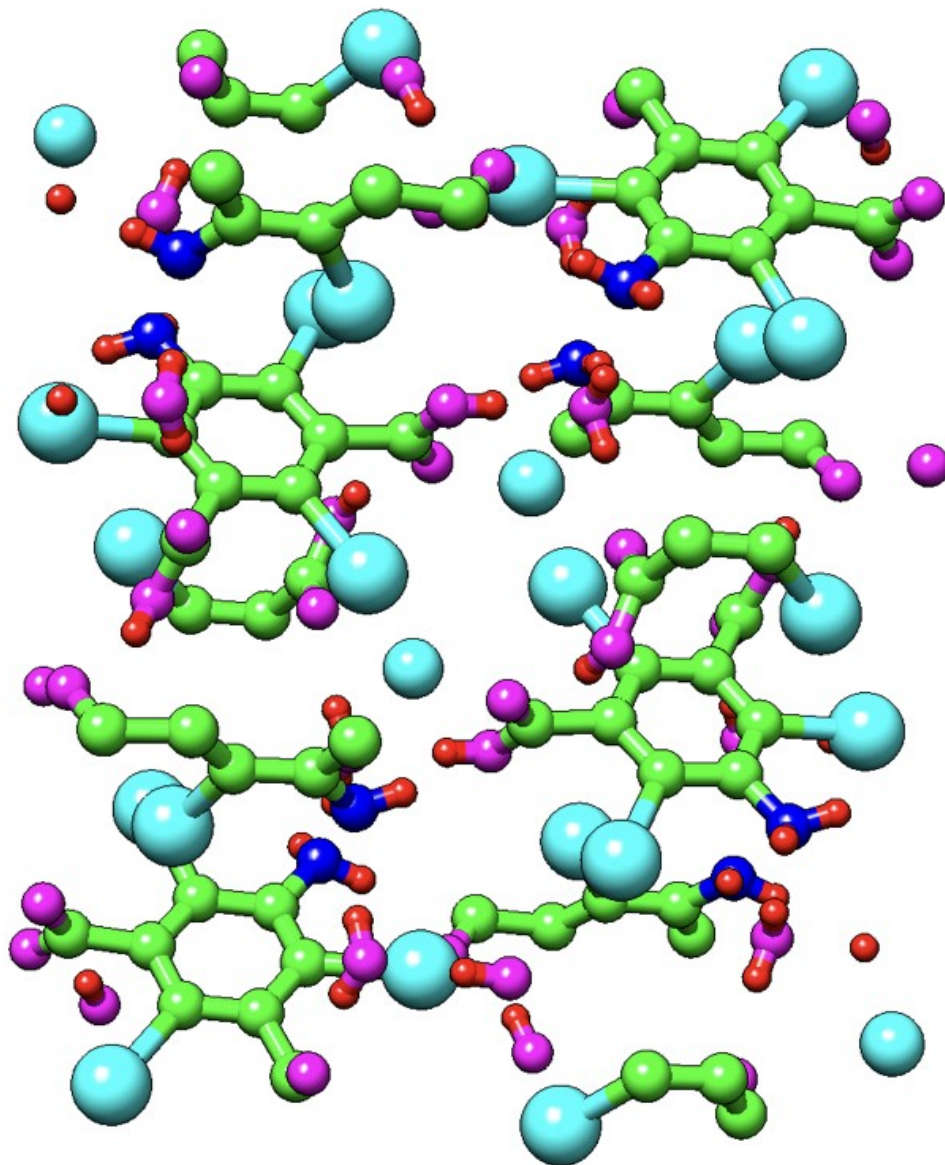
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T. Tachibana *et al.*, Scientific Reports 5, 10977 (2015).

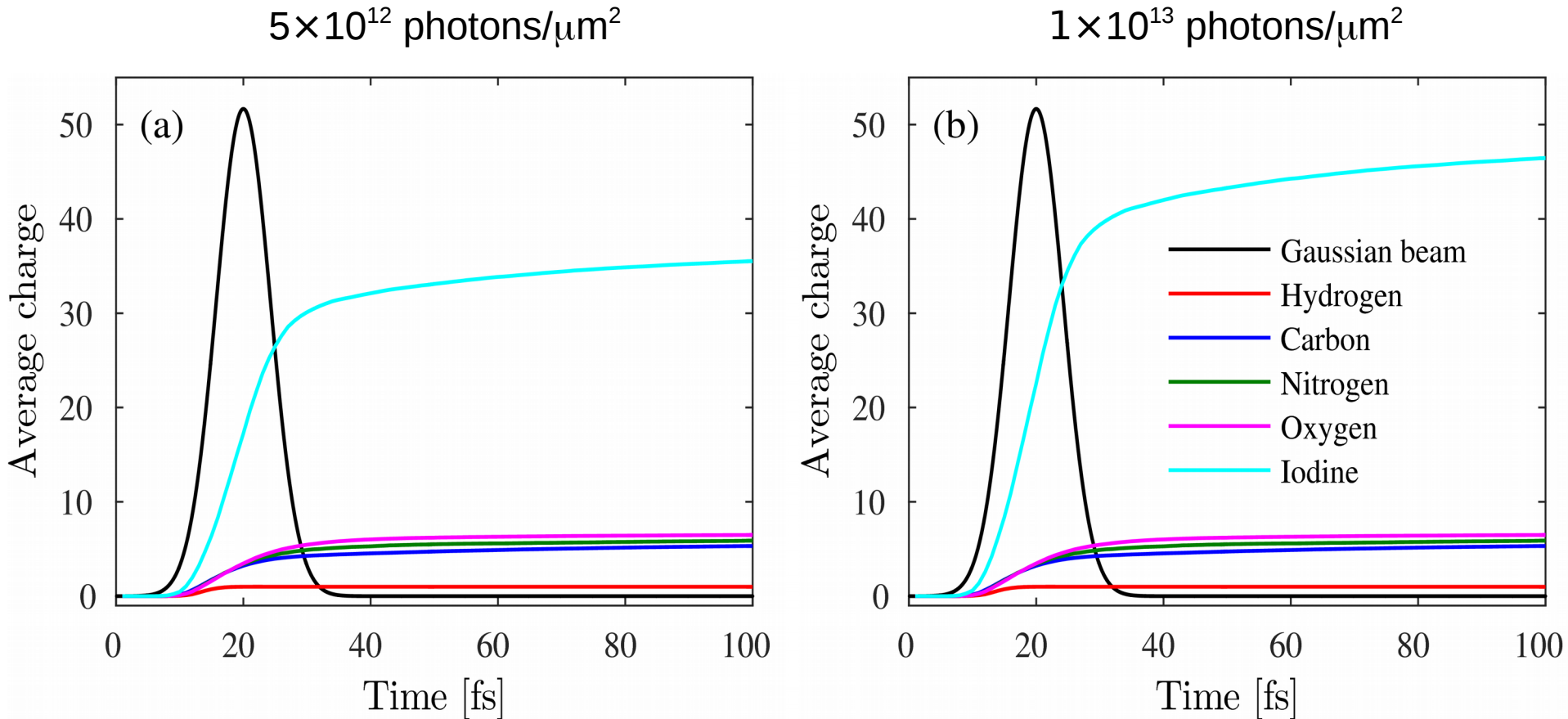


# XMDYN using periodic boundary conditions



I3C crystal  
(5-amino-2,4,6-triiodo-  
isophthalic acid,  
 $C_8H_4I_3NO_4$ )

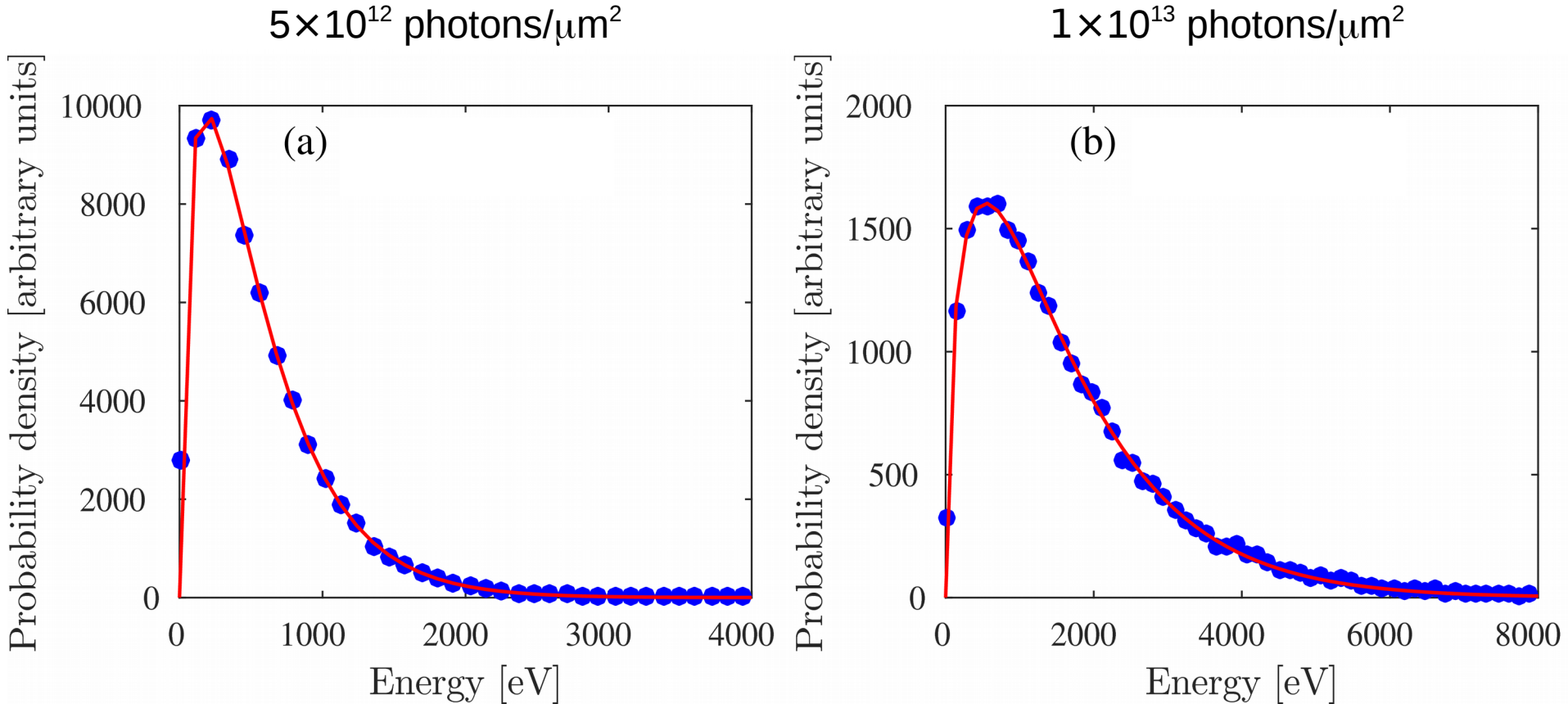
# Ionization dynamics in I3C crystal (photon energy 9.7 keV)



M. M. Abdullah *et al.*, Phys. Rev. E **96**, 023205 (2017).



# Electron thermalization in I3C crystal (250 fs after a 9.7-keV x-ray pulse)

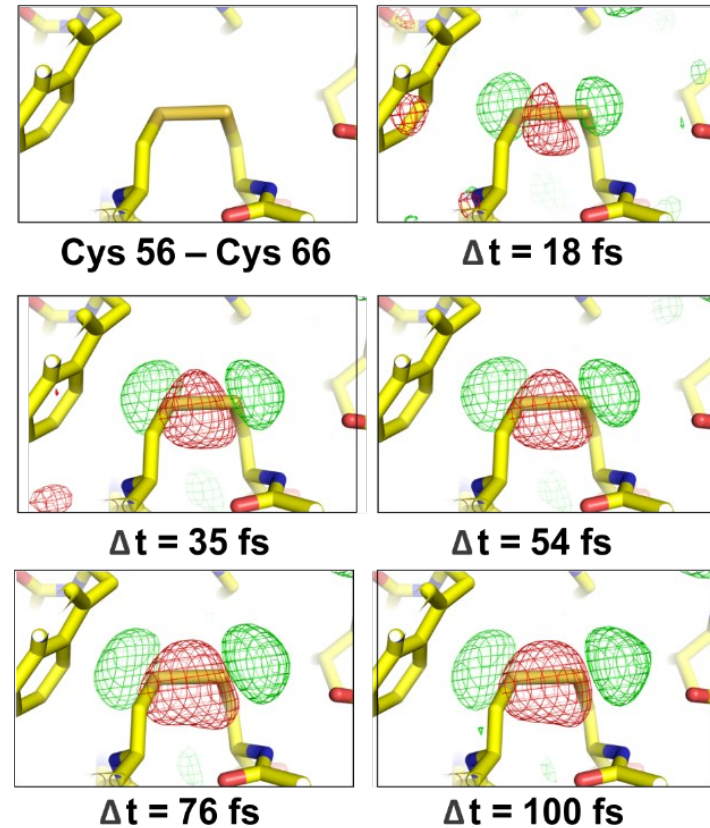
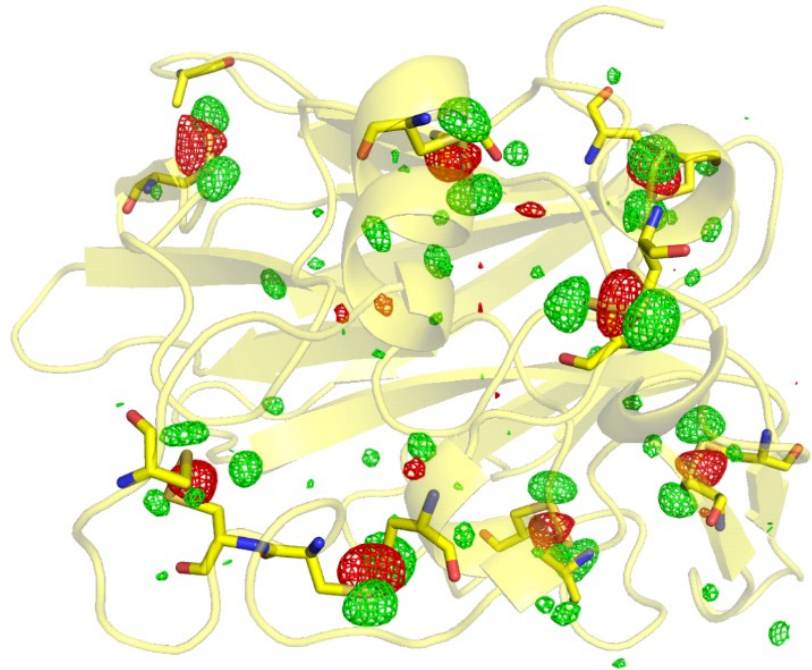


M. M. Abdullah *et al.*, Phys. Rev. E **96**, 023205 (2017).



# X-ray pump / x-ray probe SFX study of thaumatin

Collaboration with **Ilme Schlichting** et al.  
Experiment carried out at LCLS



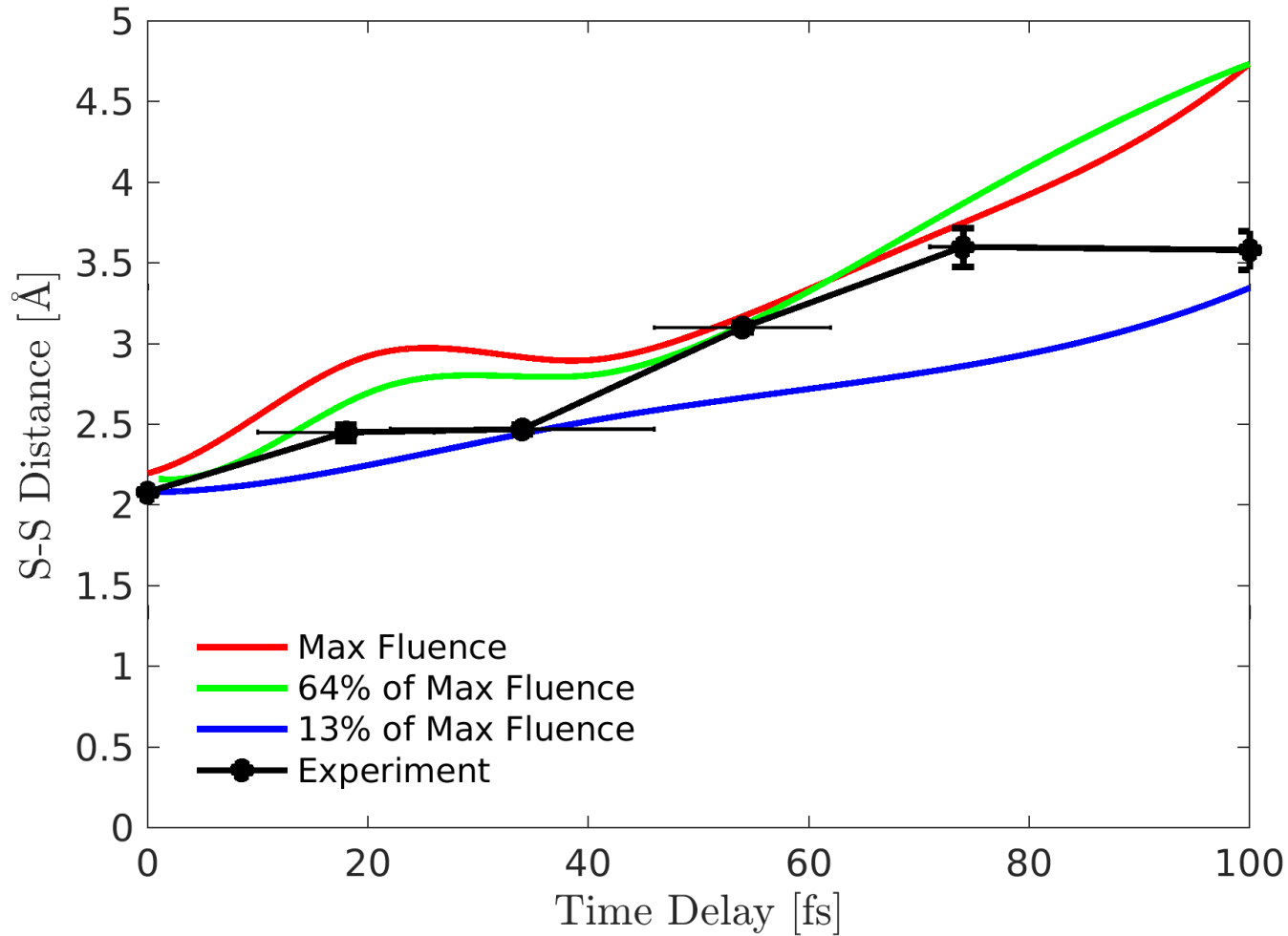
Photon energy: 7.1 keV

Pump (probe) pulse duration: 15 fs (15 fs)

Combined pulse energy: 1 mJ, shared 50%/50% (pump/probe)

Intensity in the focus:  $3 \times 10^{19}$  W/cm<sup>2</sup>

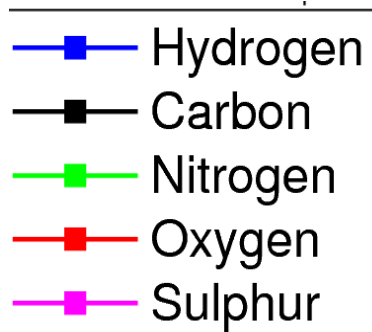
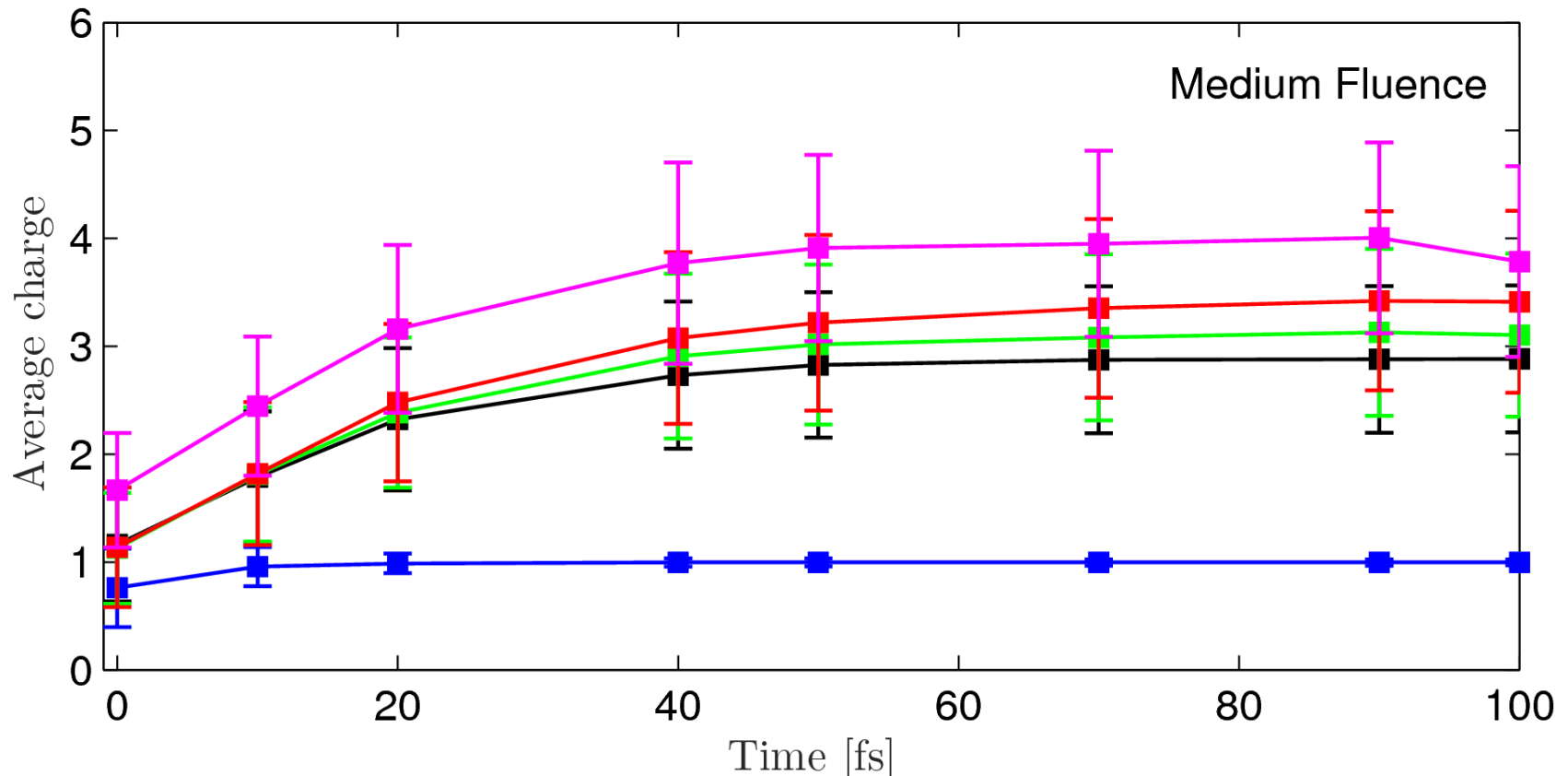
# Disulfide bond length in thaumatin



K. Nass *et al.*, Nature Commun. **11**, 1814 (2020).

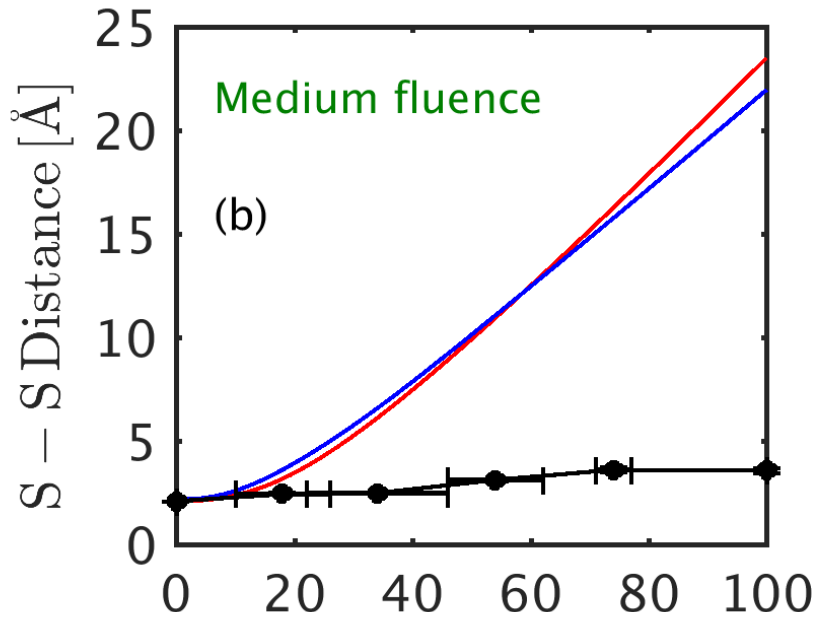


# Average charge as a function of pump-probe delay

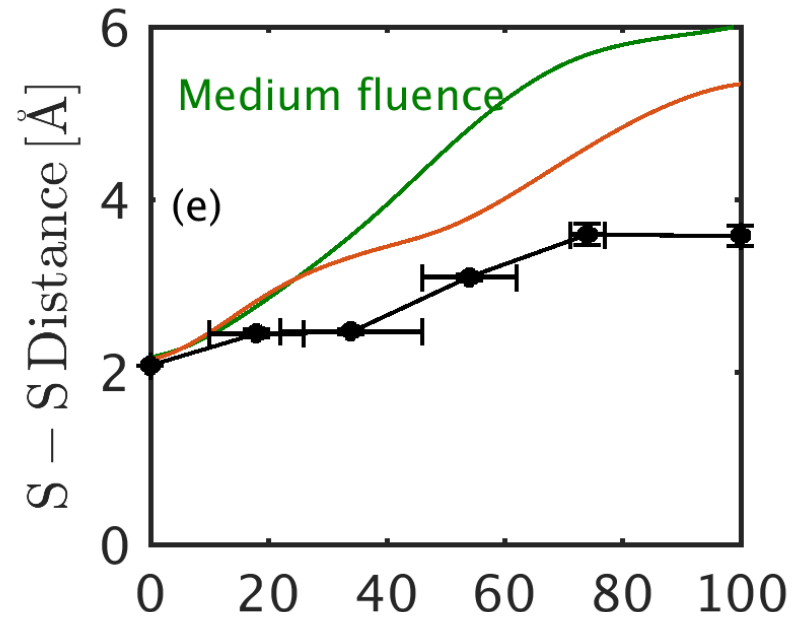


K. Nass *et al.*, Nature Commun. **11**, 1814 (2020).

# Ion caging and plasma screening



**Red:** Isolated S-S pair in vacuum  
**Blue:** Isolated S-S pair using charges in crystal environment



**Brown:** No Coulomb interaction between S atoms and plasma electrons  
**Green:** No Coulomb interaction between S atoms and non-S atoms

K. Nass *et al.*, Nature Commun. **11**, 1814 (2020).



# Effects of radiation



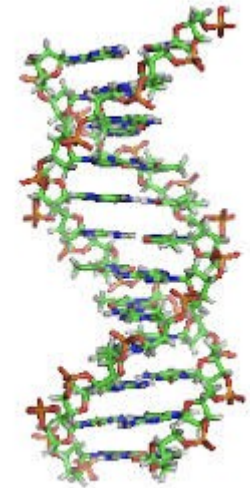
Corrosion in nuclear power plants



Medical x-ray imaging



Air travel and spaceflight

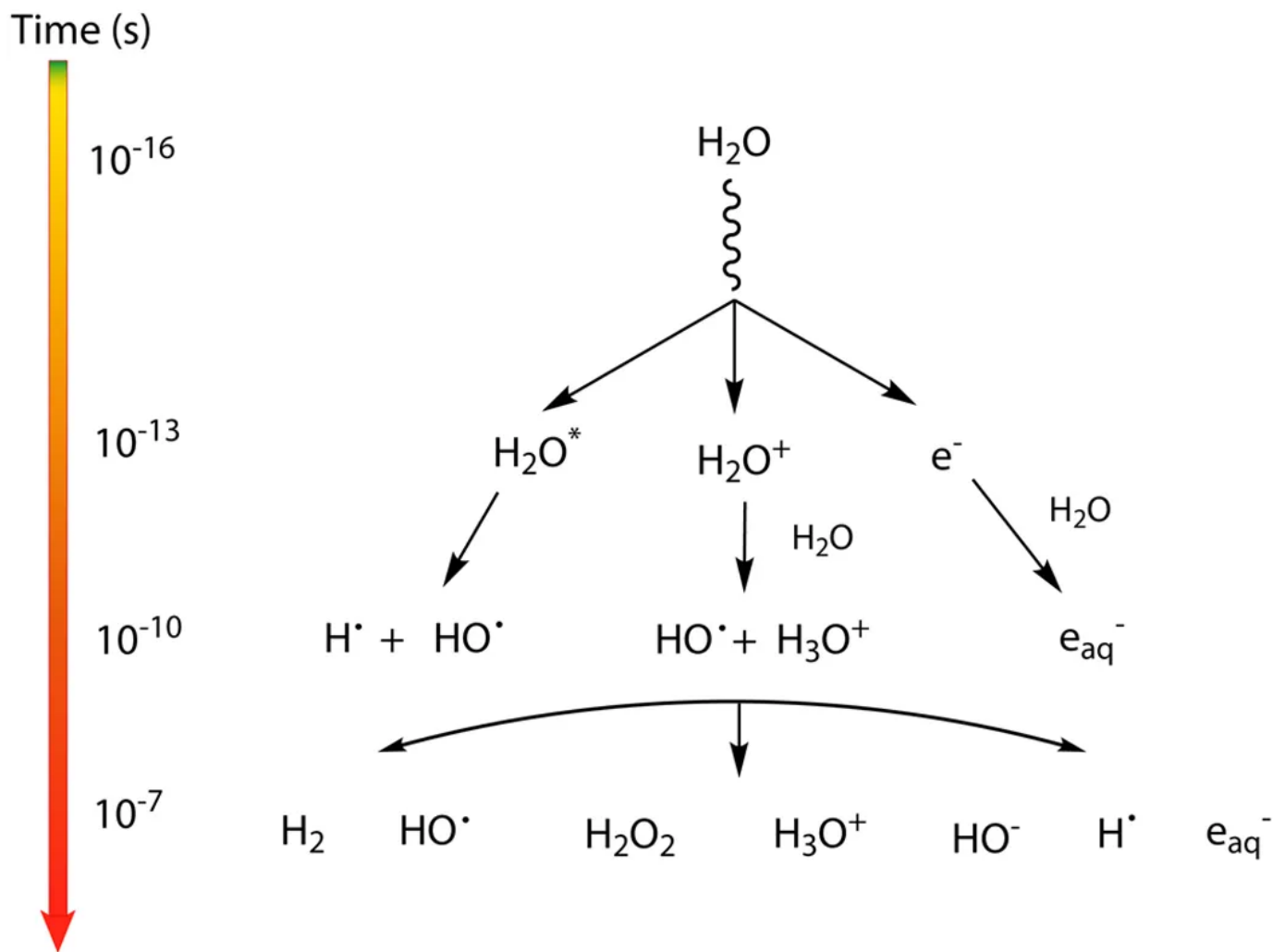


Radiotherapy

## What happens microscopically?

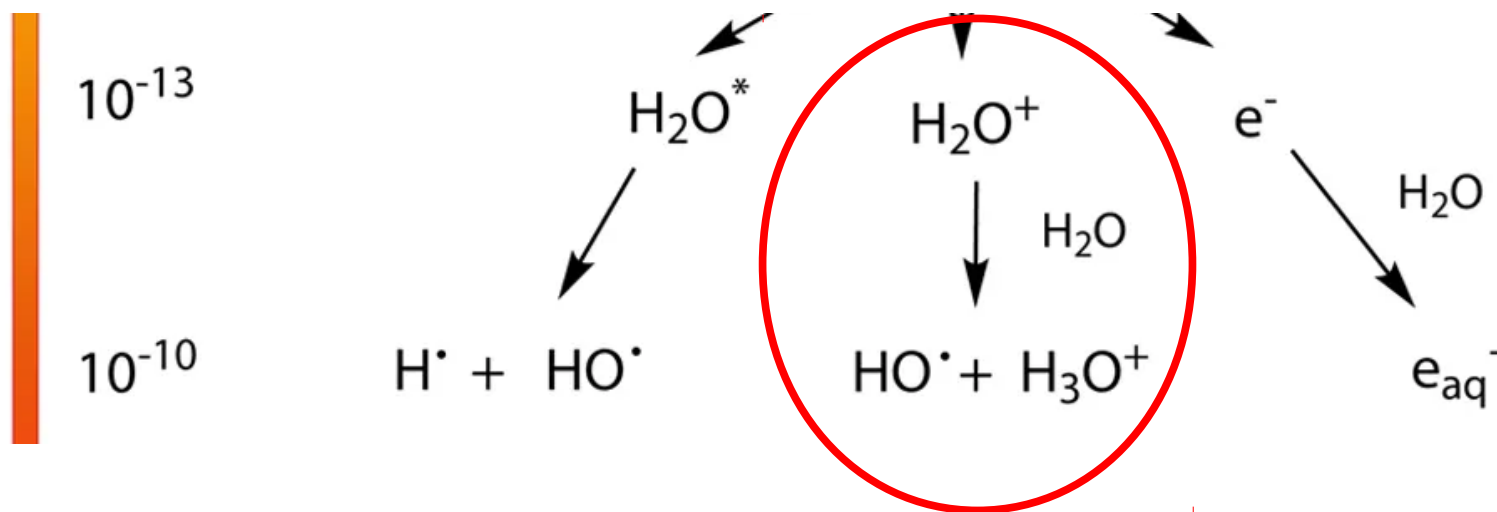
Ionizing radiation (x rays,  $\gamma$  rays, charged particles) gives rise to the **formation of highly reactive radicals**. Particularly, through the **ionization of water**, highly reactive **hydroxyl (OH) radicals** are formed.

# How does ionization of water produce OH?



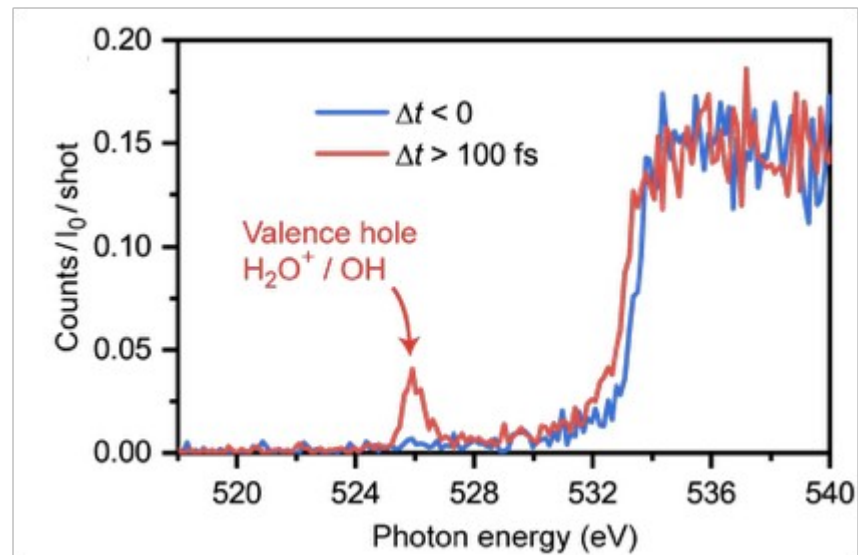
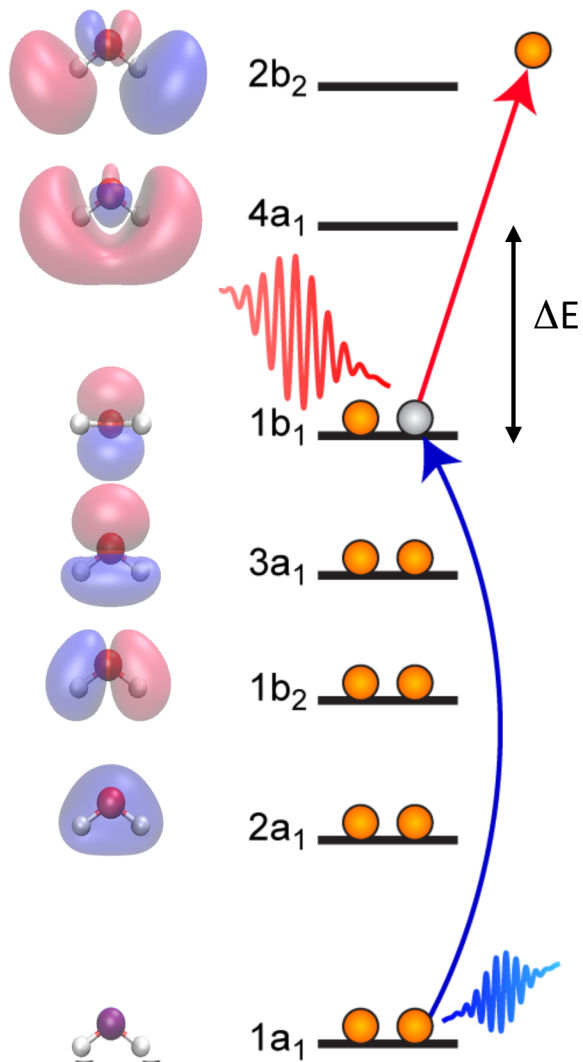
C. M. Lousada *et al.*, Scientific Reports **6**, 24234 (2016).

# Is it true that the key reaction step takes somewhere between 100 fs and 100 ps?



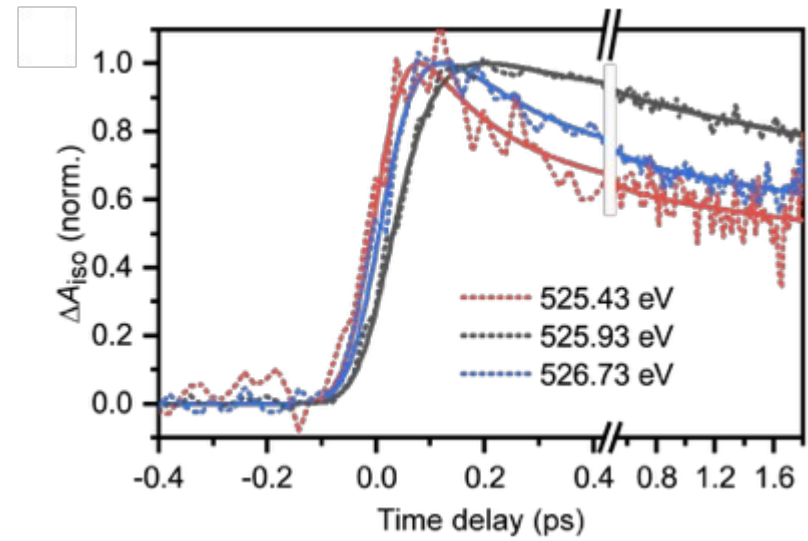
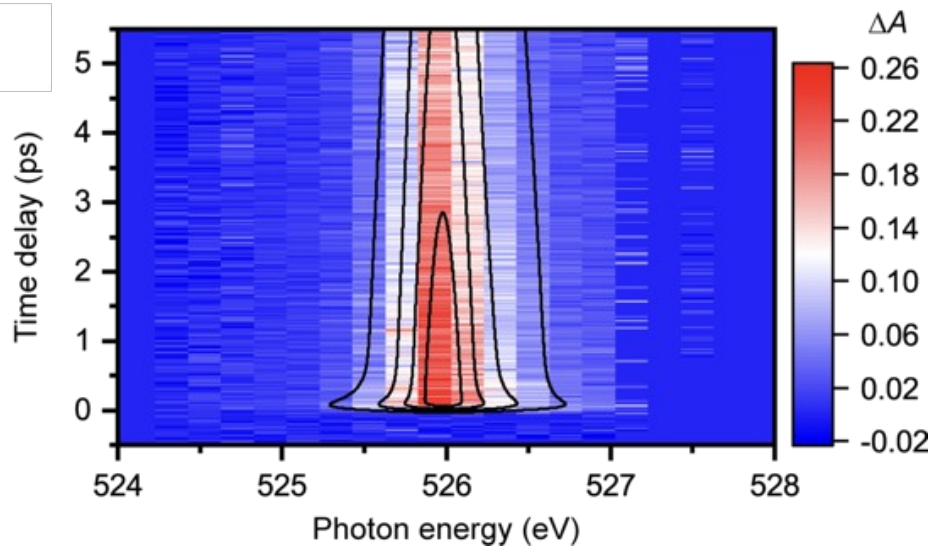
C. M. Lousada *et al.*, Scientific Reports **6**, 24234 (2016).

# Probing the hole through transient x-ray absorption

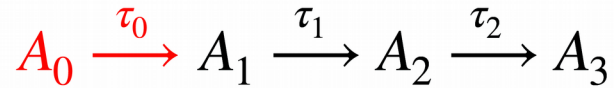


Z.-H. Loh *et al.*,  
Science **367**, 179 (2020).

# Observed delay dependence of x-ray absorption resonance



Sequential kinetics



$\tau_0 = 46 \pm 10$  fs

$\tau_1 = 180 \pm 20$  fs

$\tau_2 = 14.2 \pm 0.4$  ps

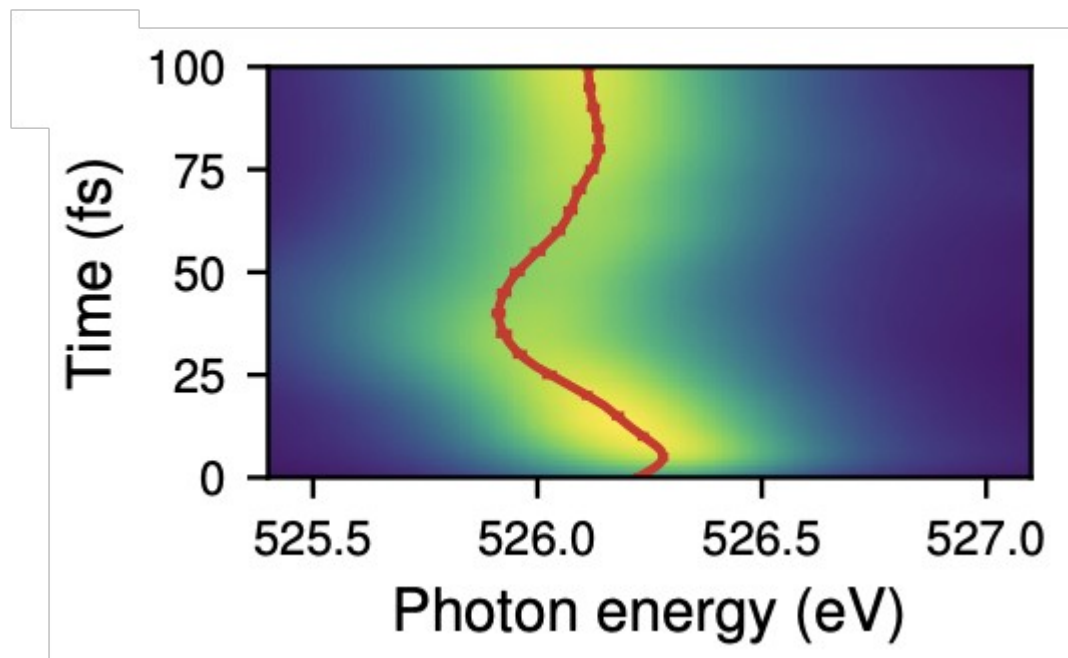
OH formation

Vibrationally hot  
OH cools

OH + e recombine

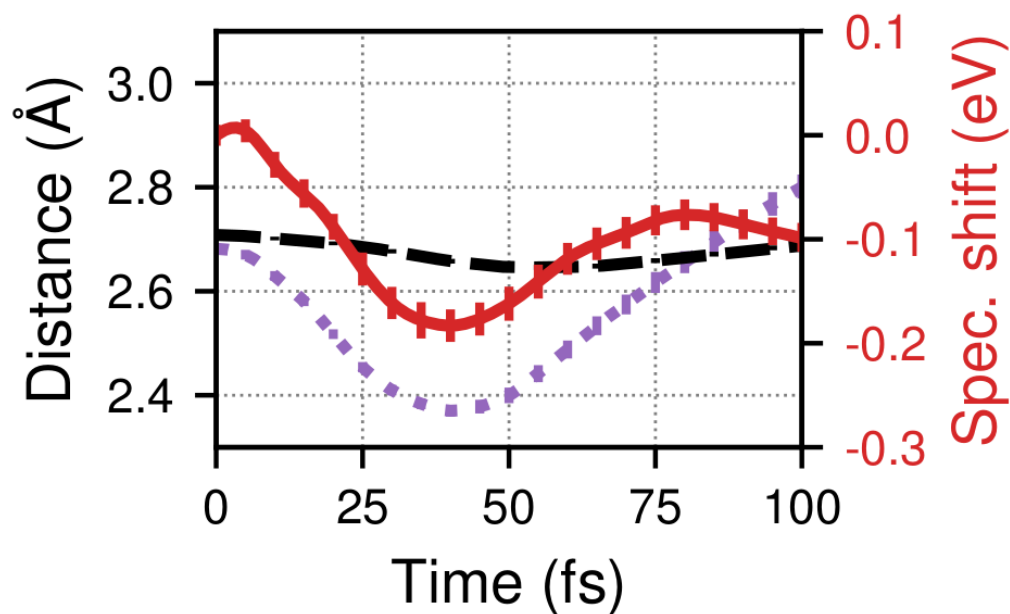
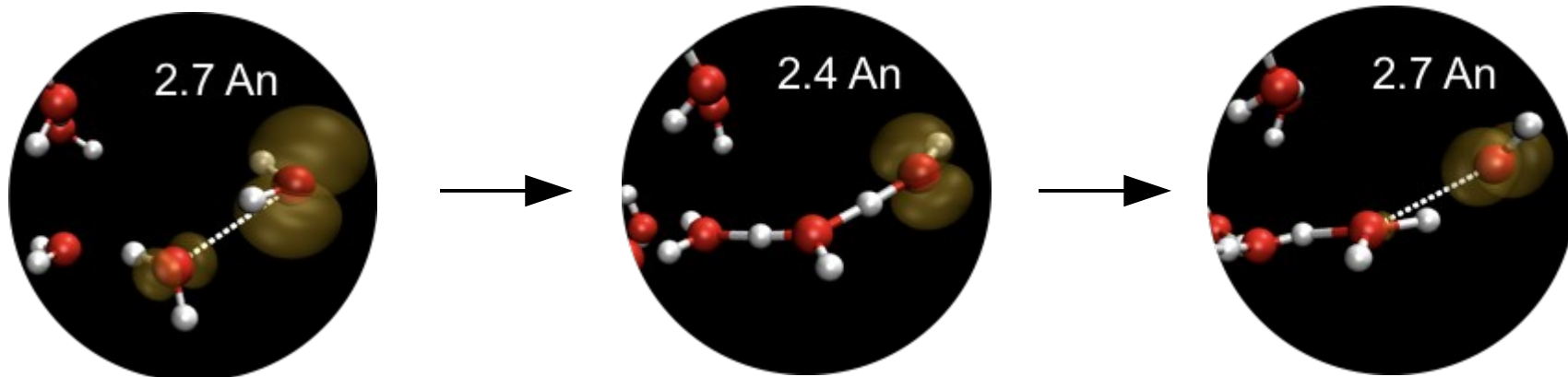
Z.-H. Loh *et al.*, Science **367**, 179 (2020).

# XMOLECULE simulation



How are these spectral dynamics connected to the decay of  $\text{H}_2\text{O}^+$  and the formation of  $\text{OH}$ ?

# Impact of chemical environment of $\text{H}_2\text{O}^+/\text{OH}$



Z.-H. Loh *et al.*, Science **367**, 179 (2020).

# Acknowledgment

- > DESY & Universität Hamburg: **C. Arnold**, R. Welsch, L. Inhester, K. Khalili (DTU)



- > Argonne National Laboratory: **L. Young**, G. Doumy, S. H. Southworth, A. Al Haddad, Y. Kumagai, M.-F. Tu, P. J. Ho, A. M. March, R. D. Schaller
- > Nanyang Technological University: **Z.-H. Loh**, M. S. Bin Mohd Yusof, T. Debnath
- > Uppsala University: L. Kjellsson, J.-E. Rubensson
- > Sorbonne Université and CNRS: M. Simon
- > University of Southern California: K. Nanda, A. I. Krylov
- > SLAC National Accelerator Laboratory: S. Moeller, G. Coslovich, J. Koralek, M. P. Minitti, W. F. Schlotter



# Conclusions

- > Electron transfer in polyatomic systems can lead to significantly enhanced ionization in comparison to independent-atom models.
- > Efficient ionization enables Coulomb explosion imaging.
- > In spatially extended systems, transient plasmas are formed.
- > In such plasmas, atomic displacements are smaller than one might naively expect.
- > Ultrafast x-ray absorption enabled the first observation of the proton transfer reaction following ionization of liquid water, giving rise to the chemically aggressive OH radical.